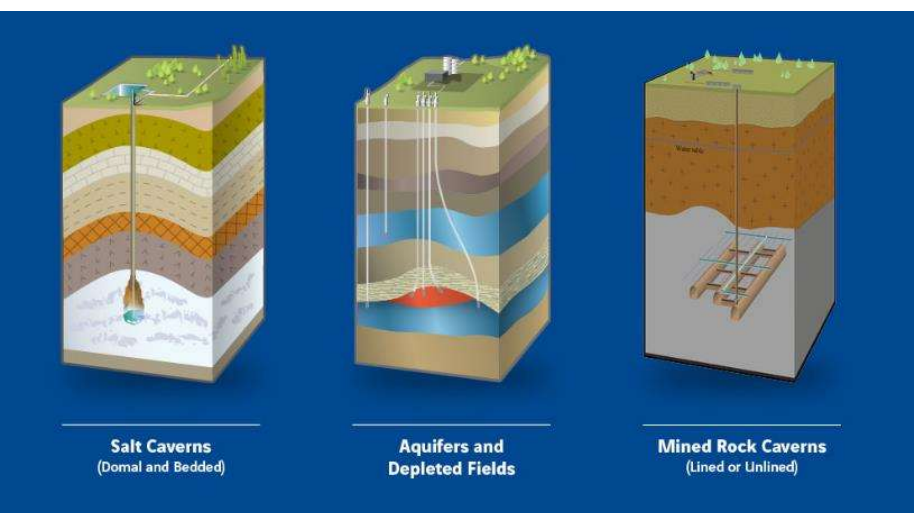




# HYDROGEN FOR THE ENERGY TRANSITION : WHY ? HOW ? PART III : LOGISTICS

Dr Vivien Esnault – IFPEN (France)

GeoIndia 2022 – Continuing Education Courses October 12th 2022



## PROGRAM OF THE COURSE (2/2)

### ● Part 3 : Toward an hydrogen network

- Technologies to transport and store hydrogen
- Safety and quality issues
- Case study : Your hydrogen strategy for India

### ● (Tea break)

### ● Part 4 : Hydrogen underground

- Underground ressources for hydrogen (native hydrogen and in-situ production)
- Storing hydrogen underground
- Questions – open debate

## PART 3 : TOWARD AN HYDROGEN NETWORK

### ● Storing hydrogen, at all scale

- Storing hydrogen in your vehicle
- Storing hydrogen on a small site
- Storing strategic quantities of hydrogen

### ● Transporting hydrogen

- Today's reality
- Network needed for the hydrogen economy

### ● Safety & quality

### ● Case analysis : your hydrogen strategy for India



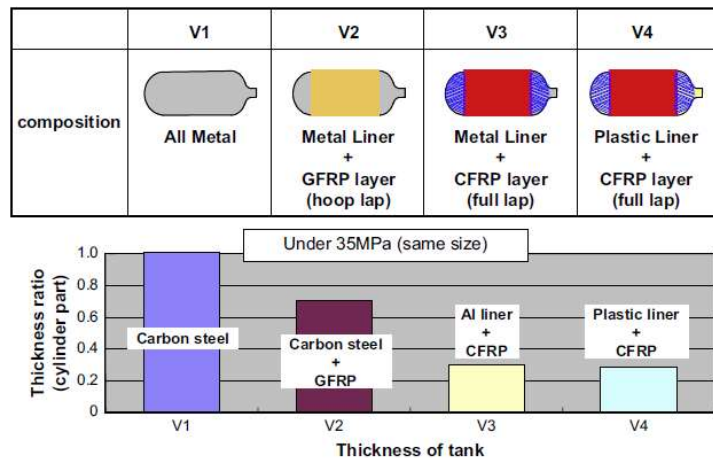
# STORAGE

# STORING HYDROGEN ON A VEHICLE : A QUESTION OF MASS

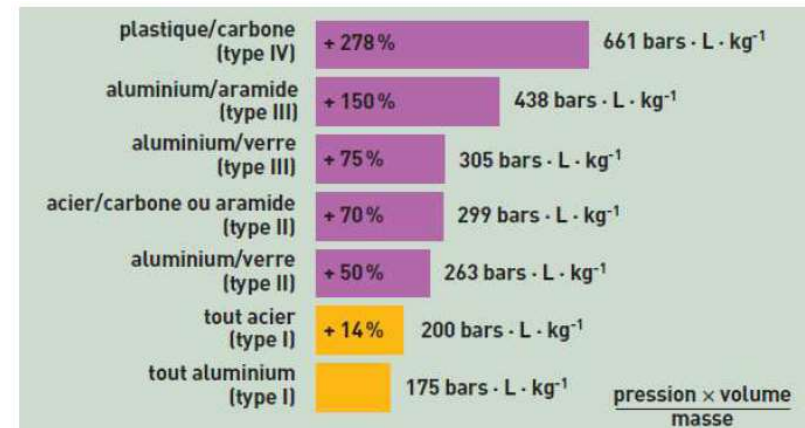
- Hydrogen is light, the reservoir isn't
  - High pressure → demanding reservoir design
- New standard : type 4 reservoir
  - Completely proven at 400 bar, up to 700 bar
  - 100% polymer design
  - >5% of hydrogen mass/ reservoir system mass



Wikipedia (Honda vehicle)



