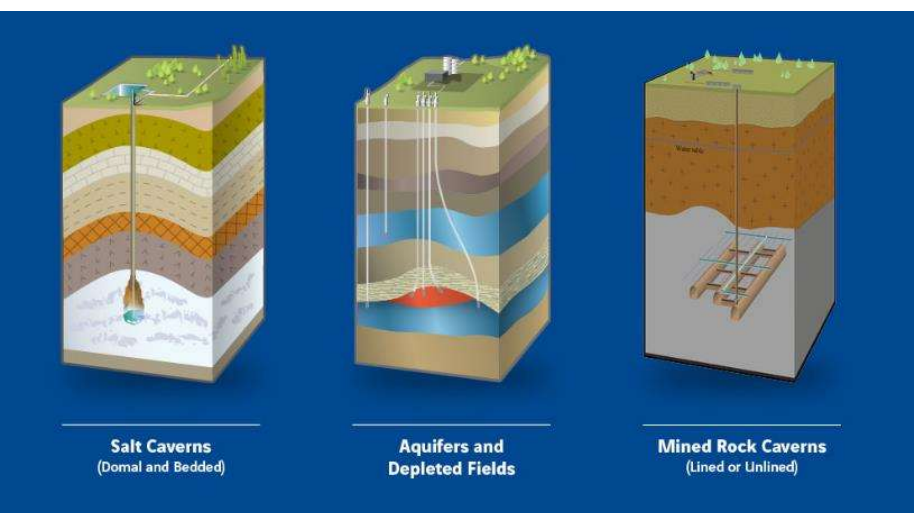




HYDROGEN FOR THE ENERGY TRANSITION : WHY ? HOW ? PART II : PRODUCTION

Dr Vivien Esnault – IFPEN (France)

GeoIndia 2022 – Continuing Education Courses October 12th 2022



PROGRAM OF THE COURSE (1/2)

● Part 1 : Why Hydrogen ?

- The role of hydrogen in the energy transition
- Hydrogen current and future usage
- Case study : Why hydrogen in India ?

● (Tea break)

● Part 2 : The « hydrogen rainbow » : means of production for hydrogen

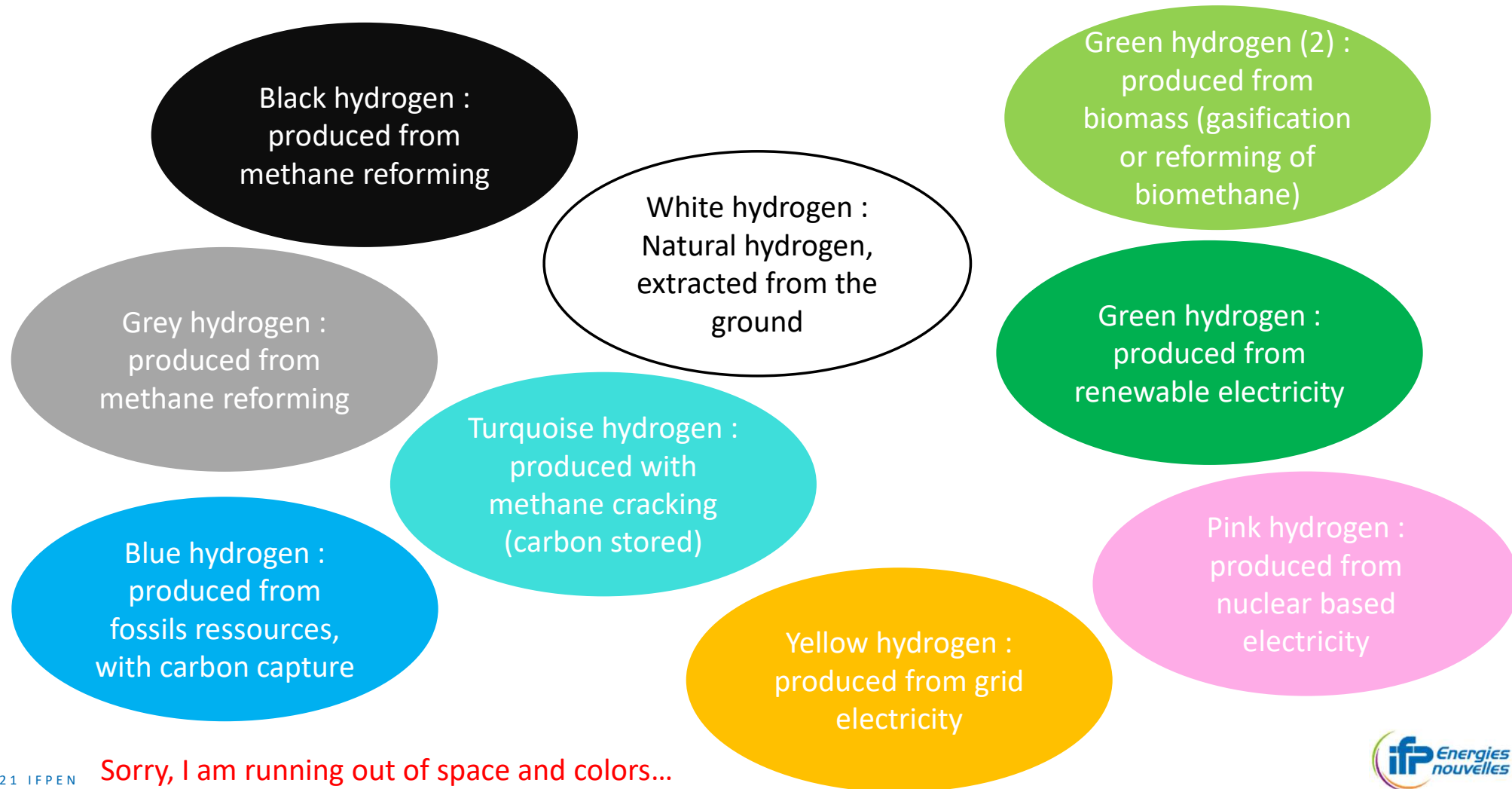
- Green vs blue/black : electrolysis or hydrocarbon-sourced ?
- Environmental impact... and economy
- Case study : India's assets for the production of H₂

● (Lunch break)

PART 2 : THE HYDROGEN RAINBOW : MEANS OF PRODUCING HYDROGEN

- The hydrogen « rainbow » : how can we produce hydrogen, now and in the future ?
 - Hydrocarbon based solutions (« grey » and « blue »)
 - Electrolysis and biomass based solutions (« green »)
- Technico-economic analysis
 - Detailing environmental impacts
 - When and how do I have cheap renewable hydrogen ?
- Case study
 - Indian case analysis : India's assets for the production of H₂

THE HYDROGEN RAINBOW : 50 SHADES OF HYDROGEN



Sorry, I am running out of space and colors...

HYDROGEN DOESN'T HAVE A COLOR, IT HAS A FOOTPRINT

● Hydrogen can be produced from hydrocarbons

- With different processes, and different feedstock
 - Natural gas is not the same as coal
 - Bio-based feedstock are of course a totally different story
- Are CO2 emissions in the environment limited ?
 - Carbone capture is a %, not a yes/no answer

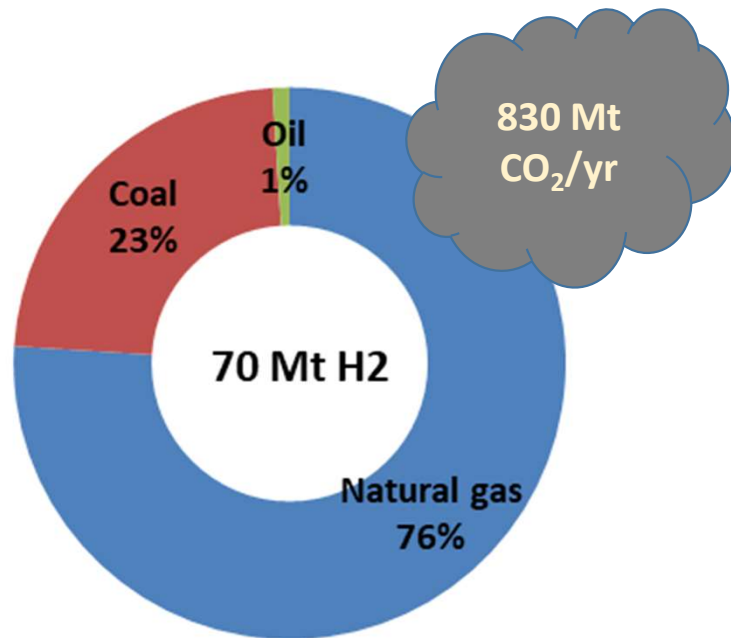
● Hydrogen can be produced through electrolysis

- Hydrogen is just as green as your electricity
 - Electricity is nowhere 100% renewable, and will stay so for a while
- How do you know which electricity went into your production ?

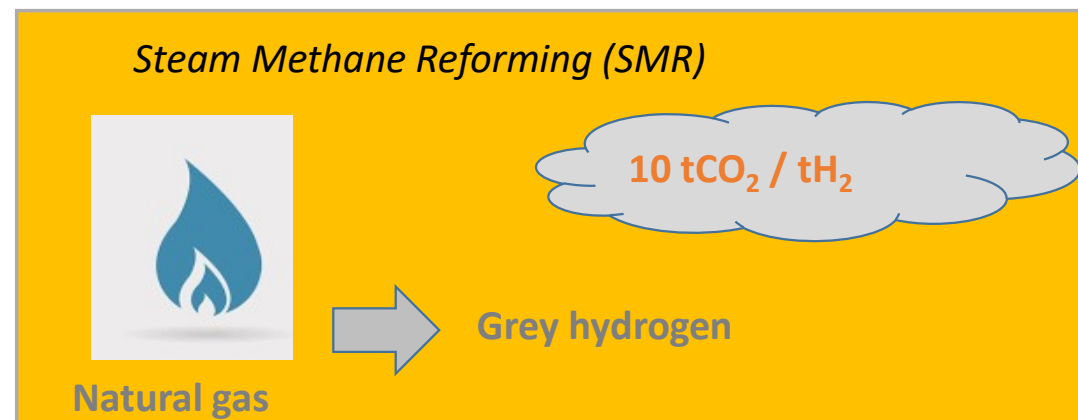
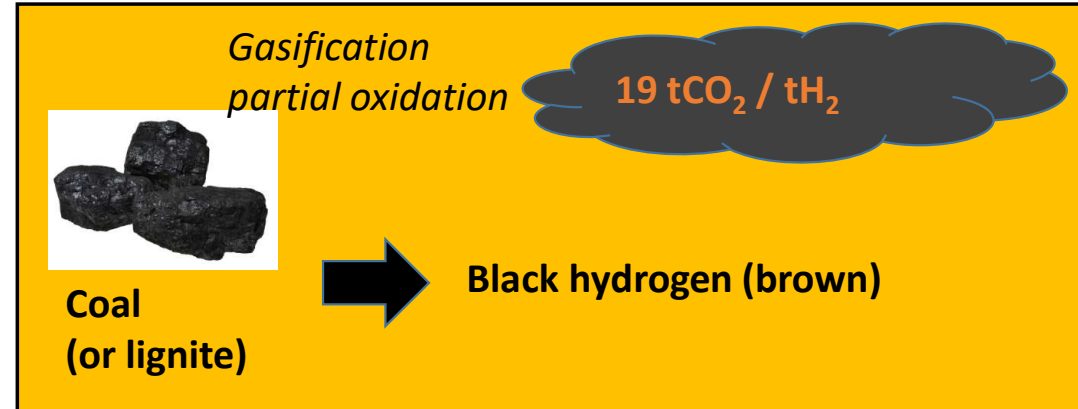
● What's your footprint ?

PRODUCTION OF H₂ FROM HYDROCARBONS : PROCESSES

NOT ALL FEEDSTOCKS ARE THE SAME



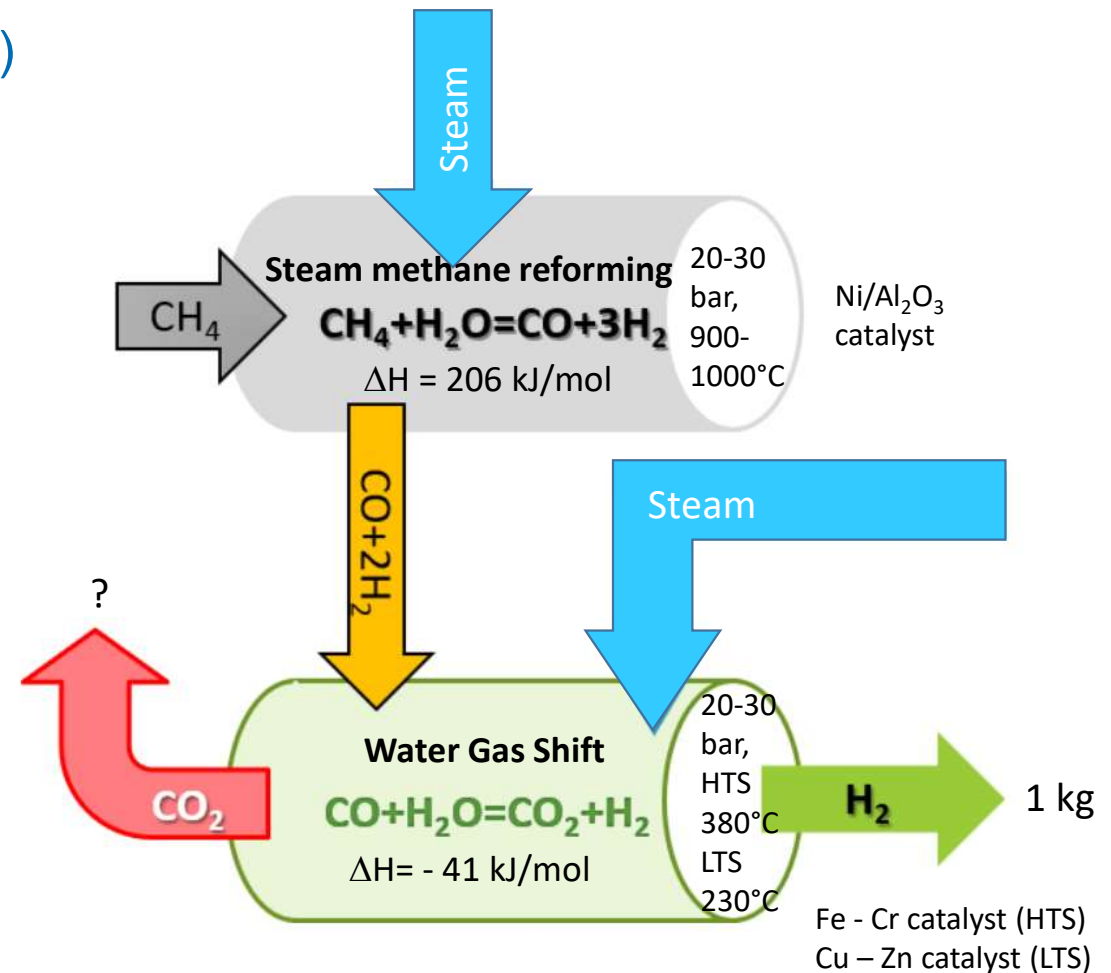
Source: International Energy Agency 2019



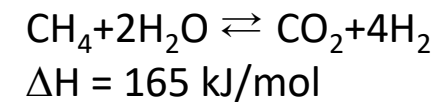
STEAM METHANE REFORMING (SMR)

● Feedstock

- Methane (CH₄)
- Natural Gas: major feedstock
 - 70-90% CH₄
 - Sulfur removal: eliminate H₂S, COS,...
 - Elimination of C₃+
- Refinery Off Gas (ROG)
- LPG (mainly propane and butane)
- Naphta (light cut C₅-C₁₀)
 - Desulfurization
 - Pre-treatment to prevent coking
 - Catalyst resistant to coking

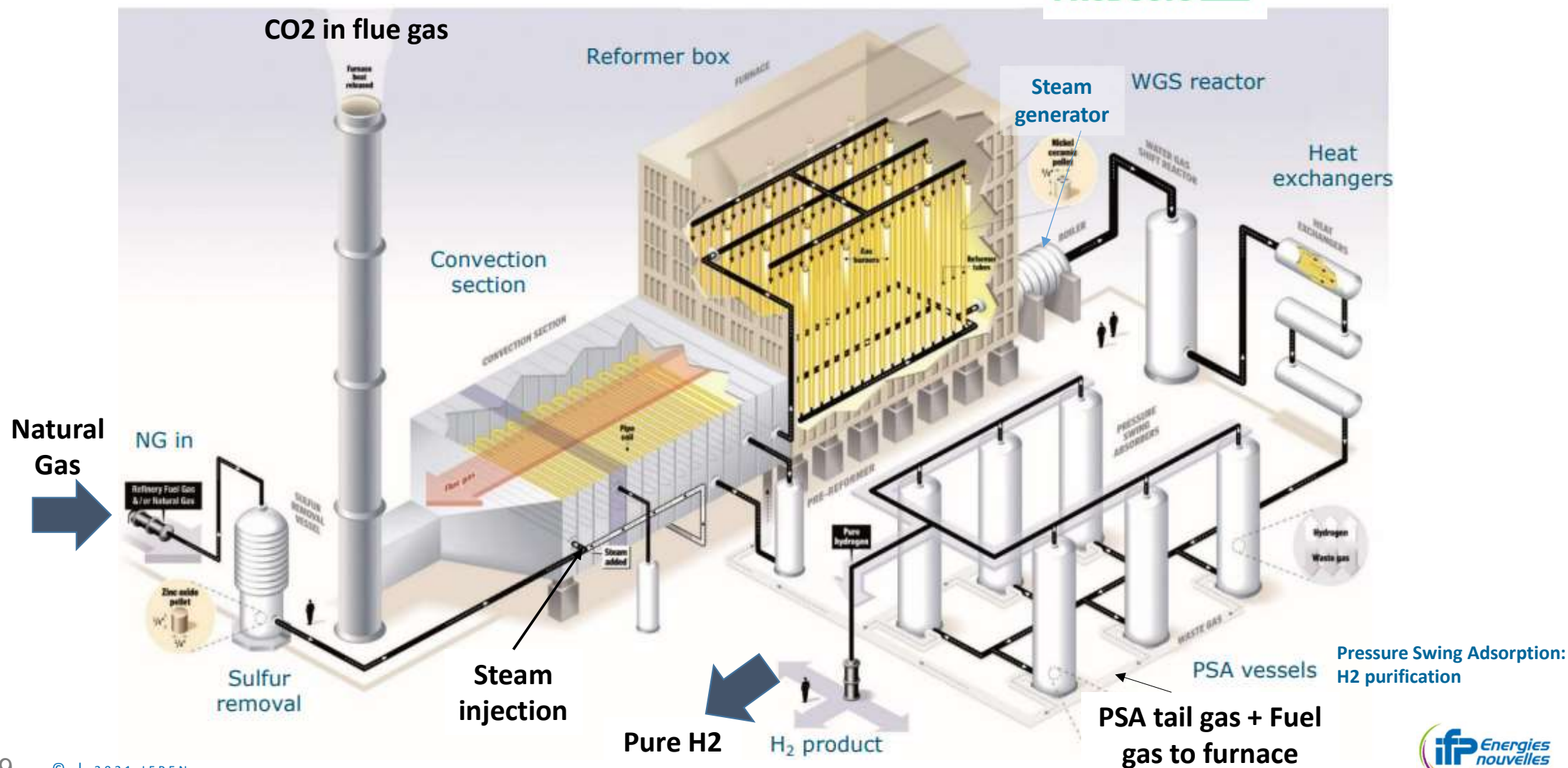


Overall balance



SMR LARGE SCALE UNIT

AIR
PRODUCTS



ALTERNATIVE : AUTOTHERMAL REFORMING (ATR) OR PARTIAL OXIDATION (POX)

- Partial oxidation



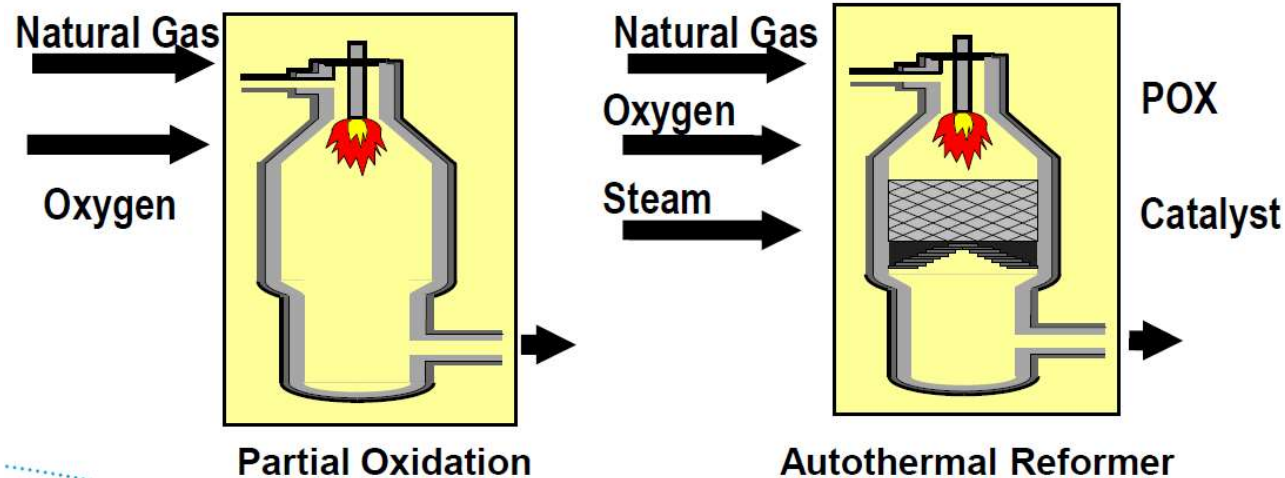
- Exothermic $\Delta H = -36 \text{ kJ/mol}$

- Steam methane reforming



- Endothermic $\Delta H = 206 \text{ kJ/mol}$

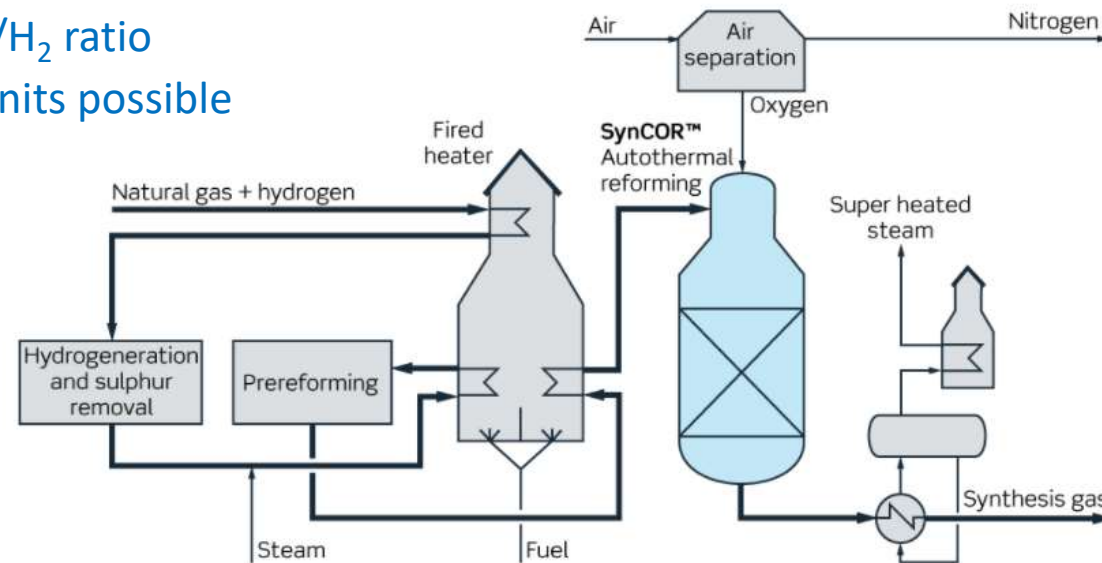
AutoThermal Reforming (ATR): combination of both reaction in the same reactor



AUTOTHERMAL REFORMING ATR

● Benefits

- Any feedstock
- Better thermal integration, yields
- Less sensitive to coking vs. SMR
- Less emissions
- Flexibility on CO/H₂ ratio
- Small compact units possible



Source: Haldor Topsoe



SOLID FUEL GASIFICATION BY PARTIAL OXIDATION

● Main reactions

● Partial oxidation: $C_nH_m + n/2 O_2 \rightarrow n CO + m/2 H_2$

● Gasification: $C + H_2O \rightarrow CO + H_2$

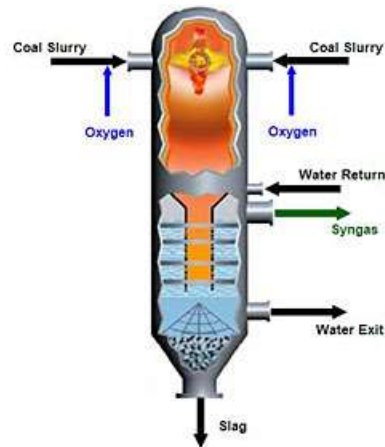
● And many side reactions

● Operating conditions

● 1200-1500°C (controled by O2 flow)

● 20-90 bar

● No catalyst => thermal conversion



Coal is essentially composed of carbon

⇒ Syngas composition : rich in CO

⇒ Attractive for « power to liquid » applications

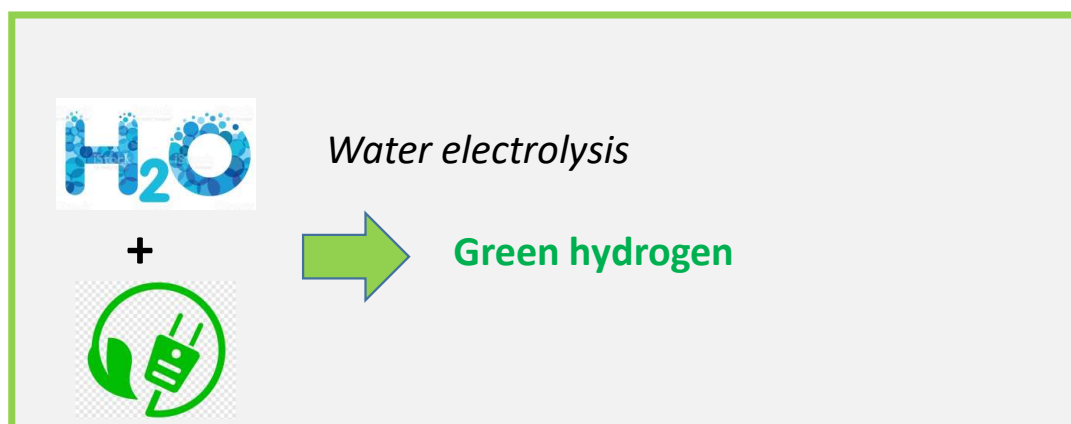
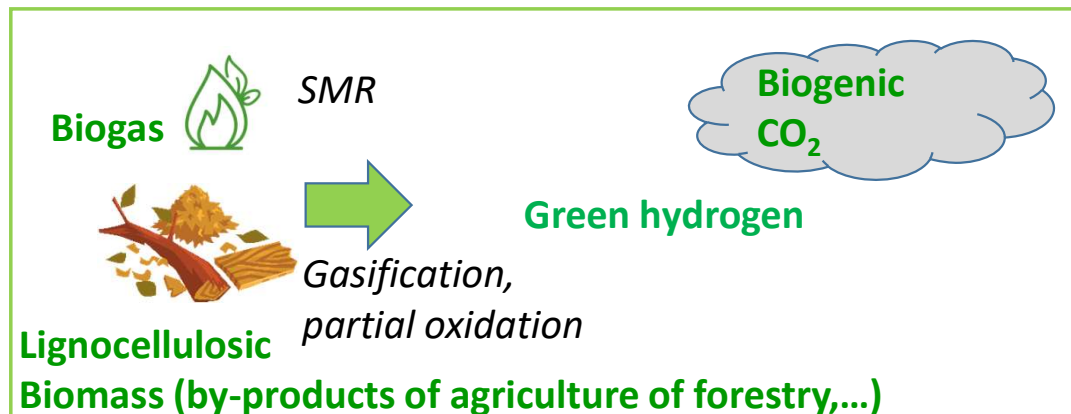
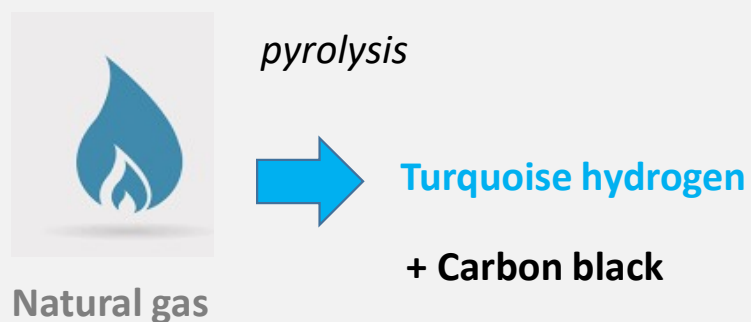
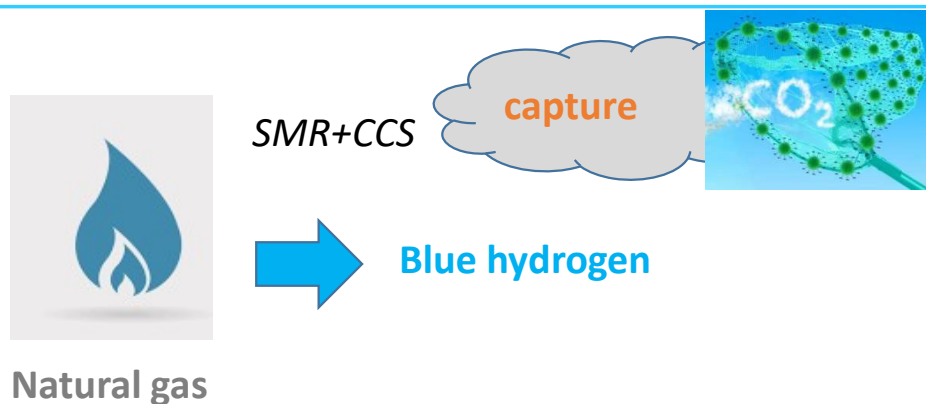
		Charbon	Coke pétrole	Lignite	RSV	Biomasse
H ₂	% vol G.S	27	22	31	45	29
CO	% vol	64	65	55	48	49
CO ₂	% vol	3	5	8	4	14
CH ₄	% vol	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1
N ₂	% vol	5.5	6.5	4.3	2.9	6.3
H ₂ S	% vol	0.46	1.3	0.2	0.1	0.12
COS	% vol	0.04	0.16	0.02	0.01	<0.1
HCN	mg/m ³	1	0.8	1	0.2	0.3
NH ₃	mg/m ³	0.4	0.3	0.24	0.4	0.4

Vs. syngas composition by SMR

Matière première	Méthane % volume	Gaz naturel % volume	GPL % volume	Naphta % volume
CH ₄	3,06	2,91	2,39	2,12
CO	12,16	12,62	13,62	14,17
CO ₂	9,66	10,40	12,73	14,19
H ₂	75,12	73,98	71,86	69,52
N ₂	-	0,09	-	-

PRODUCTION OF H₂ FROM HYDROCARBONS : MITIGATING EMISSIONS

LOW-CARBON HYDROGEN OPTIONS



CO2 CAPTURE FROM SMR (BLUE HYDROGEN)

3 options

Table 1: Levelised Cost of H₂ (LCOH), CO₂ Avoidance Cost and Overall CO₂ Capture Rate (IEAGHG, Techno-Economic Evaluation of SMR Based Standalone (Merchant) Hydrogen Plant with CCS. Technical Report 2017-02, 2017)

Capture Case	LCOH Euro Cent/Nm ³	CO ₂ Avoidance Cost Euro/t	Overall CO ₂ Capture Rate
No capture	11.4	-	-
Option 1	13.5	47.1	56%
Option 2	14.2	66.3	54%
Option 3	16.5	69.8	90%

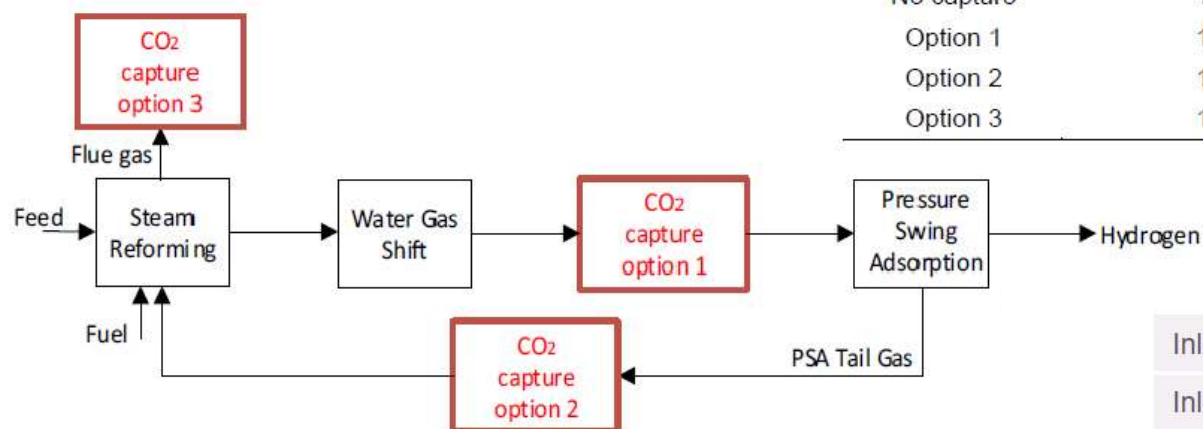


Figure 1: Steam methane reforming - CO₂ capture options

	Option 1 Syngas	Option 3 Flue Gases
Inlet Gas Pressure	30 bar	Atm
Inlet Gas CO ₂ Content	17.3%	18%
Inlet Gas CO ₂ partial pressure	5.2 bar	0.18 bar
O ₂ in the inlet gas	No – traces	Few %
CO ₂ captured on SMR	60%	90%