Mori et al. 2009, International Journal of Hydrogen Energy



France Hydrogène



# TYPE IV RESERVOIR : HOW DOES IT WORK

## ● Liner

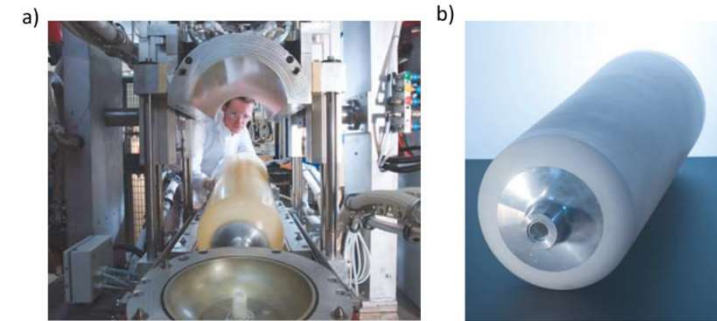
- Molded (ex : rotomolding) in specific thermoplastic
- Responsible for controlling leakage

## ● Composite enveloppe

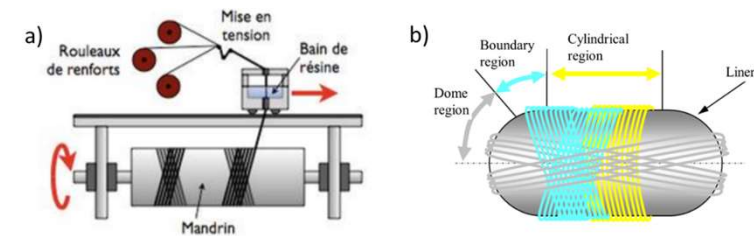
- Woven, carbon-fiber composite
- Responsible for structural integrity

## ● Boss, valve...

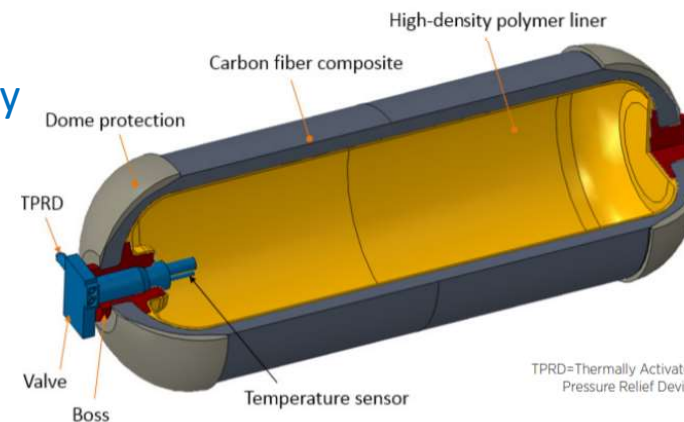
- Interactions between components is key



CEA



Quantum Technologies



US Department of Energy

TPRD=Thermally Activated  
Pressure Relief Device



## TYPE IV RESERVOIR : LINER INTEGRITY

### ● Permeability

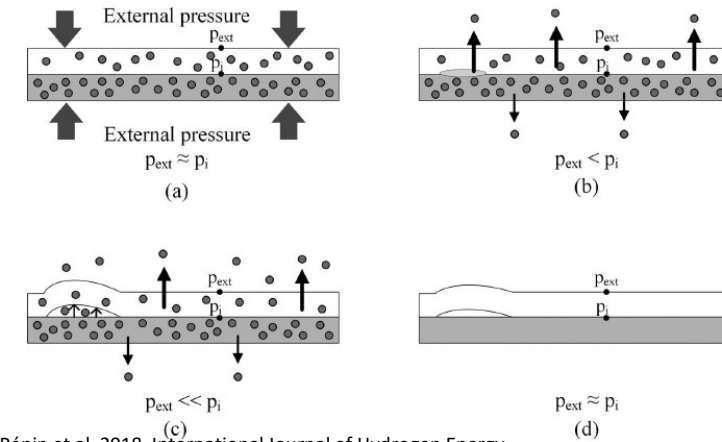
- H<sub>2</sub> penetrates and migrates (quite) easily in polymers
- Limit of  $10^{-5}$  loss per hour (order of magnitude) : 1% in a month

### ● Blistering

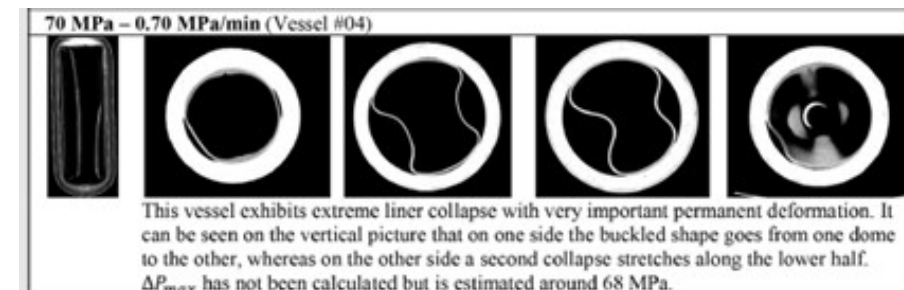
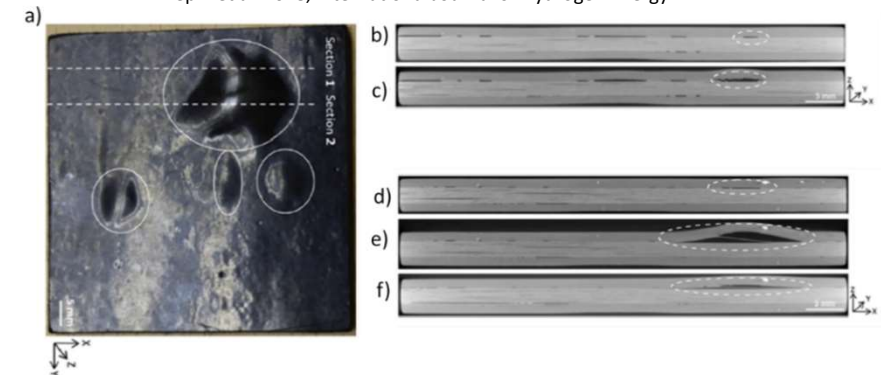
- H<sub>2</sub> accumulates at high pressure in the structure
- If pressure drops, it might not have time to escape
- Gaseous bubbles form, and eventually blisters

### ● Collapse

- Same mechanism as blistering, at the liner/enveloppe interface
- Liner eventually deform and collapse



Pépin et al. 2018, International Journal of Hydrogen Energy



Blanc-Vannet et al. 2019, International Journal of Hydrogen Energy

## TYPE IV RESERVOIR : MECHANICAL ENVELOPPE INTEGRITY

### ● Wild range of tests

- Burst
- Impact
- Bullet shot
- Bonfire
- Fatigue
- ...

### ● Correspond to real life difficult scenario

- All configurations of car crashes
- Fire in a confined space (tunnel...)

### ● « Leak before fail »



© Paweł Gasior's Web Site

Washington State University



OSTI

ifp Energies nouvelles



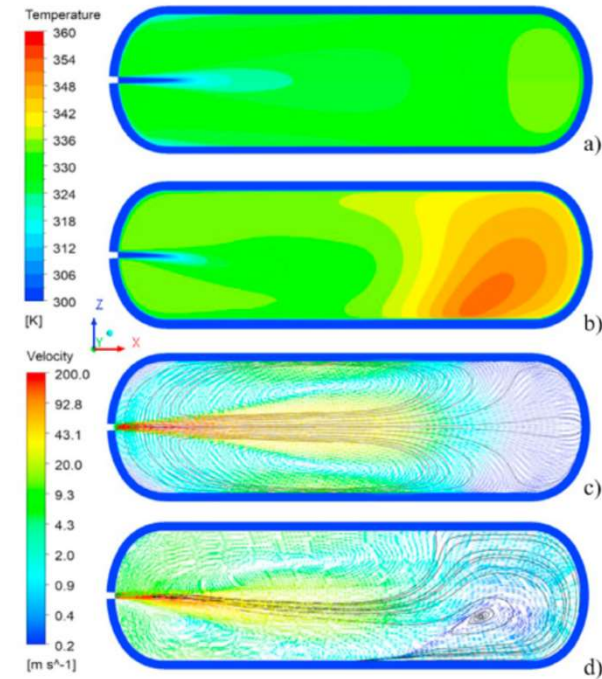
## TYPE IV RESERVOIR : REFUELING CONSTRAINTS

- Advantage of hydrogen : fast refueling

- However, compressing a gas generates heat
- 100°C of  $\Delta$  in typical operations
- Minimal refueling times for light vehicles : 5 min

- Important consequences on vehicle and station design

- Gas is refrigerated to -40°C before injection
- -40°C to 70°C : very demanding for polymers
- Cyclical stresses → material fatigue

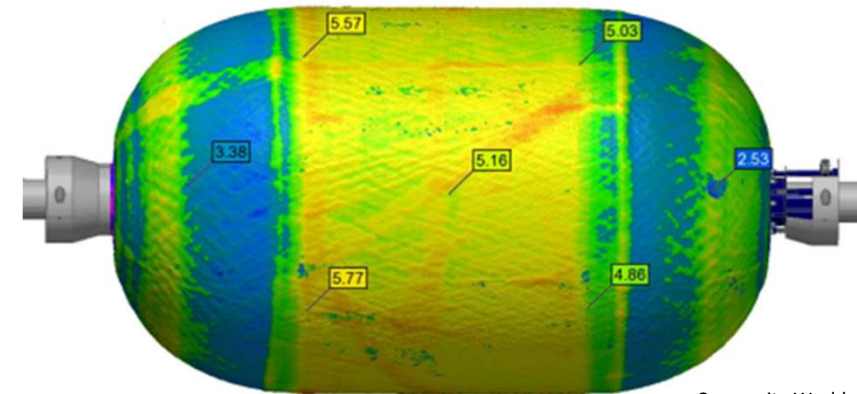


Li et al. 2021 International Journal of Hydrogen Energy



## FUTURE DEVELOPMENT IN RESERVOIRS ?

- No need for more pressure
  - 300-400 bar might be enough for a lot of applications
- Type 5 reservoirs ?
  - Get rid of liner, full composite
  - No additional capacity, but easier process
- Simplify process
  - Current process is very time and resource consuming
  - Partially responsible for FCEV high prices
- Adaptable shapes for the reservoir
  - Would ease vastly vehicle integration
  - Good luck to come out with anything else then a cylinder, though



Composite World

# LIQUID HYDROGEN STORAGE

- Denser than 700 bar gaseous

- Not so much, on a physics point of view (x1.7)
- Very much so, taking into account the reservoir which is lighter

- -253°C...

- LNG -160°C, liquid nitrogen -196°C, absolute zero -273°C
- Relatively unknown levels of cold (except niche industries)
- Energy consumption : +20% (33% of energy yield)

- Who will need it ?