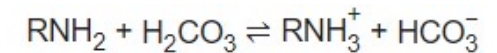
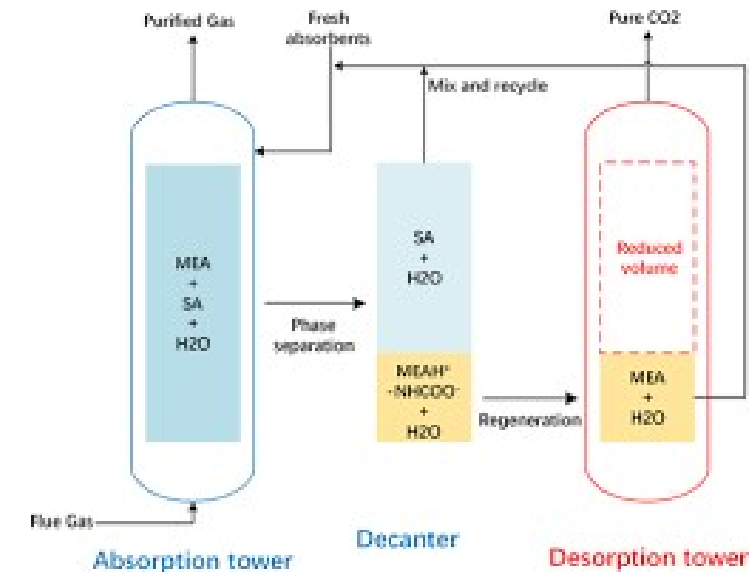
CARBON CAPTURE PROCESS

● Two-step for methane reforming

- Purification : PSA (Pressure Swing Adsorption) will capture all impurities from hydrogen (CH₄, CO₂, CO...)
- CO₂ capture : Most common method use amine to specifically scrub CO₂

● Principle of amine carbon capture

- CO₂ reacts with the amine to form ions (including HCO₃⁻) in the absorption section
- The ions are soluble in water → extracted from the gas phase
- The amine is regenerated through heating in the desorption section



CARBON SEQUESTRATION

- A large variety of geology can trap CO₂

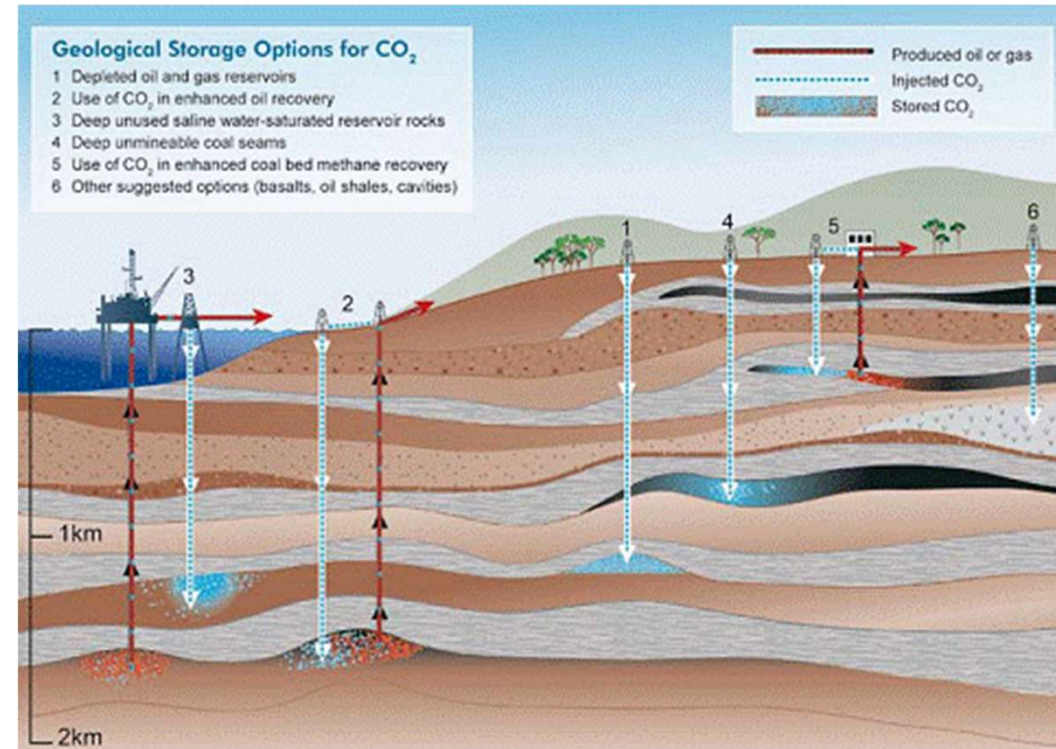
- O&G wells
- Coals seams
- Saline aquifers

- Still a lot of questions...

- Durability of the trapping (cracks, degraded wells...)
- Capacity to pump large quantities
- Logistics

- Are we doing it or not ?

- A lot of demos
- No industrial scale ecosystem



HYDROGEN, AN IDEAL CANDIDATE FOR CCUS ?

- A perfect case for CCUS :

- The process is « naturally » producing streams for almost pure CO₂
 - Not the case for many industrial flue gases
- Large centralized units → easier logistics
- Connection with gas producers... when gas wells are ideal candidates for capture

- Explain why the economical case sounds good :

- CO₂ could be captured and stored for between 75 to 150 \$/t
- Including CAPEX? It represents a premium around 0.5 \$/kg of H₂, against a typical production cost of 2\$/kg

- Only problem : there is an alternative

- Is blue hydrogen necessary ?
- Is blue hydrogen going to be available before green hydrogen ?



IEA



MAN Energy Solutions

BIOMASS GASIFICATION

Renewable
energies

● High temperature gasification

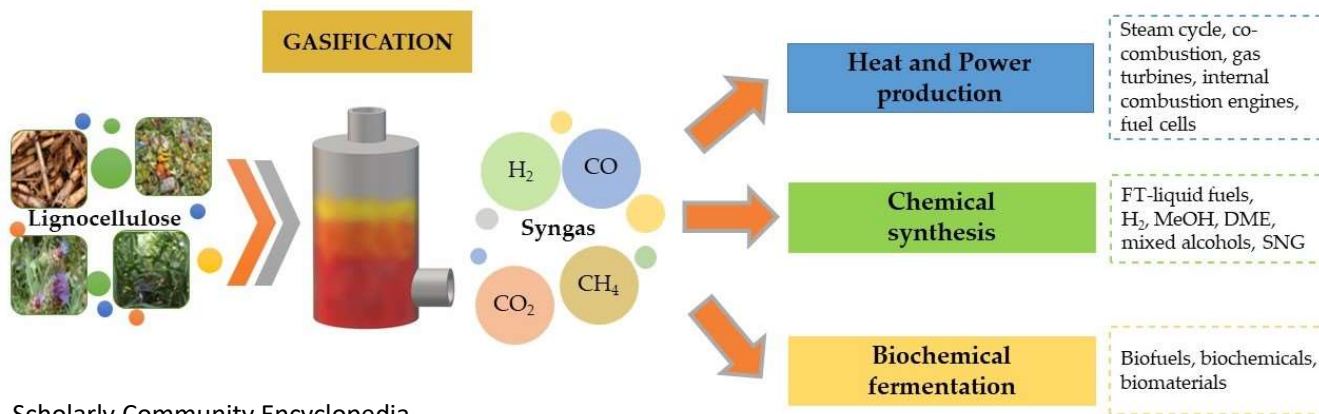
- Robust, well-known technology
- Produce a CO-rich syngas

● Feedstock

- Wastes rich in lignin
- Wood chips, variety of agricultural waste



US Department of Energy



Scholarly Community Encyclopedia



ifp *Energies nouvelles*

HYDROGEN FROM BIOMASS, DISCUSSION

● Mature processes

- « Simply » transposing processes already existing for fossil feedstock
- Not at the same scale, though

● Economy is not favorable, at first sight...

- Economy of scale is an important factor of what makes SMR/gasification cheap
- The feedstock itself is expensive !
- And limited in volume...

● More than just hydrogen production

- Possibility to generate negative emissions with carbon capture
- High CO syngas → Relevant process for e-fuel/ methanol production
- Has a role to play in the waste economy

METHANE PYROLYSIS

- Methane decomposition:



- Compared to Steam Methane Reforming

- $\Delta H = 75 \text{ kJ/mol}$ vs. $\Delta H = 206 \text{ kJ/mol}$ for SMR
- Use more natural gas than SMR ($2 \text{ H}_2/\text{CH}_4$ vs. $4 \text{ H}_2/\text{CH}_4$)
- Produces C instead of CO_2 : easier to store
- Use electricity as a source of energy

- Compared to water electrolysis

- Use 3 to 5 times less electricity than electrolysis

- Overall energy conversion efficiency of methane and electricity combined into hydrogen is 40-45%

- Additional revenue by selling carbon black

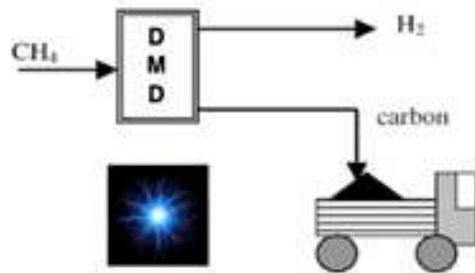
- For rubber, tires, printers
- Actual demand of carbon black is 16 Mt corresponding to 5 Mt of H_2

Options for turquoise H_2 production

Liquid metal	Thermo Catalytic decomposition	Plasma
Liquid metal / salts Catalytic / non catalytic <ul style="list-style-type: none"> • Hazer (Australia) • Czero (CA, USA) • Ember-TNO (The Netherlands) • KIT (Karlsruhe) • IASS (Posdam) • TNO (The Netherlands) • BNL (USA) 	Most R&D in Germany Fixed bed / Fluidized Bed / <ul style="list-style-type: none"> • BASF • Linde Group • ThyssenKrupp • TUD (Dortmund) • TU Bergakademie (Freiberg) • RUB (Bochum) 	Thermal / Non thermal <ul style="list-style-type: none"> • MONOLITH Materials (US) • MPT (FR) LEADERSHIP POSITION TRL = 8

PERSEE

PLASMA TECHNOLOGY FOR METHANE PYROLYSIS



+ 75 kJ.mole⁻¹
5.2 kWh / kg H₂

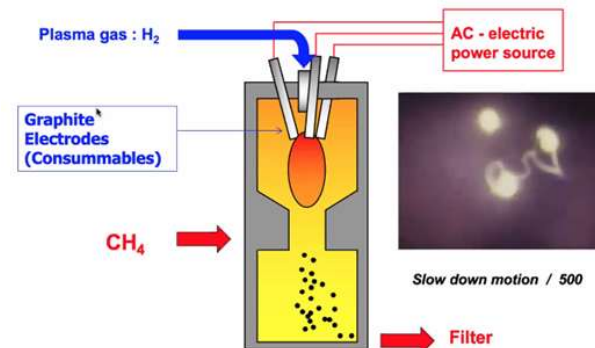
10-20 kWh / kg H₂

PERSEE

Laurent FULCHERE - Methane Pyrolysis - IFP - 2011, March 17th

3-phase AC

PSL ★



IDECC

Industrial development

PSL ★

2012-2018: Demonstration Pilot 1 MW, Mountain View USA-CA

2021:

- Commissioning 1st commercial CB&H₂ unit, *Olive Creek One (OC1)*, Nebraska
- OC2, + NH₃ production « in the pipe »

vs. Water electrolysis : 50-60 kWh /kg H₂

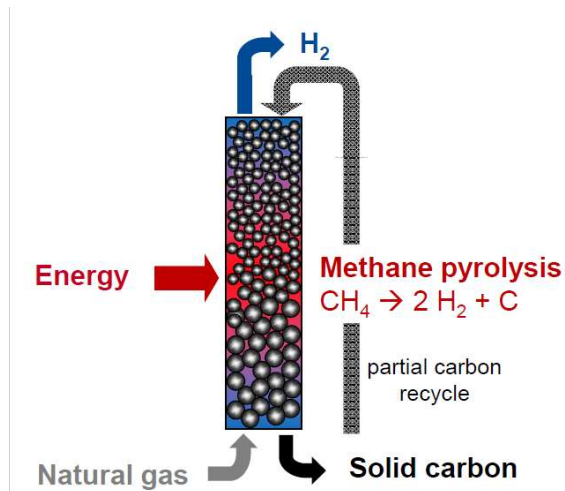
Monolith Materials

<http://monolithmaterials.com/>

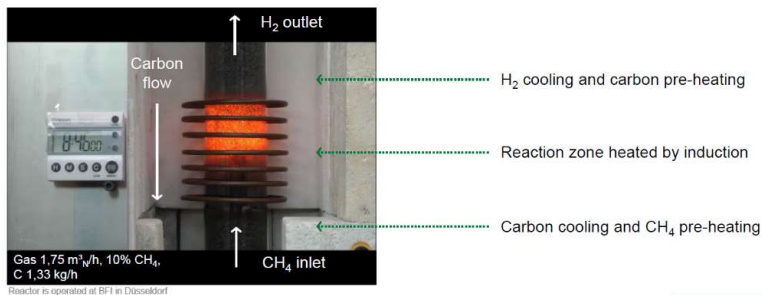


METHANE PYROLYSIS TECHNOLOGIES AT LAB SCALE

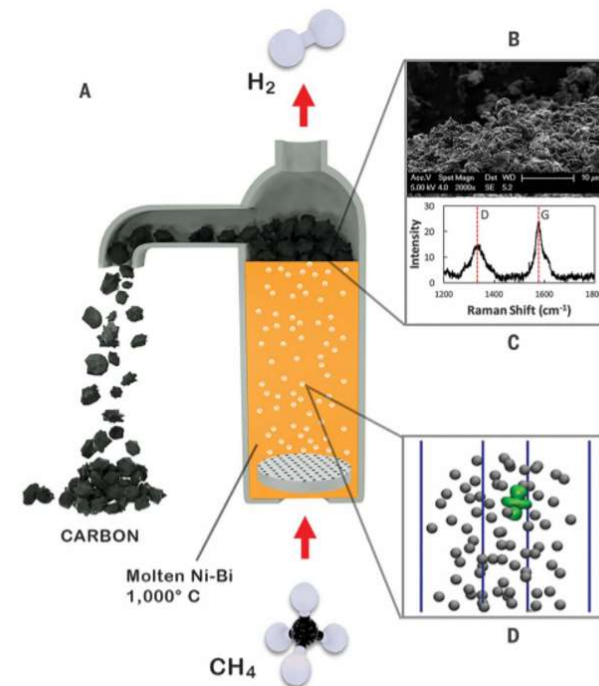
● Moving bed (BASF)



First video observation of methane pyrolysis on a moving carbon bed in glass lab reactor