- Airplanes : certainly
- Heavy trucks, naval vessels : perhaps
- Long range transport ? (road, naval)



LH2 - TANK SYSTEM



# LIQUID HYDROGEN : BOIL-OFF ISSUES

## ● The ortho/para issue

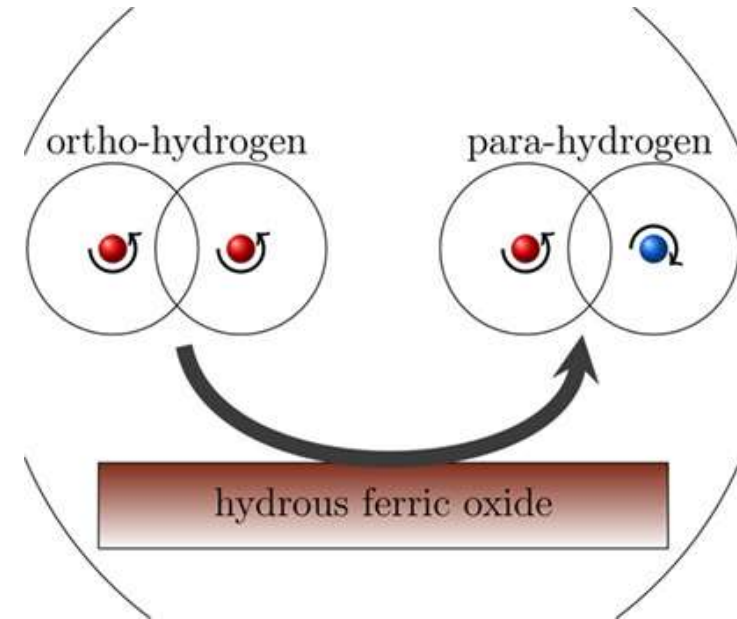
- « Ortho » and « para » are two quantum-spin related state of hydrogen
- At low temperature, hydrogen transitions from ortho to para, generating heat
- Liquid hydrogen needs to be forced into para beforehand

## ● Nevertheless, partial boiling can't be avoided

- In the 1% order of magnitude
- Depends of vessel size

## ● What do you do with boiled-off hydrogen ?

- Large vessels : keep the pressurized gas
- Small ones : vent it (safety ? carbon footprint?)



Donaubauer et al. 2019 Chemical Engineering Technology



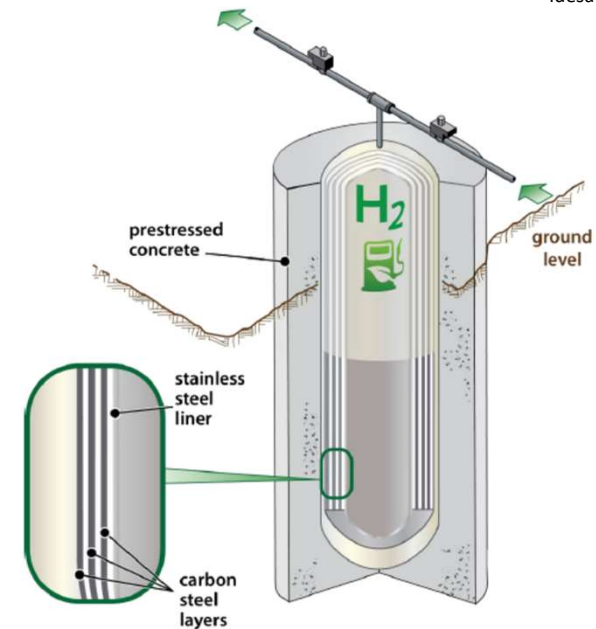


## FACILITY-SIZED, STATIONNARY STORAGE

- Less demanding than mobile applications
  - Density requirements not as important
  - More cost-sensitive
- No rocket technology needed
  - Working pressure : 100-200 bar ?
  - Additional pressure for mobility ? : only in a small buffer
  - All steel vessels mainstream
  - Buried, concrete vessels an interesting possibility
- Large quantities of gaseous hydrogen: a good idea ?
  - As hydrogen leaves industrial areas to urban areas ?
  - Already tensions about regulations in Europe



Idesa



US Department of Energy



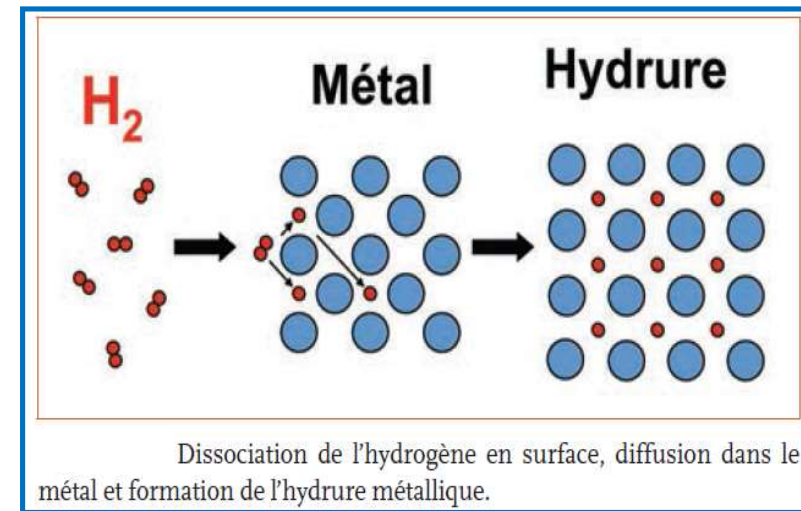
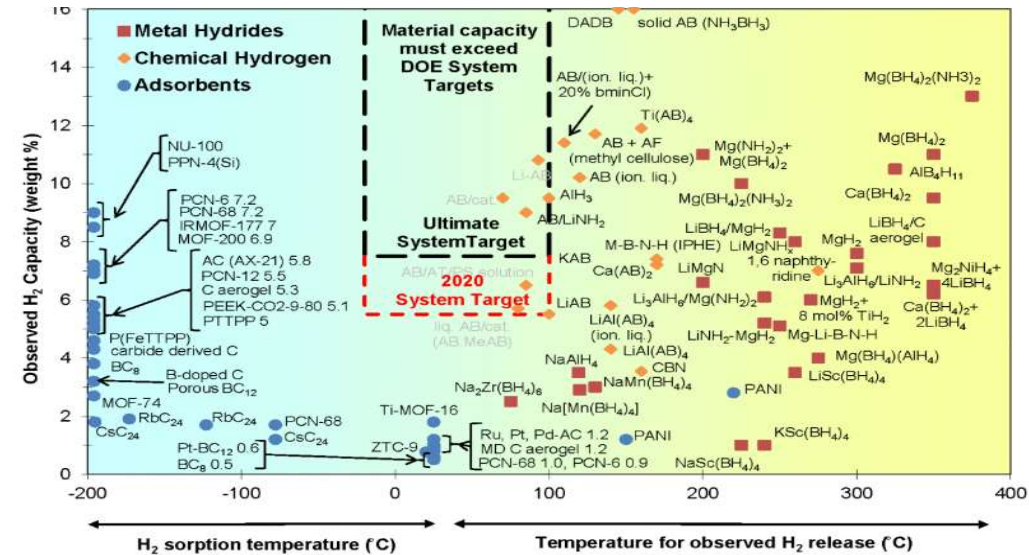
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## ● Metal hydrides

- Hydrogen can form complexes with a wide variety of metals
- Hydrogen stored in a dissociated form

- Complex to reach the targets

- Difficult to reach high levels of mass %
- The best systems only works at high T°
- Other issues, such as reaction speed, durability of cycling



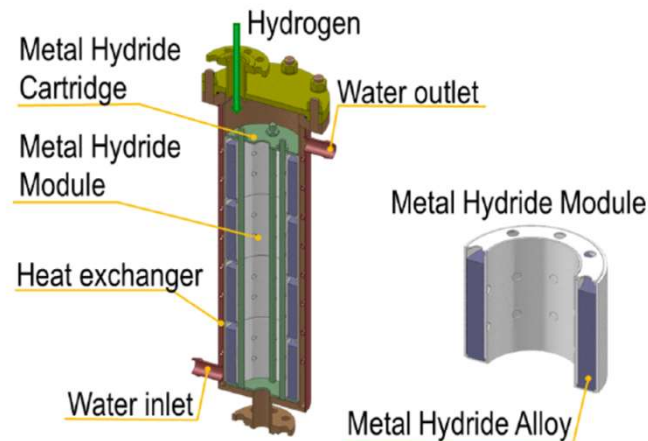
## METAL HYDRIDES : A FORMER GOOD IDEA ?

### ● Mobility usage : a lost battle ?

- Main drawback : heavy, and slow
- In the meanwhile, pressured gas options made dramatic improvement

### ● Still an interest for stationary applications ?

- Cycling speed issues and speed not so important
- Stability of the hydride could be very important for safety



Blinov et al. Energy System Research



US Department of energy



## OTHER SOLID STORAGE OPTIONS

### ● Adsorption storage

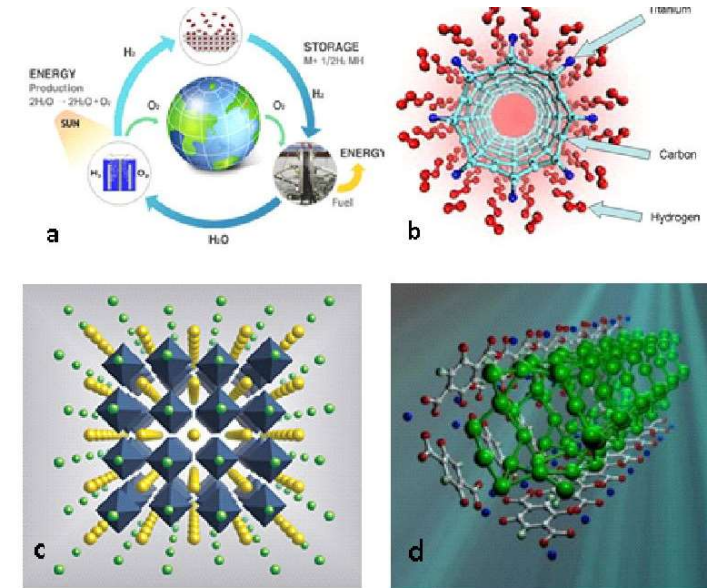
- Hydrogen stored in nanoporous structure
  - Zeolites, MOFs, nanotubes
- Hydrogen stored in its molecular form
- Dense only with high pressure or low T°

### ● Hydrogen hydrates

- H<sub>2</sub> like CH<sub>4</sub> can form ice-like structures with water
- Once again : forms significantly at extremely low T° and high pressure

### ● Liquid Organic Hydrogen Carriers

- Hydrogen stored in liquid organic molecules
- Speak about it later for transport applications



Wikipedia



## STORAGE WRAP-UP

### ● Vehicle level storage

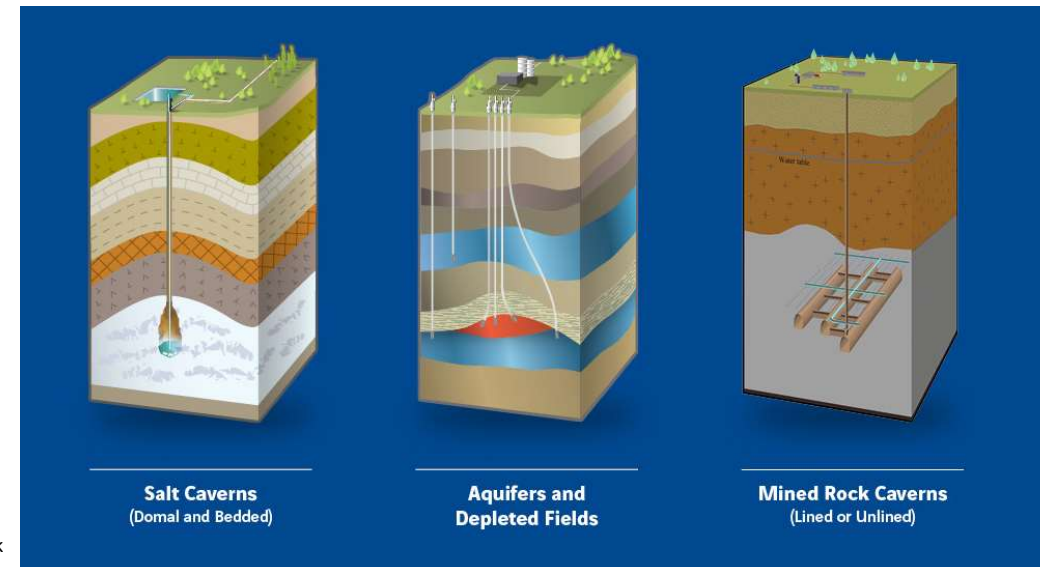
- Dramatic breakthrough of gaseous storage under pressure
- Liquid hydrogen an (expensive) option for denser storage