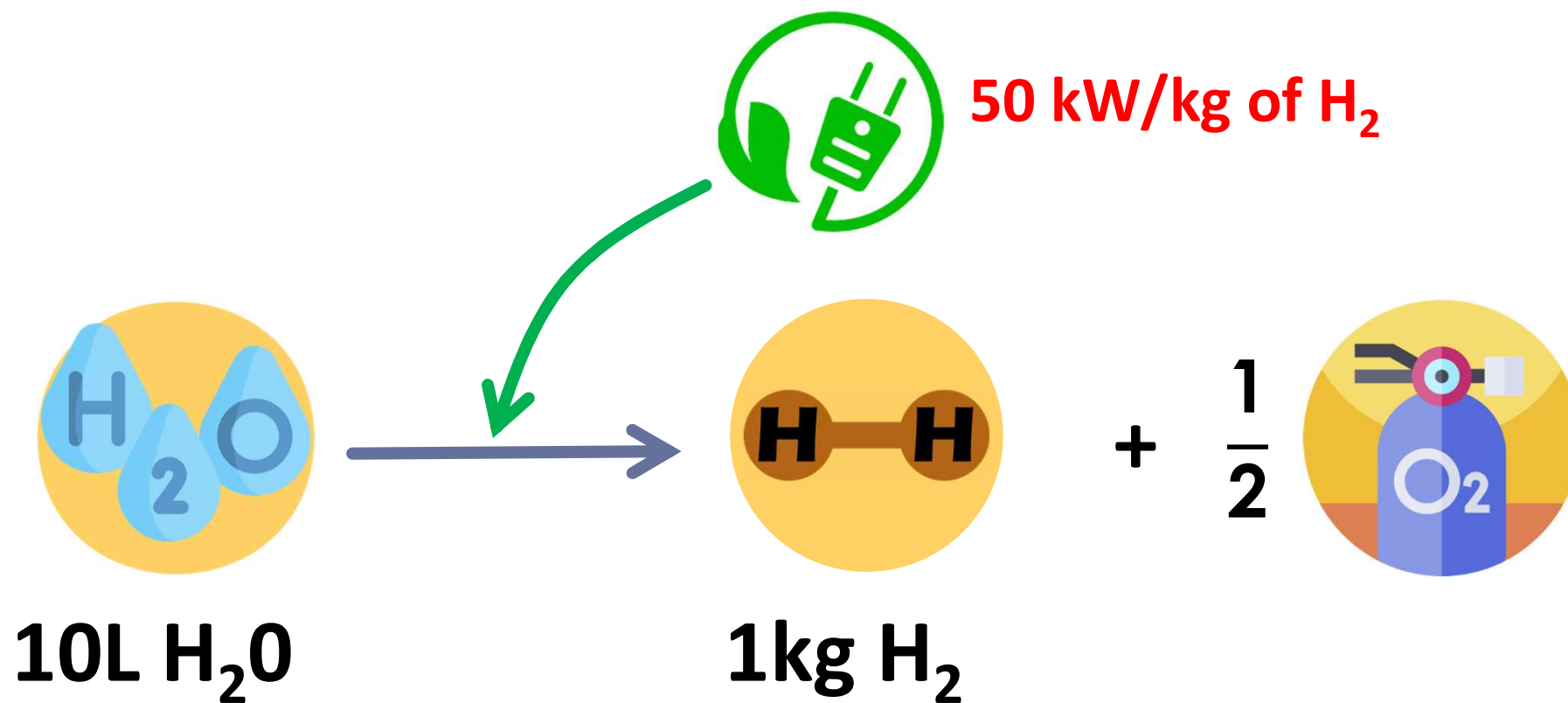


● Molten salts (KIT)



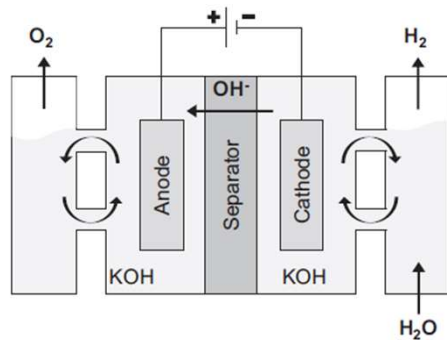
PRODUCTION OF H₂ : ELECTROLYSIS

WATER ELECTROLYSIS - POWER TO GAS



ELECTROLYZER TECHNOLOGIES

Alkaline AEC



Stack energy
efficiency / LHV

60-68%

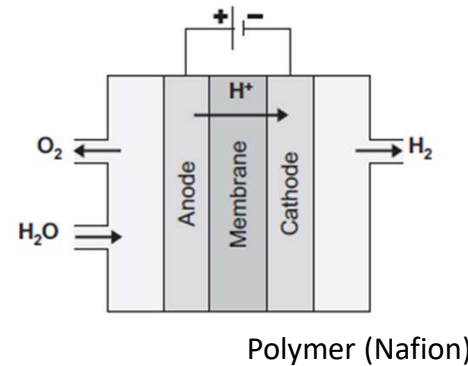
Temperature

60-90 °C

Technological
maturity

Mature

Proton Exchange Membrane PEMEC

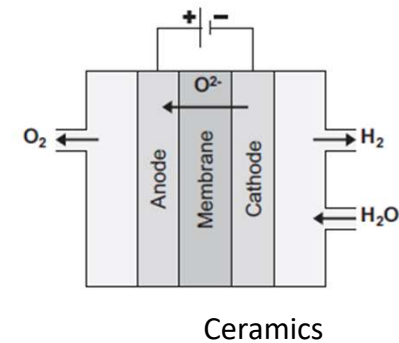


50-76%

50-80 °C

Commercial

Solid Oxide SOEC



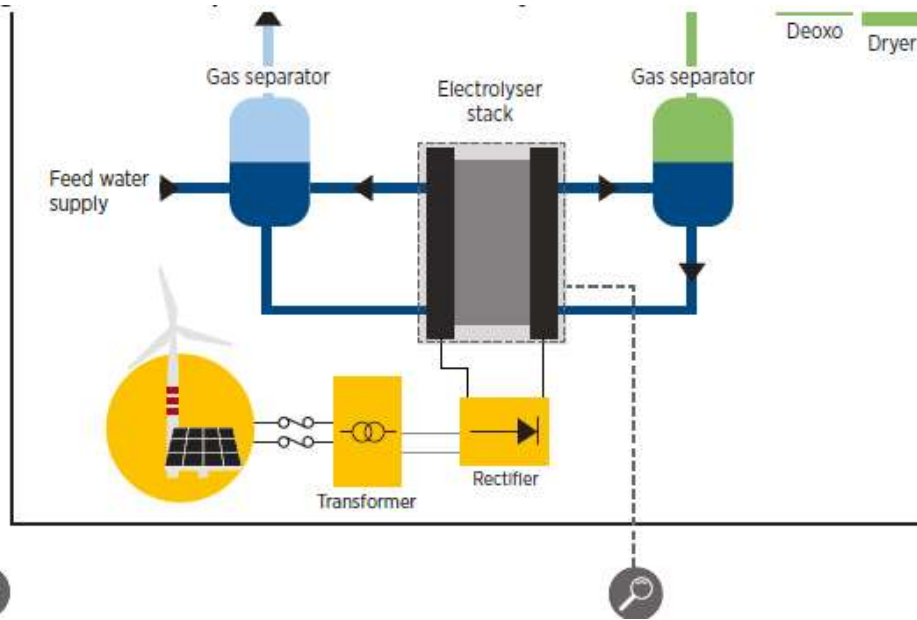
96%

700-900 °C

Demonstration

And some (less mature) others : Anion exchange membranes, photoelectrolysis, electrolysis of biomass

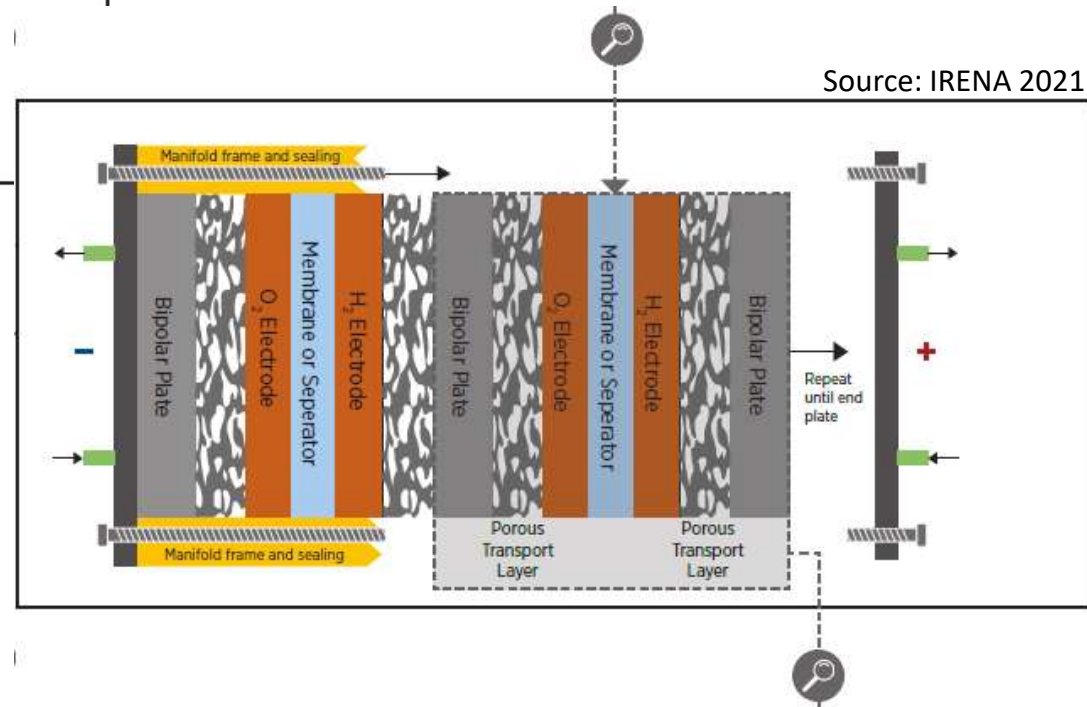
BASIC COMPONENTS OF WATER ELECTROLYZER AT DIFFERENT LEVELS



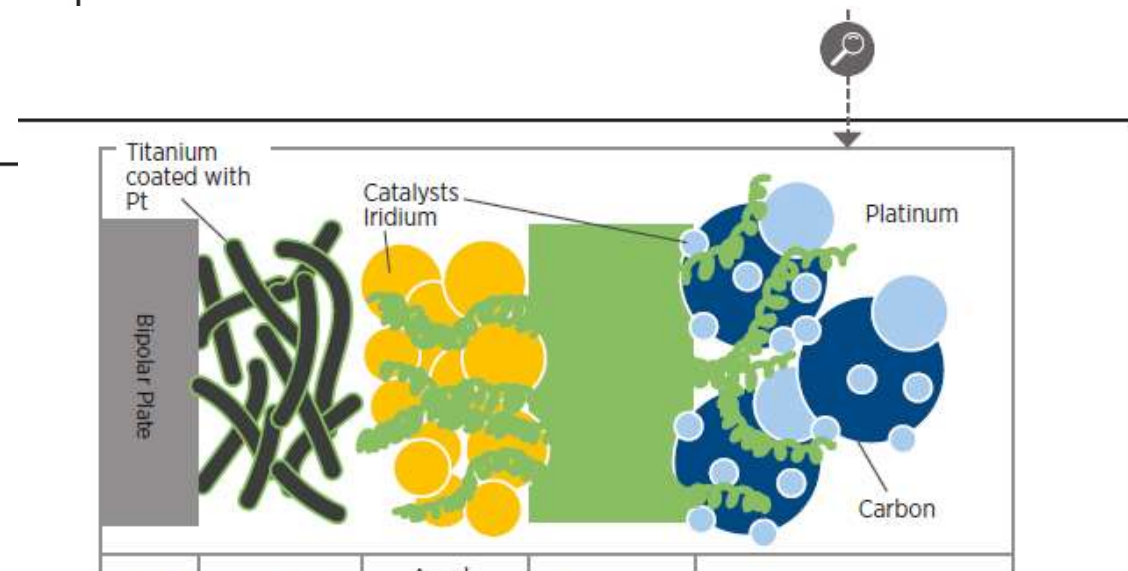
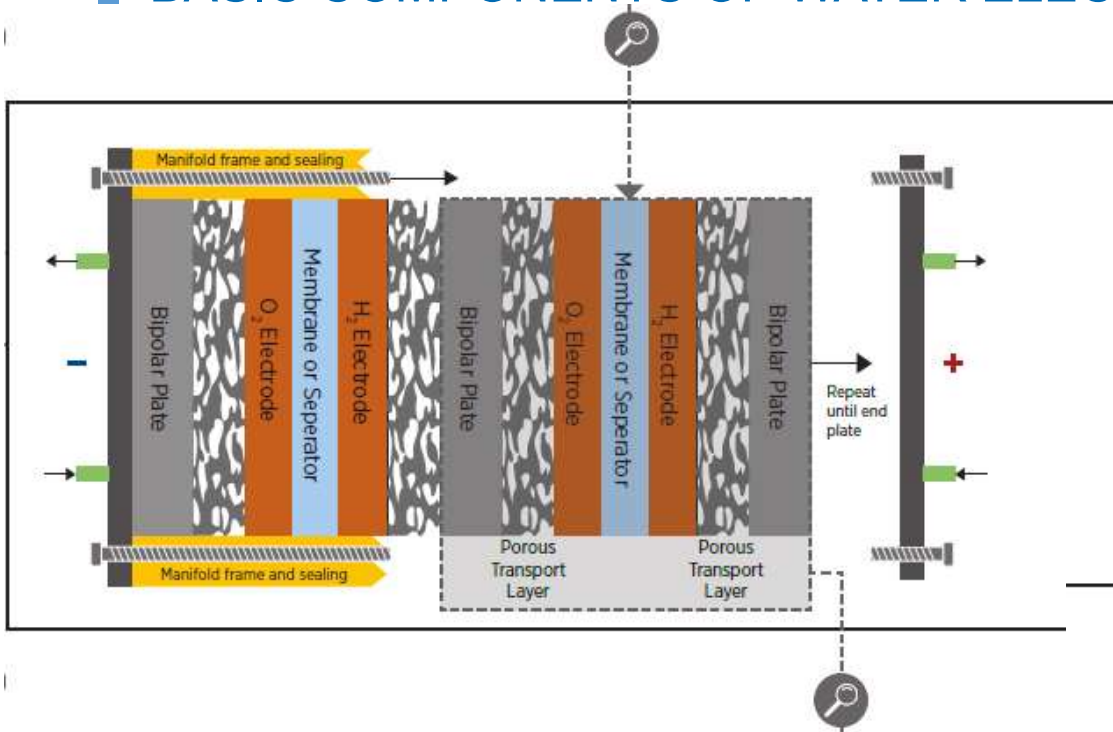
H₂ 99.5- 99.9% (after drying)

O₂ 99- 99.8% (after deoxo & drying)

	AE	PEM	SOE
η stack	60-68%	50-76%	96%
η plant (utilities, rectification, compression)	40-60%	40-60%	76-81%



BASIC COMPONENTS OF WATER ELECTROLYZER AT DIFFERENT LEVELS

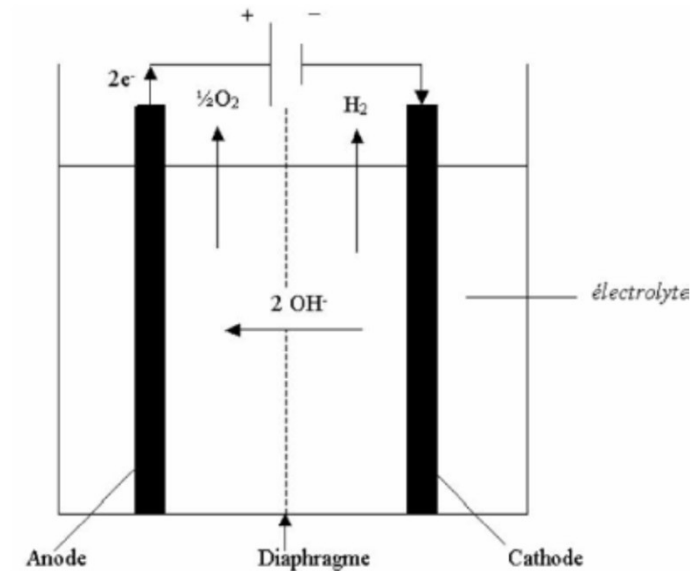


Source: IRENA 2021

O_2 = oxygen; BP = bipolar plates; PTL = porous transport layer
 based on IRENA analysis.

ALKALYNE ELECTROLYSIS

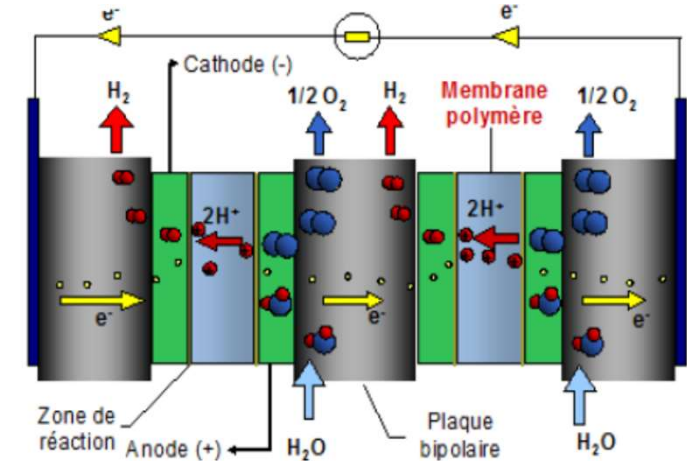
- Transport of OH^- ions in water solution
- An « old » technology
 - Used for decades for Cl_2 and NaOH production
 - Not very compact, not very high pressure...
- Don't dismiss too fast old technologies...
 - Efficiency vastly improved and now close to PEM
 - No expensive catalysts !