### ● Station level storage

- Technology ready, robust, and not very complex
- Potential interest for solid options, for safety ?

### ● Large-scale, strategic storage ?

- Underground storage of course
- Will be discussed in part IV of the course





# TRANSPORT



## WHY TRANSPORT HYDROGEN, AGAIN ?

### ● When you can produce it onsite ?

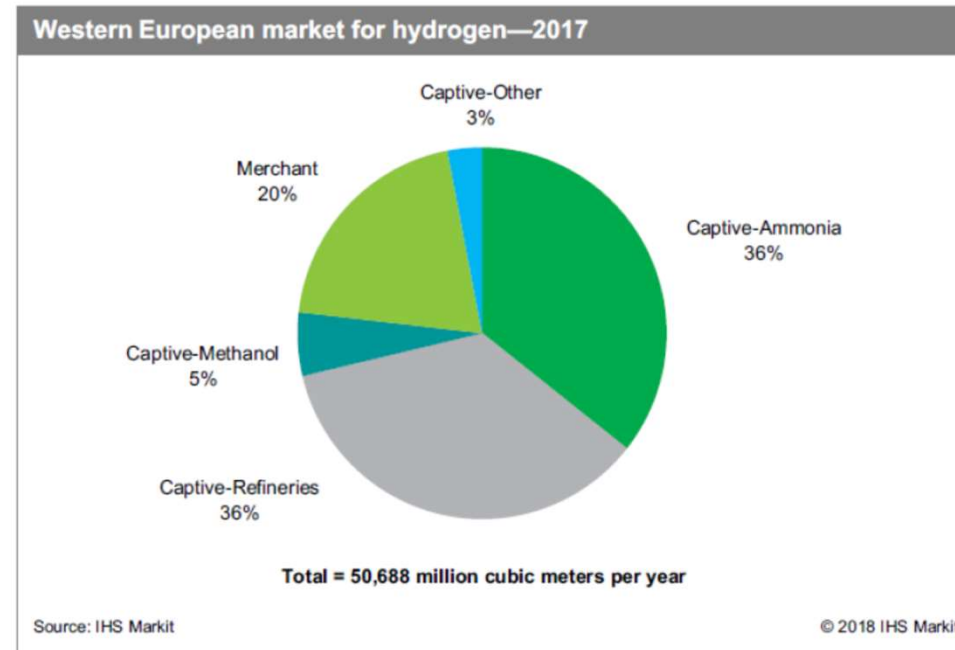
- Today, 80% of H<sub>2</sub> production is «captive»
- Coherent for large industrial users, using methane as a commodity

### ● Future usage

- Large industrial users are still going to be a thing
- Mobility usage are much more decentralized

### ● Future of production

- Premium to large production farms
- Driven by access to cheap green electricity, not proximity to an industrial site



Future hydrogen economy will see important transport of hydrogen

## TRANSPORTING HYDROGEN : TRUCKS

- Often the only solution to deliver hydrogen today
- Progressive increase of capacity per truck
  - Steel tubes : 380 kg per truck
  - Composite tube trailers : 560-900 kg per truck
  - Liquid hydrogen trucks : around 3 t per truck
- A compromise to be found
  - Longer range : strong interest in larger capacity
  - CAPEX rise sharply with more demanding transport



## THE ECONOMY OF TRUCK TRANSPORT

### ● Compressed gas trucks : short distance

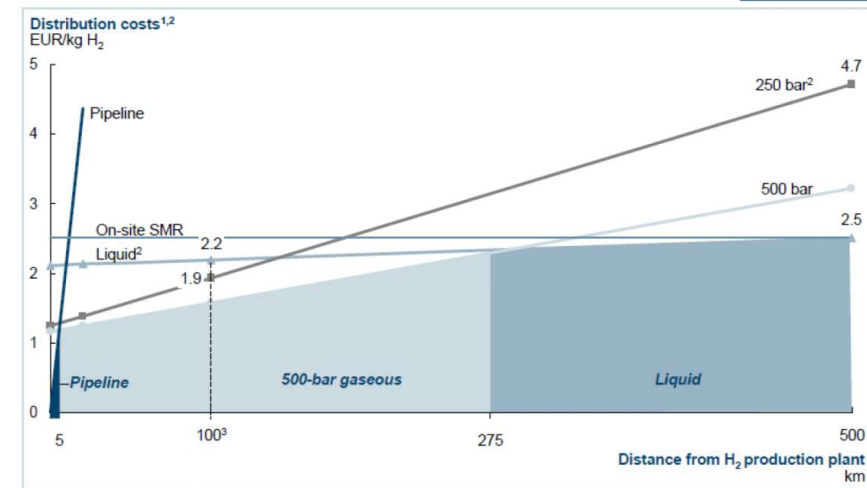
- Relatively low CAPEX
- Sharp increase of cost per km

### ● Liquid hydrogen truck

- Costly CAPEX and liquefaction OPEX
- Low cost per km

### ● Anyway : too expensive for the H<sub>2</sub> economy

- Current delivery price in station is 10\$/kg
- Technology not compatible with a 2\$/kg world



	Gaseous tube trailers		Liquid tankers	
Capacity	450 kg/day	1,000 kg/day	450 kg/day	1,000 kg/day
Hydrogen delivery and dispensing cost	\$9.46/kg	\$8.17/kg	\$11.35/kg	\$8.31/kg

US Department of Energy

## GAS IS BEST TRANSPORTED IN PIPE

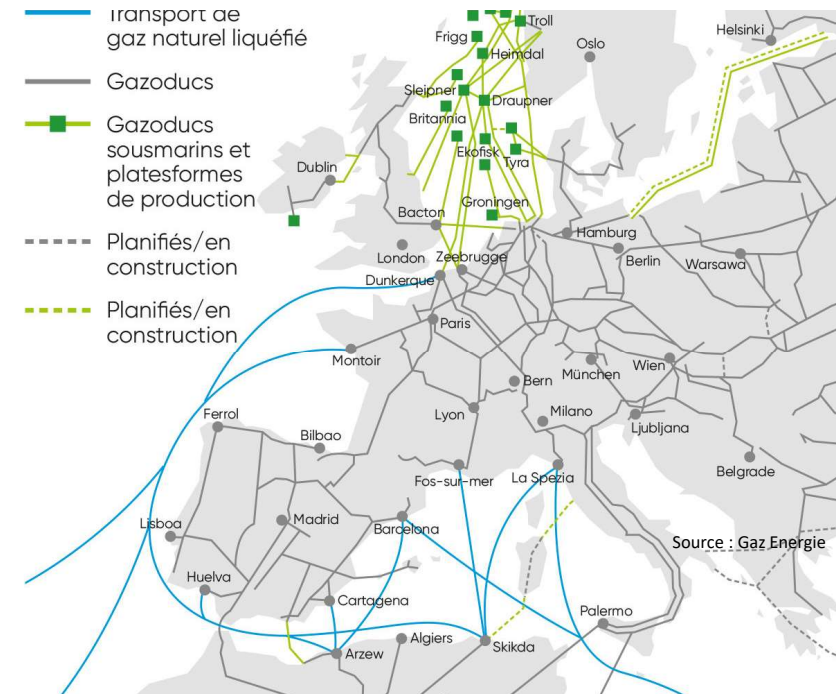
### ● The natural gas benchmark

### ● A few figures for the USA :

- 4.5 millions km of pipelines
- 70 millions households delivered
- Several thousands of billions \$ of investment

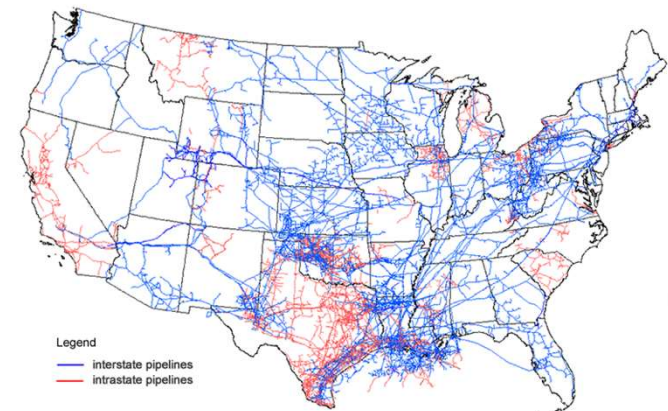
### ● More than a benchmark, an opportunity

- The gas industry has a strong interest in recycling its assets



Gaz Energie

Map of U.S. interstate and intrastate natural gas pipelines



Source: U.S. Energy Information Administration, About U.S. Natural Gas Pipelines

IEA

**Energies nouvelles**

# HYDROGEN TRANSPORT IN PIPELINE

- Infrastructures are similar to natural gas pipeline

- Typical diameter: 300 mm
- Pressure : 10-100 bar
- Flow velocity up to 40 km/h
- Meaning 65 kg/s with methane...
- ... but only 7 kg/s with hydrogen !

- Mostly underground pipeline: buried for safety issue

- Key equipments:

- Boosting stations : recompress gas every 200 km
- Maintenance valves : every 8 to 30 km
- Distribution stations...



Source : euractiv.org



Source : loe.org



## H2 PIPE NETWORKS HAVE EXISTED FOR A LONG TIME

- Rare, because rarely needed

- But operated for decades in some very dense industrial regions

Northern Europe Air Liquide network :  
First segment built: 1938  
Length : 240 km  
Capacity : 25 000 t/year



Figure 4 – Réseau de gazoducs Air Liquide au Texas et en Louisiane



Figure 2 - Réseaux pipelines d'Air Liquide en Allemagne



France Hydrogène

GRT Gaz

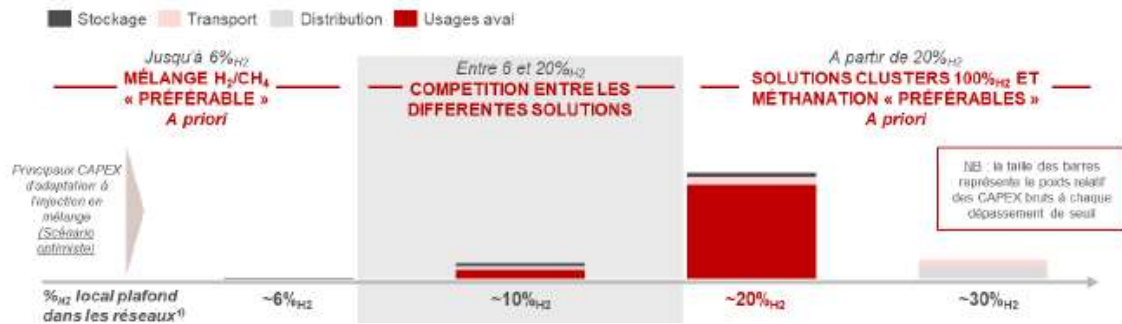
## FIRST OPTION : INJECTING HYDROGEN ON THE GAS GRID

### ● Injecting H<sub>2</sub> on the methane network can be seen as a transition technology

- Could solve issues of disconnection between supply and demand in early H<sub>2</sub> economy
- A storage potential of up to 40 TWh/year in France, according to the operators