France Hydrogène

PEM ELECTROLYSIS

- Transport of H^+ ions in polymer membranes
- Scaling-up the technology
 - Known for a long time for small units (submarines)
 - Now scaling up to mass production
 - Compact and efficient, high pressure (70 bar)
- Too reliant on strategic components ?
 - Platine and iridium catalysts needed (for now)
 - In competition for the resource with PEM fuel cells



France Hydrogène



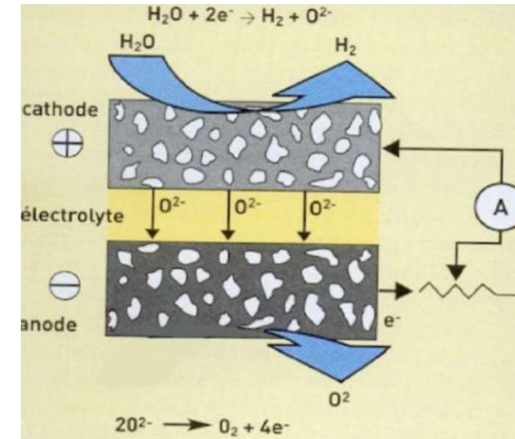
NEL



Air Liquide

SOEC/SOFC ELECTROLYSIS

- Transport of O^{2-} ions in a ceramic membrane
 - Still in development
- The « hot » electrolyzer
 - Operations at 400°C to 1000°C, depending of tech
 - Super efficient... if you have access to waste heat
 - Low pressure of hydrogen → not space efficient
- Nuclear reactor's best friend
 - Adapted to large units, need for waste heat...



France Hydrogène



Encyclopedia Britannica



CEA/LITEN

ELECTROLYZER SIDE EQUIPMENTS

● Water purification

- Electrolysis needs water... but not so much
- Requirements are hard but technologies exists
- If needed, water desanilization not a big deal in the global balance

● Gas purification

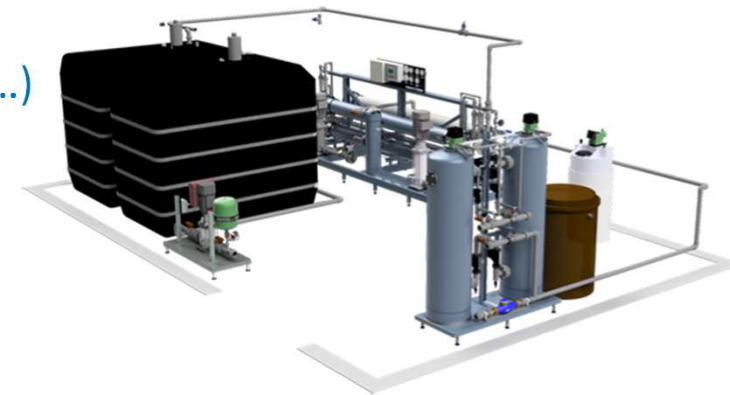
- Very strict requirements
- Less pollution then with reforming, but different (H_2O , $O_2...$)

● Compression

- Efficient transport and usage of hydrogen needs pressure
- Low pressure → larger installations and CAPEX
- Pressure of operation of the electrolyzer is key



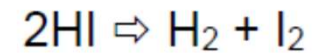
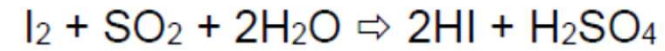
CSIC



Eurowater

THERMOCHEMICAL SPLITTING : ANOTHER WAY TO PRODUCE H₂ FROM H₂O

- At high temperature, water will decompose in H₂ and O₂
 - 3000°C : not very practical
 - The same result can be obtained using reactional intermediate, at lower temperatures : 400°C for Iode/sulphur cycle
- You still need large amounts of heat
 - Initially developped by the nuclear industry
 - Potentially attractive for thermal solar
 - Direct use of heat, no electricity
- Is this too late to the party ?
 - Electrolysis is industrializing, reducing costs...
 - Same apply to other innovative ways of producing hydrogen



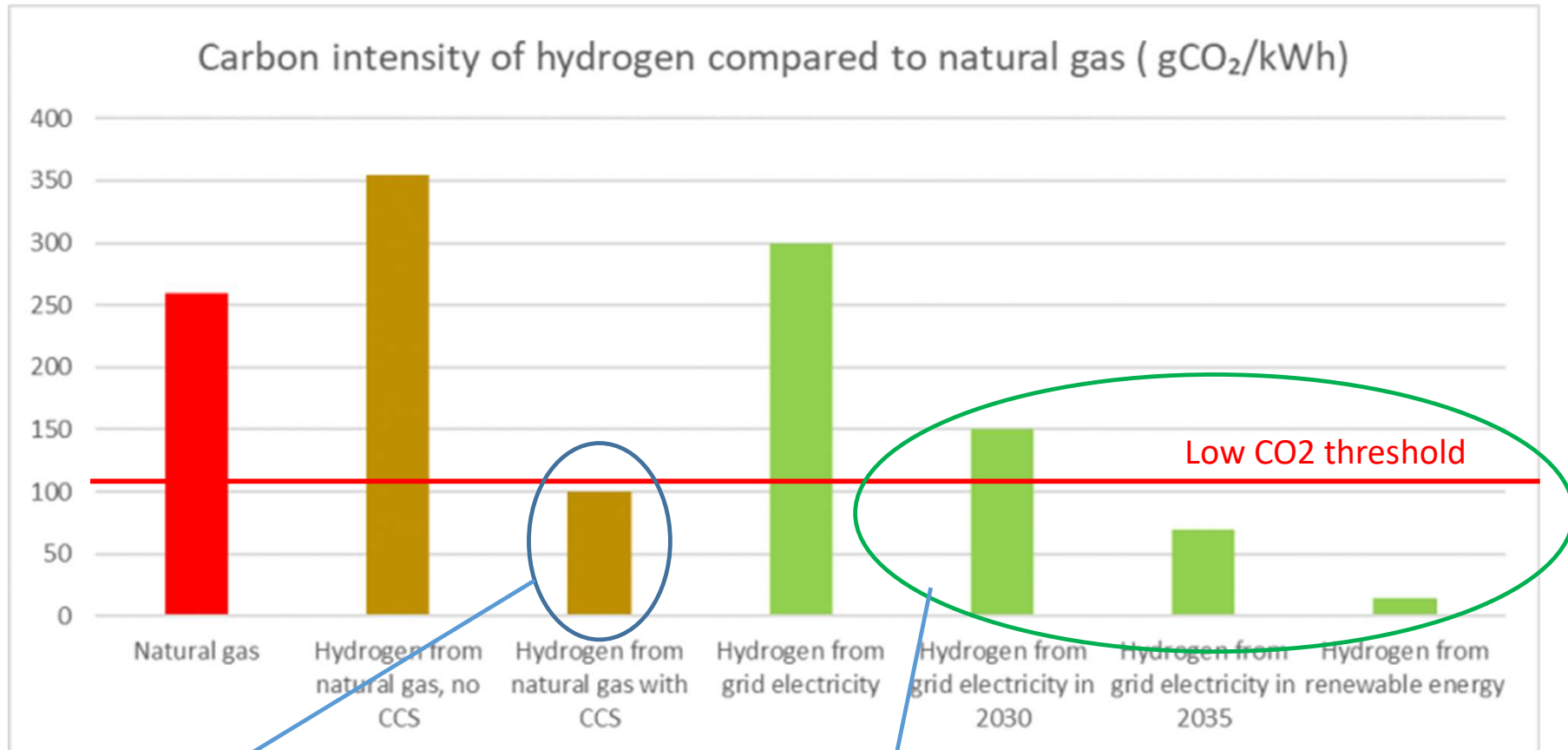
EDF

CARBON FOOTPRINT AND ECONOMIC CONSIDERATIONS

WHAT IS GREEN ?

- Every process has a carbon footprint
 - CAPEX : amortized consumption to produce the system
 - OPEX : carbon footprints of intrants : electricity, methane
- Different approaches to define « acceptable » hydrogen
 - By type of process : complicated, and not satisfying
 - By carbon footprint :
 - International standard for a threshold
 - International standard for a methodology to measure the footprint
- EU most likely definition : low-carbon hydrogen is less than $3 \text{ t}_{\text{CO}_2}/\text{t}_{\text{H}_2}$
 - Arbitrary, but a balance between sufficiently low and reachable by both blue and green

WHAT IS GREEN ?



Friends of the Earth Institute

How good is your capture ?

How good is your electricity ?

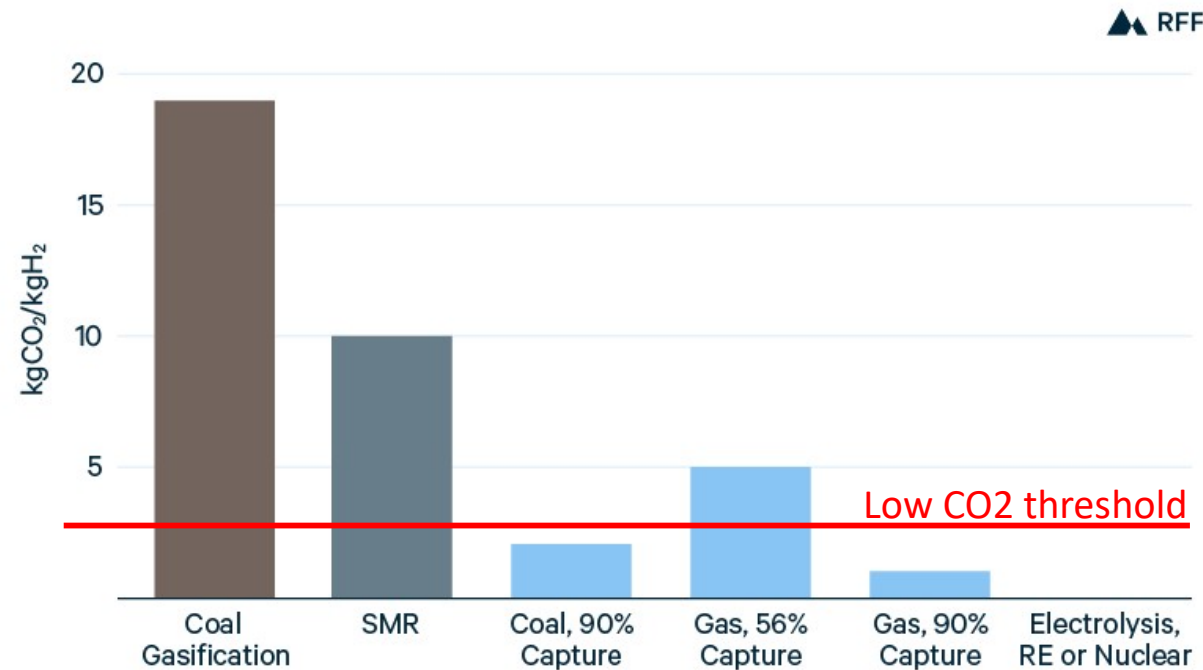
HYDROGEN FROM HYDROCARBON : CCUS NEEDS

● It can be reached...

- But you need « perfect » CCUS
- The process is favorable, but still

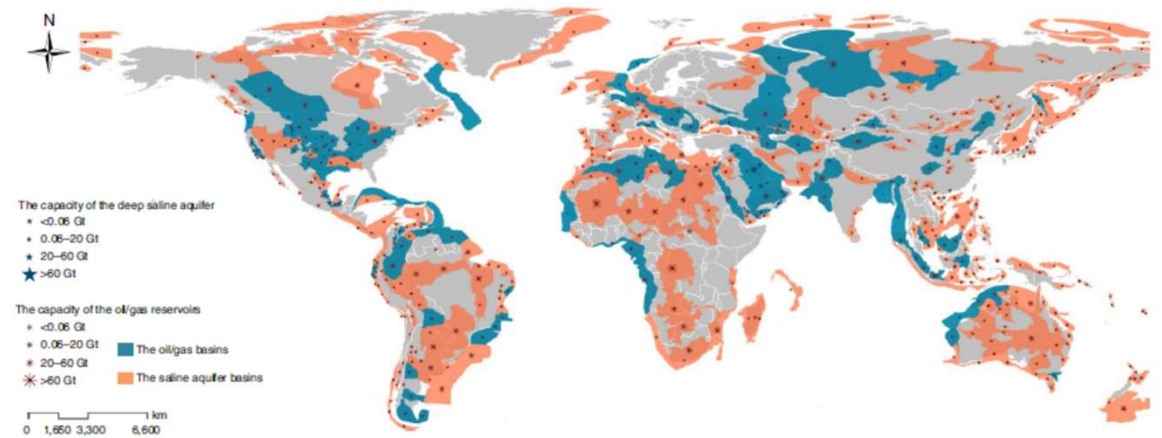
● A « good enough » hydrogen ?

- Aggressive discourse toward ever greener products
- When electrolysis has the reputation of being , « 0 emission »
- What place for hydrogen at reduced footprint, but above 3 t/t ?

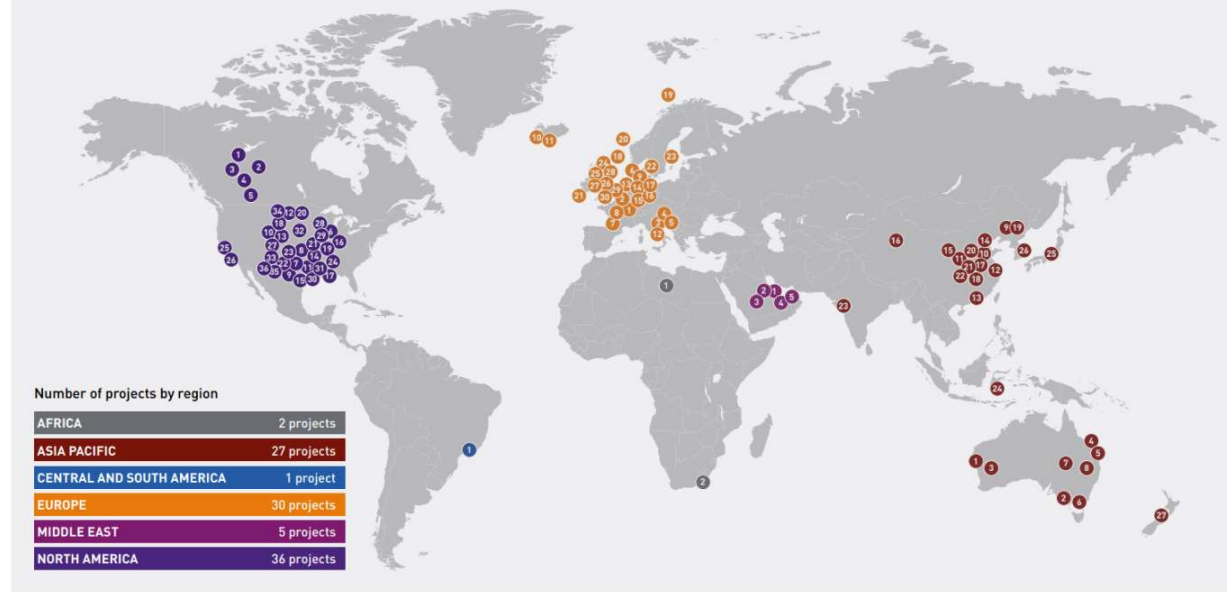


CCUS POTENTIAL

- Widely available around the world
- Still , logistics to reach the fields is key
- Actual projects are much rarer...
 - And largely dependant of O&G industry (enhanced oil recovery)



Overview of existing and planned CCUS facilities



Source: Global CCS Institute and IOGP data

HYDROGEN FROM HYDROCARBON : METHANE RELATED EMISSIONS

● Methane is an extremely powerful greenhouse gas

- Especially short term : 80x at 20 years

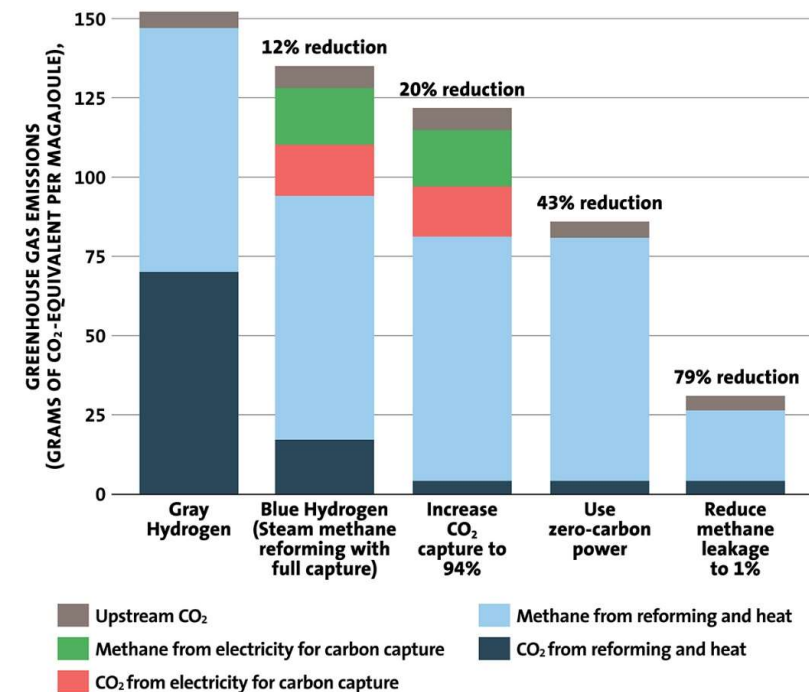
● Recent debates around methane leakage on extraction sites

- Estimations varying by a factor of 100...
- In truth, very site and process dependant
- US shale gas has a poor footprint, North Sea gas much better

● Important consequences for hydrogen

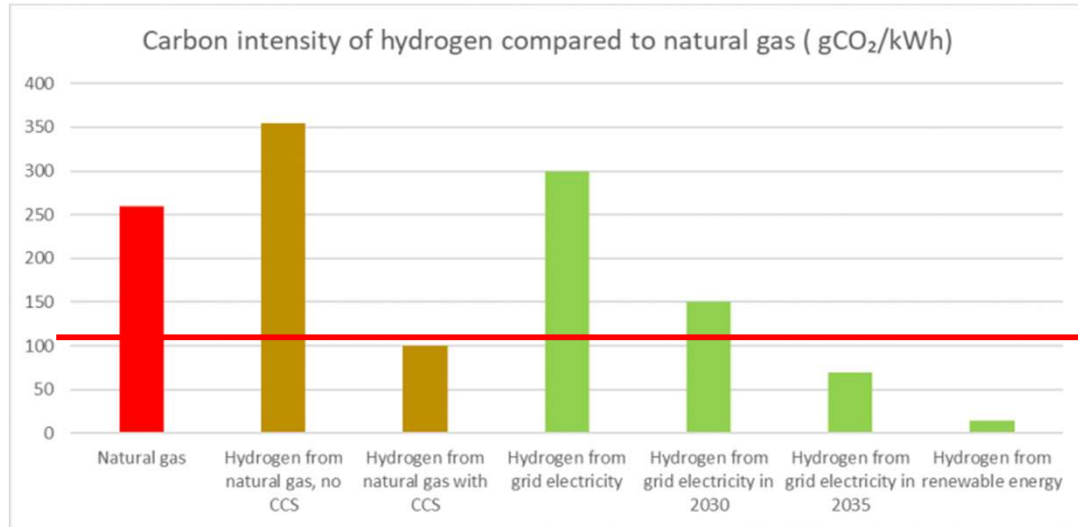
- Methane footprint can kill all by itself the hydrogen footprint if poor
- Difficult to take footprints at face value...

FIGURE 1A: LIFE-CYCLE HYDROGEN PRODUCTION EMISSIONS, 20-YEAR TIME FRAME

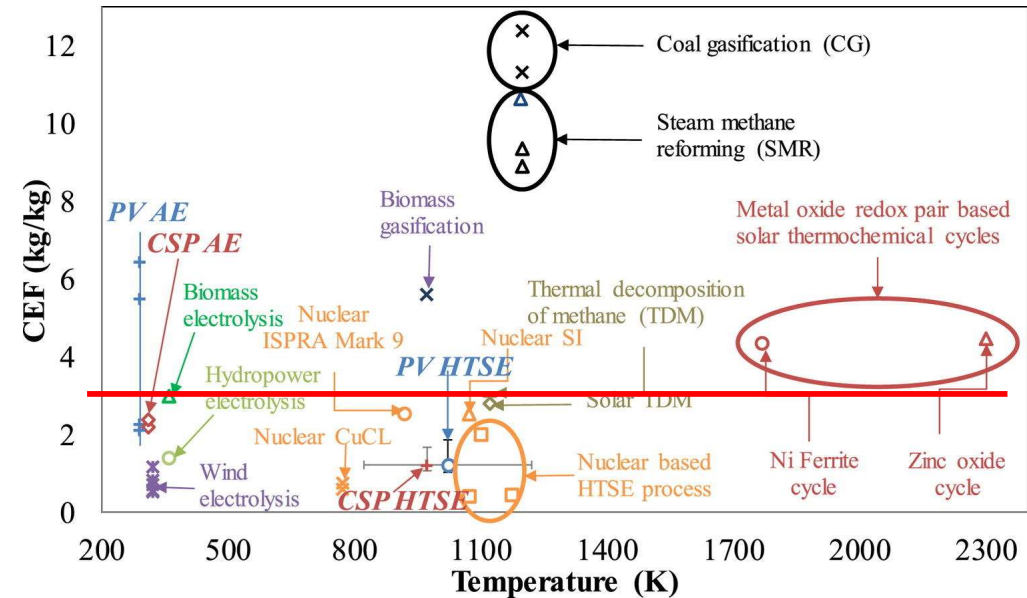


SOURCE: Howarth and Jacobson (2021)

HYDROGEN FROM ELECTROLYSIS : ELECTRICITY FOOTPRINT



Friends of the Earth Institute

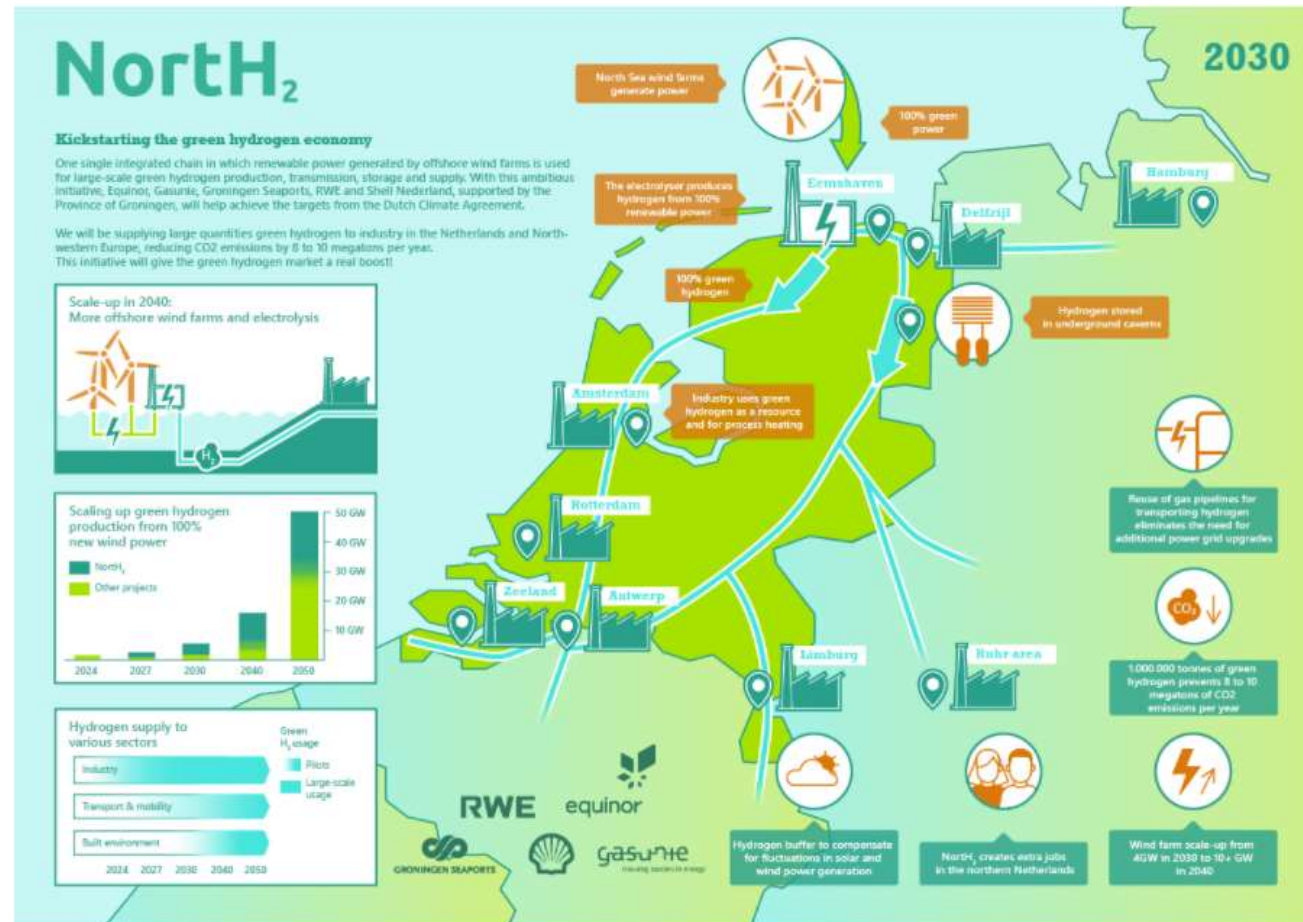


Yadav and Banerjee 2020 Applied energy

- Any « green » electricity will do
 - Wind and hydropower is better than solar, though
- The grid is not there...
 - And won't be there any time soon

HOW DO YOU GUARANTEE GREEN ELECTRICITY ? DEDICATED CAPACITY

- Local hub, with dedicated power source
- No balance of intermittence, so you better still be plugged to a grid

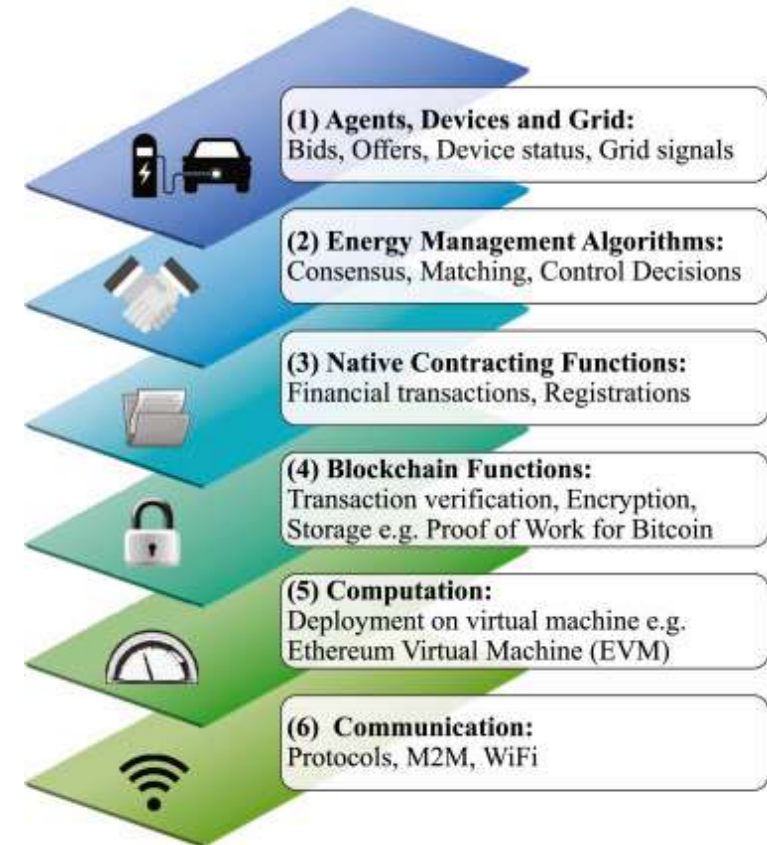


Source: RWE

LAST BUT NOT LEAST : HYDROGEN DIRECT FOOTPRINT

- Plug on the grid, get the electricity you want
 - The producer divide its production in lots
 - Lots are sold separately
 - Legit if the same green lot is not sold two times
- Technology helps if you want to make it flexible
 - Communication, blockchains...
- Does not make anything greener...
 - Won't be cheap to get the best lots
 - Who get to consume the junk electricity ?

6 Layers of Smart Contracting in Energy Systems



Kirli et al. 2022 Renewable and Sustainable Energy Reviews

LAST BUT NOT LEAST : HYDROGEN DIRECT FOOTPRINT

- Recent polemic around H₂ greenhouse potential
 - Complex mechanism involving the stabilisation of methane in the atmosphere
 - Studies made a huge polemic stating H₂ greenhouse potential was 200x of CO₂ !
 - Yes, but « immediate potential », which does not make sense
 - Latest studies : 10x time the potential of CO₂ at 100 years, the current benchmark
- Current leak practices are not up to the task
 - On average, electrolyzer leaks 9.4%, logistics up to 20 %...
 - 30% leakage = equivalent footprint 3 tCO₂/tH₂ = you are not green
- In reality : a more mature industry will (and have to) leak less
 - Larger, more continuous electrolysis could leak as little as 0.5%
 - Logistics needs to benchmark methane wich leaks < 1%



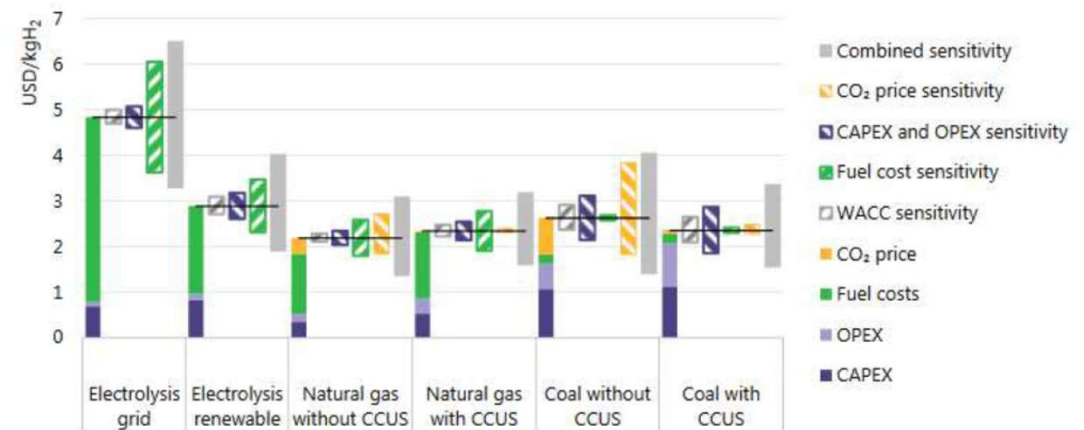
NTS



SSE

2 \$/KG : « AFFORDABLE HYDROGEN »

- Considered the current baseline price
 - Centralized large SMR unit
 - Methane at 7 \$/MBTU or 20 \$/MWh
- Main argument for blue hydrogen : it is not much more expensive
 - 50/150 \$/t of CO₂ treated
 - 0.5 to 1.5 \$/kg price premium
- No logistics in those calculations !
- When is electrolysis going to deliver at 3 or even 2 \$/kg ?



Notes: WACC = weighted average cost of capital. Assumptions refer to Europe in 2030. Renewable electricity price = USD 40/MWh at 4 000 full load hours at best locations; sensitivity analysis based on +/-30% variation in CAPEX, OPEX and fuel costs; +/-3% change in default WACC of 8% and a variation in default CO₂ price of USD 40/tCO₂ to USD 0/tCO₂ and USD 100/tCO₂. More information on the underlying assumptions is available at www.iea.org/hydrogen2019.

Source: IEA 2019. All rights reserved.

STABLE METHANE PRICES, REALLY ?



« SOVEREIGN HYDROGEN »

- But there is a crisis right now !
 - But are you sure we are not running into the next one ?
- A matter of national sovereignty
 - States don't like to be dependant for something as crucial as energy
 - At least in Europe : very severe clampdown on any technology which is natural gas dependant
- Electrolysis is much more reassuring on this point
 - Locally produced (possibly)
 - Stable price defined at commissioning (for offgrid systems)



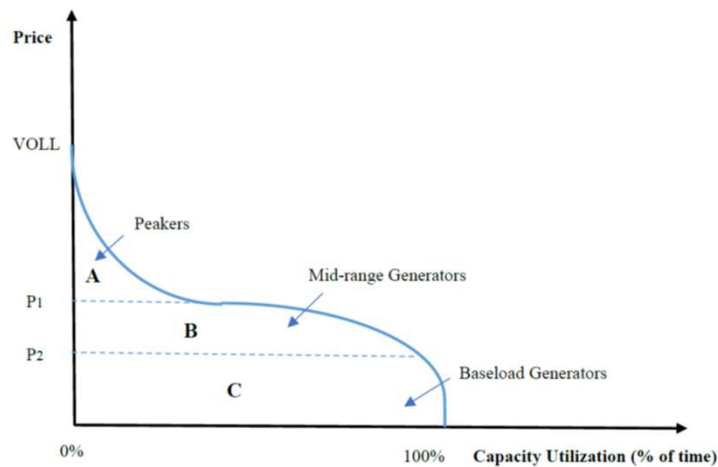
Wall Street Journal



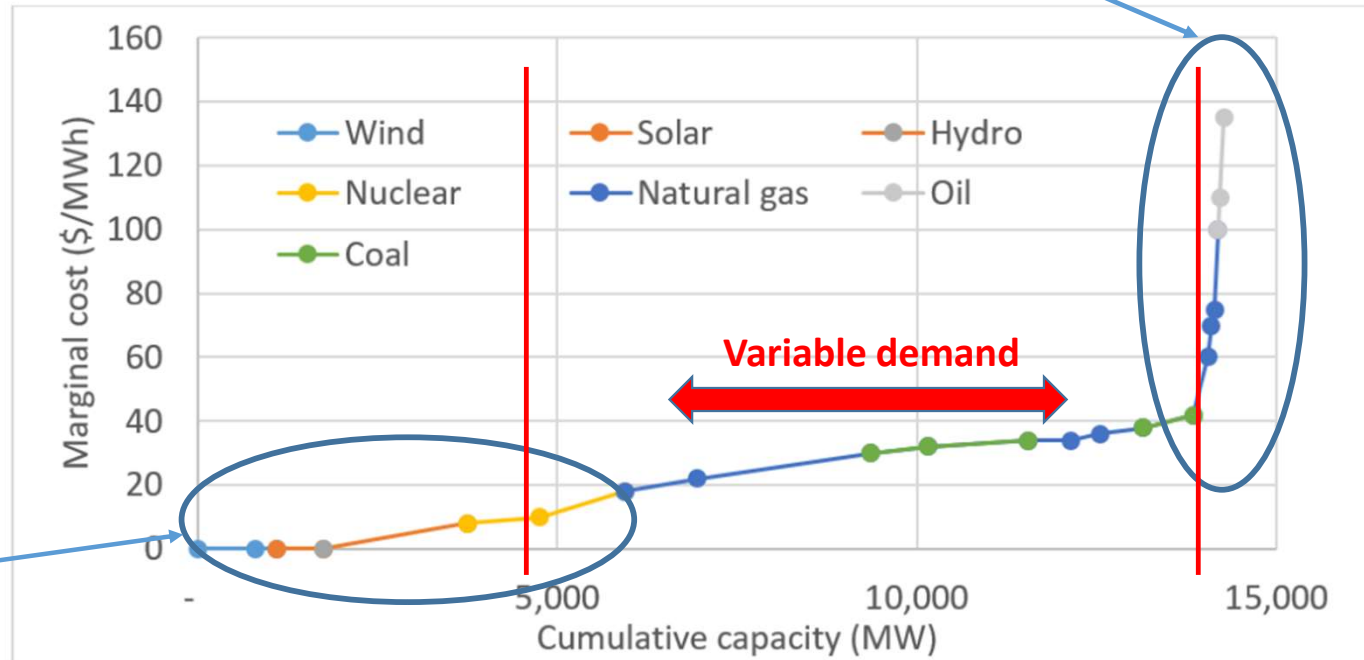
Euronews

CHEAP ELECTRICITY FOR HYDROGEN

- The plan : produce during peak production
 - You are solving the intermittence problem
 - And the electricity is cheap



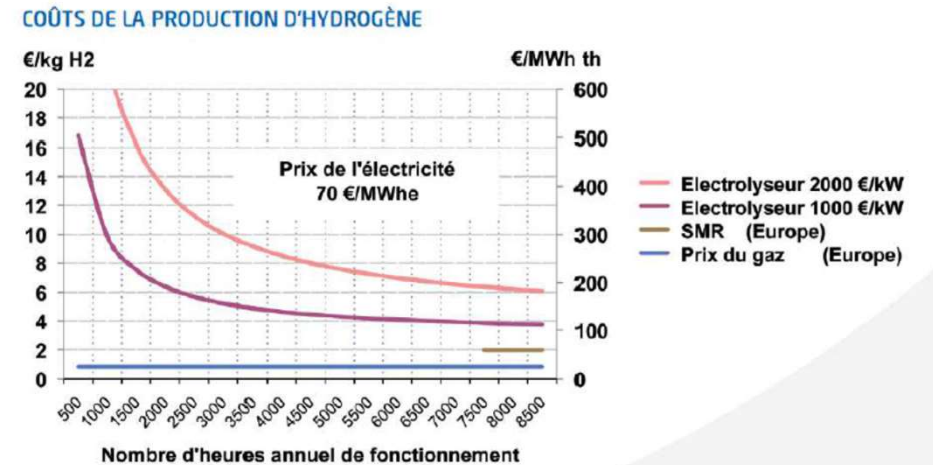
Low marginal cost : the place you want to operate producing hydrogen



Peak prices : key to the price you pay if you want a regular supply

CURRENT PICTURE : KILLED BY BOTH ELECTRICITY PRICES AND CAPEX

- CAPEX are currently too high to allow for that...
 - It is actually cheaper to run full-time with expensive electricity !
- Baseline electricity prices are far from allowing a 2\$/kg hydrogen
- And current cost struggle to go below 10 \$/kg



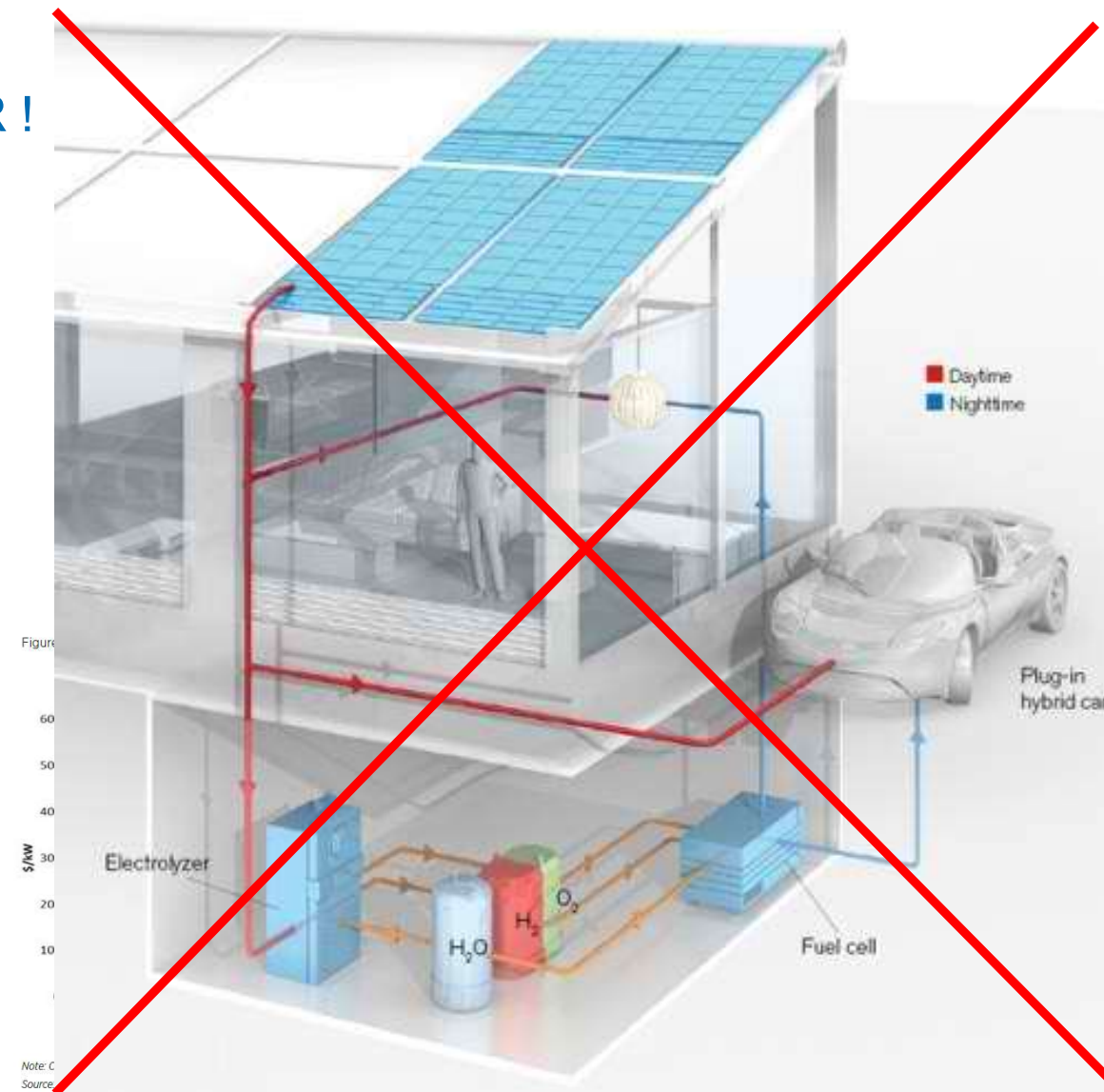
COÛTS DE PRODUCTION DE L'HYDROGÈNE PAR ÉLECTROLYSE SUIVANT DIVERS SCÉNARIOS

Scénario		1	2	3	4	5	6
Coût de l'électrolyseur	€/kW	2 000	2 500	800	800	800	800
Rendement électrolyse		60 %	50 %	80 %	80 %	80 %	80 %
Production annuelle	kWh	7 000	2 000	2 000	1 000	500	7 000
Coût de l'électricité stockée	€/MWh	70	70	70	140	0	60
Coût de l'H2 produit	€/kg	7,0	18,3	6,1	12,2	10,5	3,7
	soit \$/MBtu	68	177	59	118	102	36
	soit €/MWh	178	463	154	309	267	94

Source : CGSP

WE NEED CHEAPER ELECTROLYZER !

- And they are coming !
 - Price could go as low as 200 \$/kW
- But the main driver is economy of scale
 - Massive mega-farms
- Important conclusions regarding processes :
 - Intermittent production become relevant !
 - But you won't get it in your backyard



THE LONG TERM PICTURE ?

	2020				2050			
	Alkaline	PEM	AEM	SOEC	Alkaline	PEM	AEM	SOEC
Cell pressure [bar]	< 30	< 70	< 35	< 10	> 70	> 70	> 70	> 20
Efficiency (system) [kWh/kgH ₂]	50-78	50-83	57-69	45-55	< 45	< 45	< 45	< 40
Lifetime [thousand hours]	60	50-80	> 5	< 20	100	100-120	100	80
Capital costs estimate for large stacks (stack-only, > 1 MW) [USD/kW _{el}]	270	400	-	> 2 000	< 100	< 100	< 100	< 200
Capital cost range estimate for the entire system, >10 MW [USD/kW _{el}]	500- 1000	700- 1400	-	-	< 200	< 200	< 200	< 300

Note: PEM = polymer electrolyte membrane (commercial technology); AEM = anion exchange membrane (lab-scale today); SOEC = solid oxide electrolyzers (lab-scale today).

Based on IRENA analysis.

RENEWABLE ELECTRICITY PRICE

- Intermittent production : direct plug to a renewable source realistic
- Very exciting perspectives in terms of electricity price !
 - But strong debates on the realism of those estimates

Conditions for a 2\$/kg H₂ :

- Electrolyzer at 200 \$/kW, running 2000 hours a year (1/4 of the time) : 0.5 \$/kg
- Electricity at 20 \$/MWh : 1 \$/kg

Logistic costs 0.5 \$/kg ?



NUCLEAR-BASED ELECTRICITY FOR HYDROGEN ?

● How cheap/expensive is nuclear electricity ?

- Countries with an existing, amortized park pay relatively low marginal prices : 40-50 \$/MWh
- Hinkley Point (UK) new reactor : 120 \$/MWh
- More controlled new generation : 60-70 \$/MWh

● Not cheap, but its a baseline, steady supply

- Very nice for a grid which relies heavily on renewables

● Good for hydrogen today, not so much tomorrow

- Countries with a large nuclear park are the only one today producing low-carbon hydrogen from the grid
- Too expensive in the future to be of interest for gneralized H₂ production

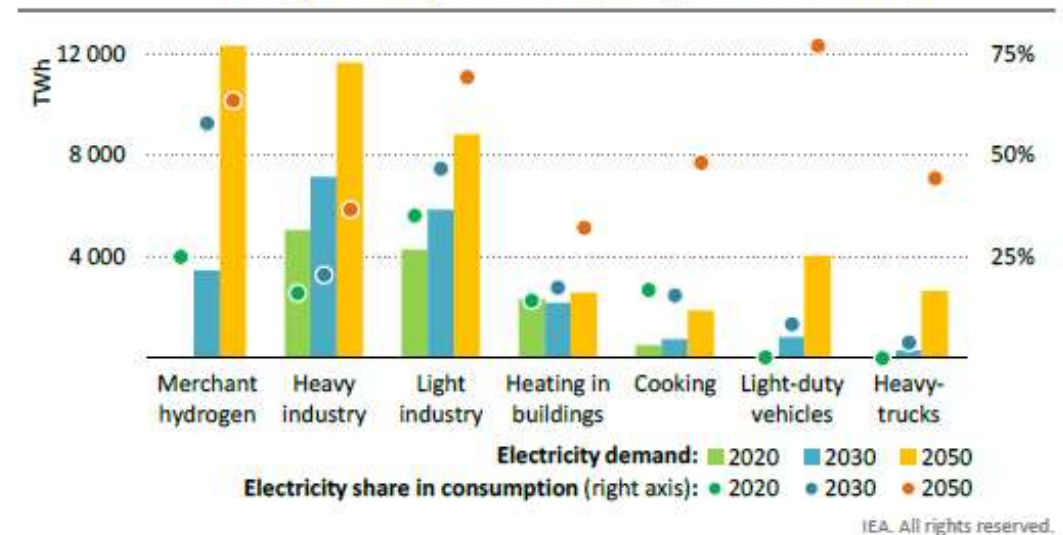


Financial Times

EVERYBODY WANTS GREEN ELECTRICITY...

- Electricity needs are up, everywhere
 - Mobility is there, but only a small share
 - Massive industrial swift to electricity
- Hydrogen comes on top of that...
 - To become the first consumer in 2050 ?
- Cheap green electricity, really ?
 - Production costs are going down, sure...
 - But we are in for a massive offer crisis

Figure 2.16 ▶ Global electricity demand and share of electricity in energy consumption in selected applications in the NZE

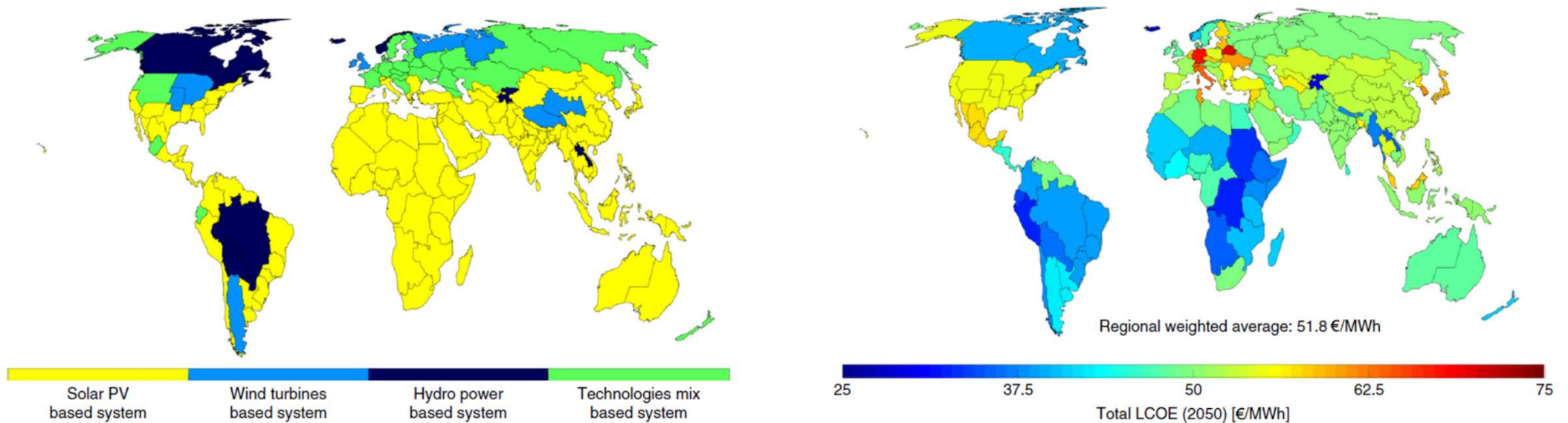


Global electricity demand more than doubles in the period to 2050, with the largest rises to produce hydrogen and in industry

Notes: Merchant hydrogen = hydrogen produced by one company to sell to others. Light-duty vehicles = passenger cars and vans. Heavy trucks = medium-freight trucks and heavy-freight trucks.

Source IEA 2021 –
Net Zero by 2050

YOU WON'T FIND 20 \$/MWH ELECTRICITY EVERYWHERE



Bogdanov et al. 2019 Nature Communications

● But that will be for next chapter...

CASE STUDY

YOUR TIME TO WORK !

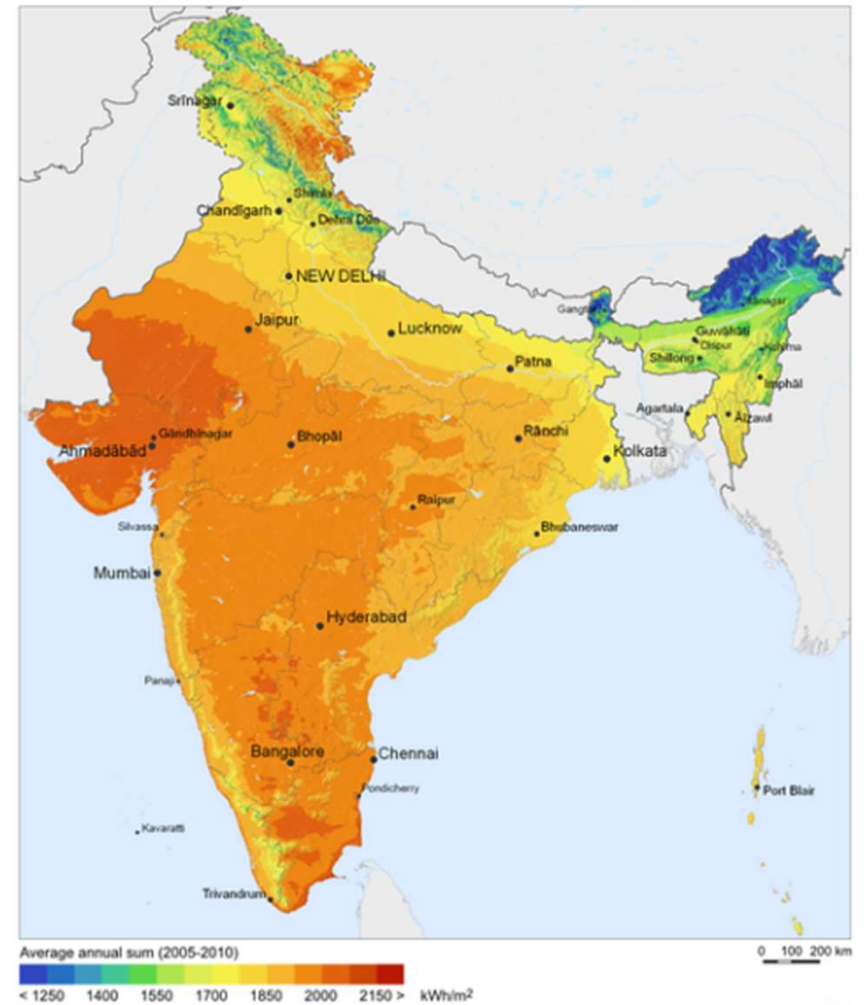
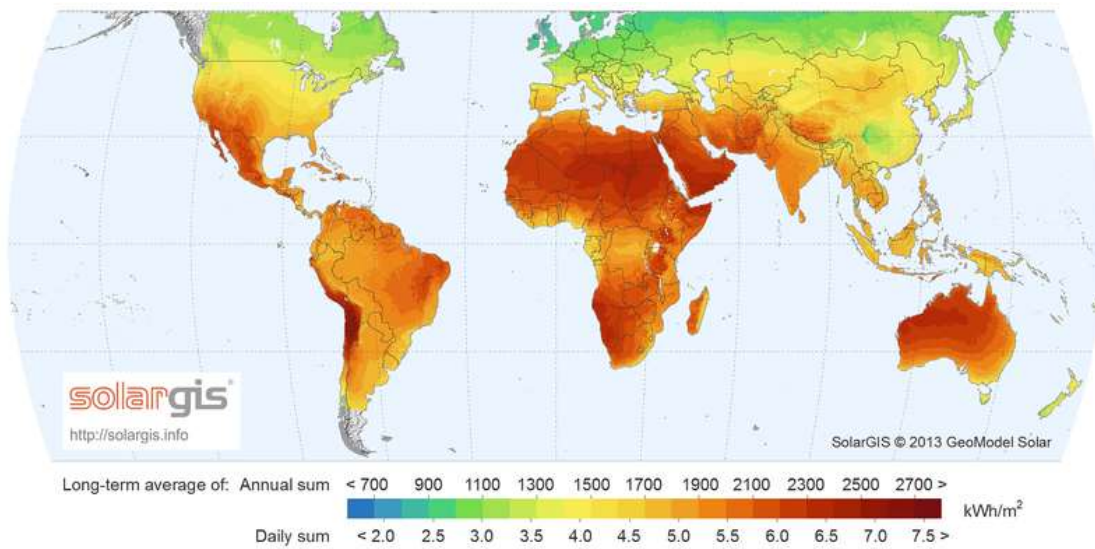
- India's hydrogen strategy : offer

- Identify the opportunities in India to produce low-carbon hydrogen
 - Type of production ?
 - Geography ?

- I don't have a correction !

- You are the experts concerning India, not me...
- But I have a few éléments and we can debate

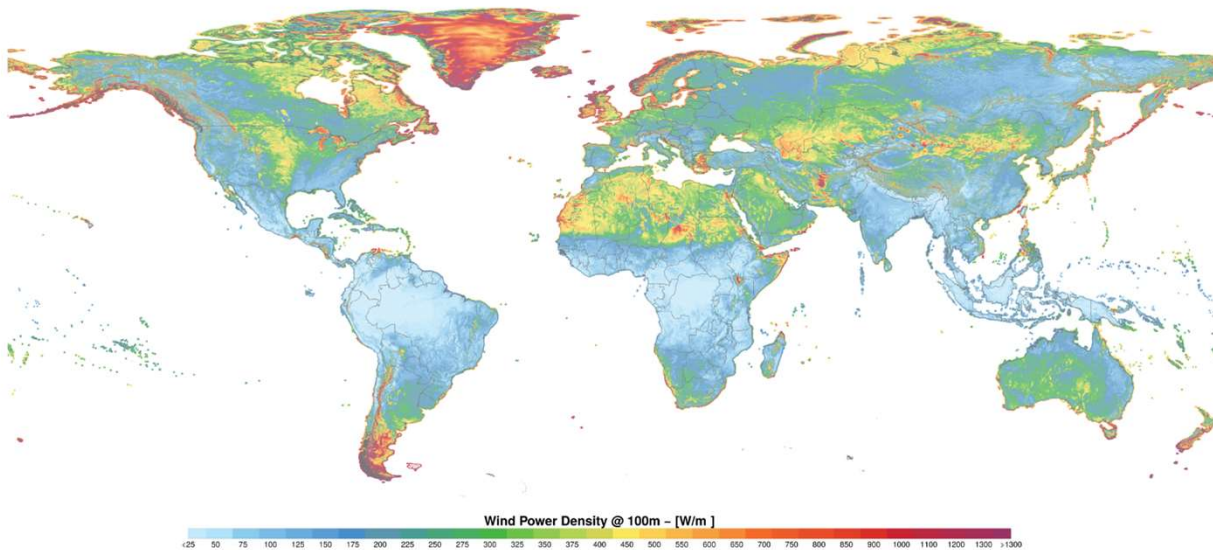
RENEWABLE POTENTIAL OF INDIA : SOLAR



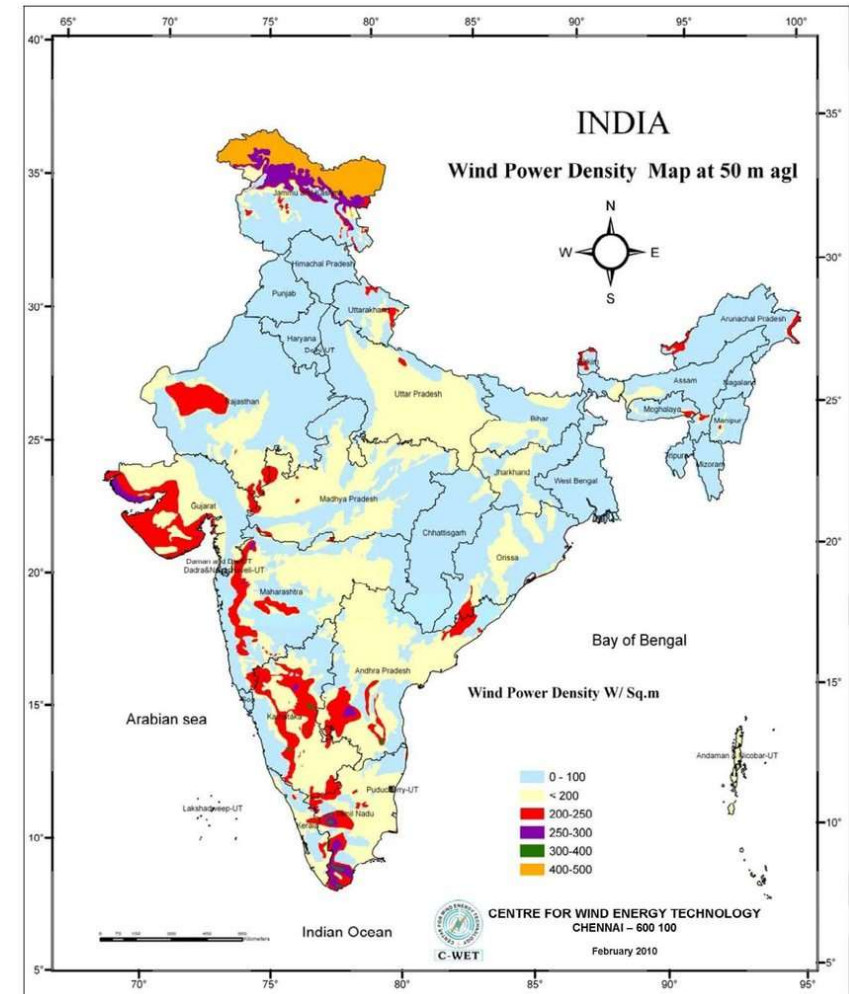
RENEWABLE POTENTIAL OF INDIA : WIND

ONSHORE & OFFSHORE WIND RESOURCE MAP

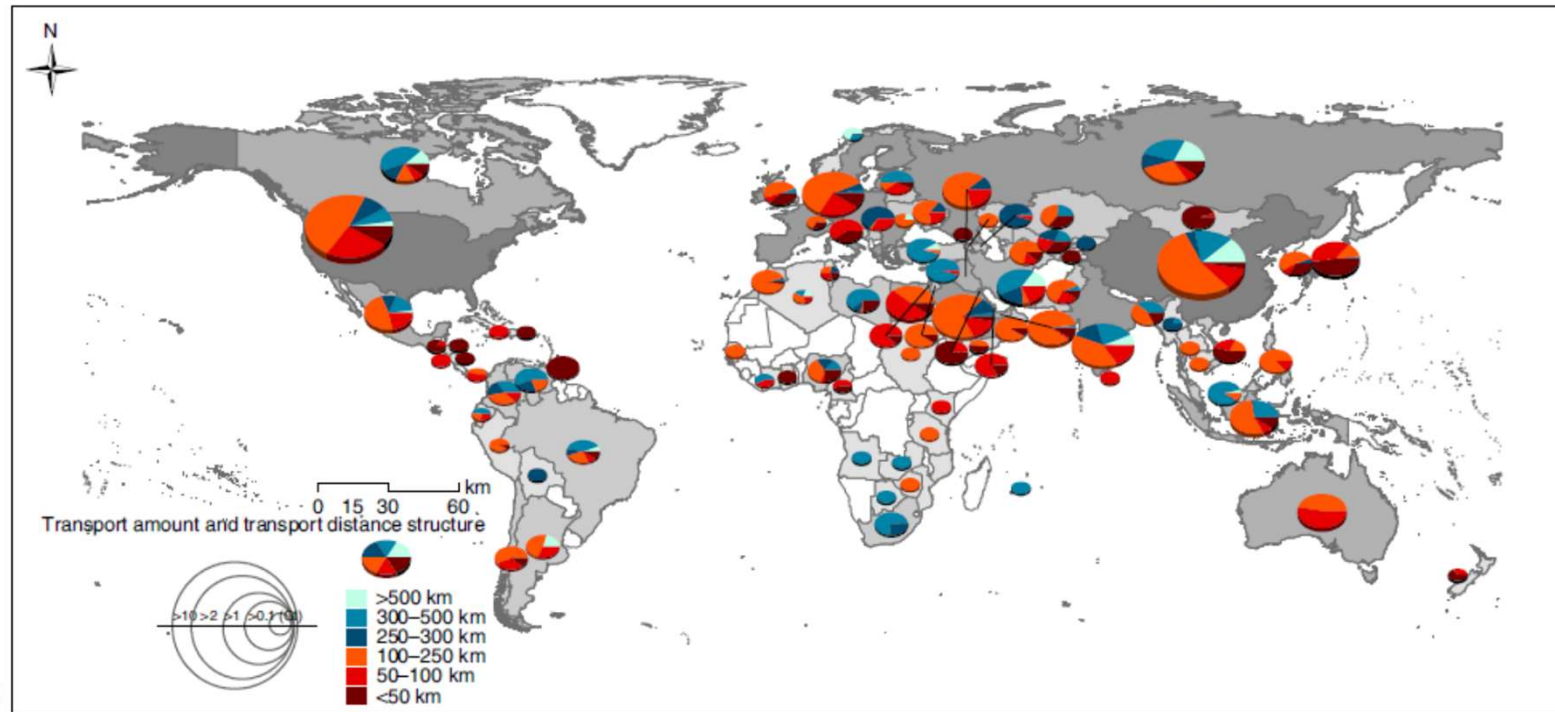
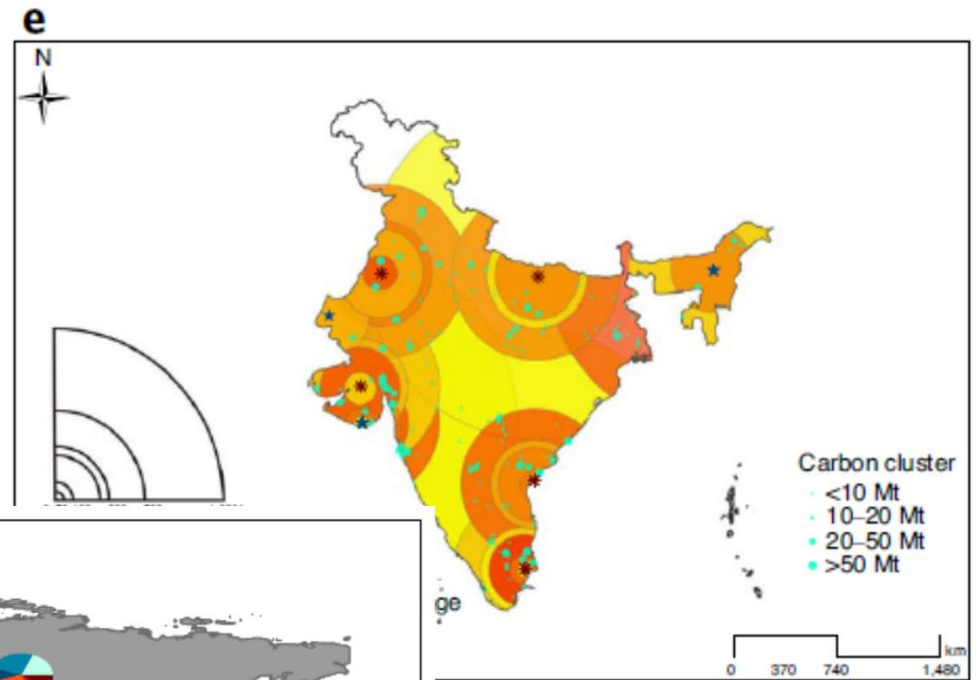
WIND POWER DENSITY POTENTIAL



This map is published by the World Bank Group, funded by ESMAP, and prepared by DTU and Vortex. For more information and terms of use, please visit <http://globalwindatlas.info>



CCUS POTENTIAL OF INDIA



WRAP-UP

- India is a sun country (and a good one)
 - Solar likely dominant in the renewable mix
 - Solar thermal on the table due to local conditions
 - With SOFC electrolysis or thermochemical splitting ?
 - Doesn't mean hydro or wind can't be locally deployed
- India's CCUS potential is decent but...
 - India has no domestic methane production
 - No project started yet
 - Coal gasification ?
- India is a big country
 - We will talk about logistics in the next chapter...





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