### ● Adaptation costs hinder the profitability of the solution beyond a 20% vol. load

- Main issue : adapting end users installations



Source : « Conditions techniques et économiques d'injection d'hydrogène dans les réseaux gaz naturel », juin 2019



GRHYD  
Régulation des Énergies

ifp Energies  
nouvelles

## THE ISSUES WITH THE H<sub>2</sub>/CH<sub>4</sub> MIX

- The presence of H<sub>2</sub> is a hassle for the CH<sub>4</sub> users

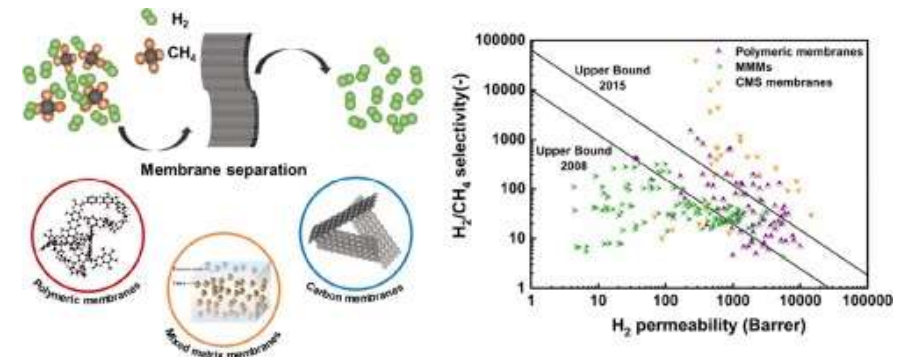
- Especially the industrials
- Extracting H<sub>2</sub> before sensitive CH<sub>4</sub> user is compulsory

- Doesn't mean you can easily use this hydrogen...

- Extraction needs to focus on producing pure CH<sub>4</sub> or H<sub>2</sub>, can't do both efficiently
- Standards for fuel cell usage are very, very pure

- Most of hydrogen will burn with methane

- Which, in the general case, is a nonsense economically



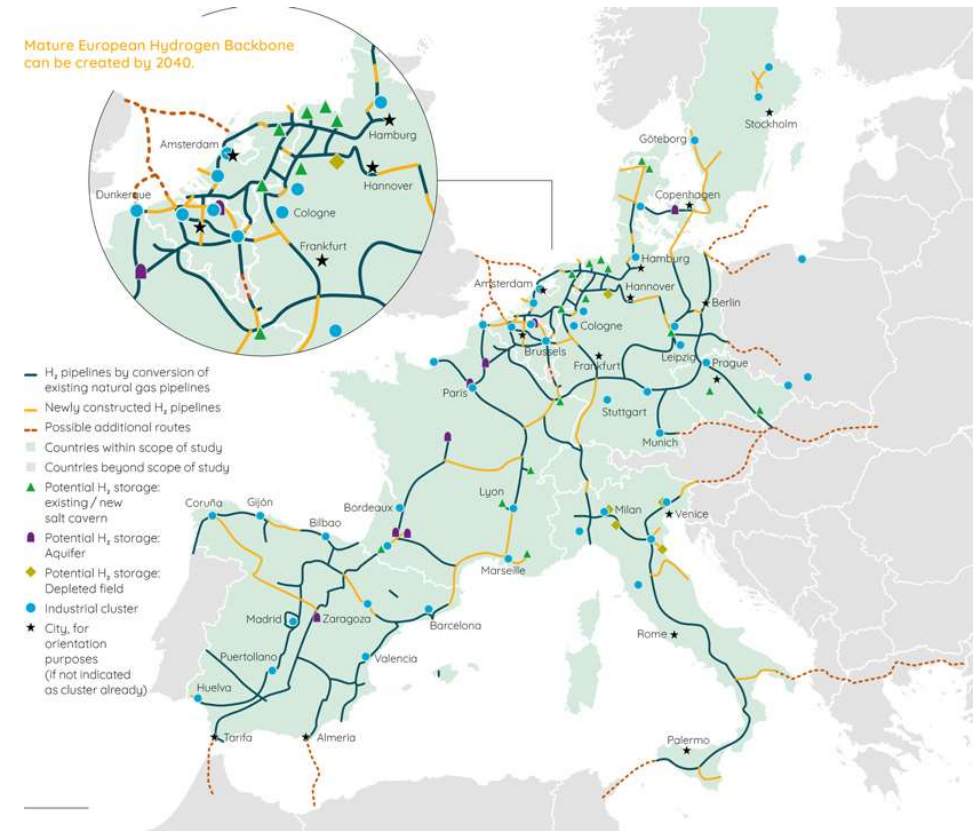
## TOWARD A 100% HYDROGEN NETWORK

### ● Example of the EU Hydrogen backbone

- Will ramp from 2030 to 2040
- 80% of existing recommissioned pipes

### ● Very attractive cost

- Pipe transport is almost % CAPEX and very scale dependant
- Transport costs as low as 0.1 or 0.2 \$/kg/1000 km
- Compatible with an hydrogen economy and balancing offer/demand balance at continental scale



Hydrogen Backbone Consortium

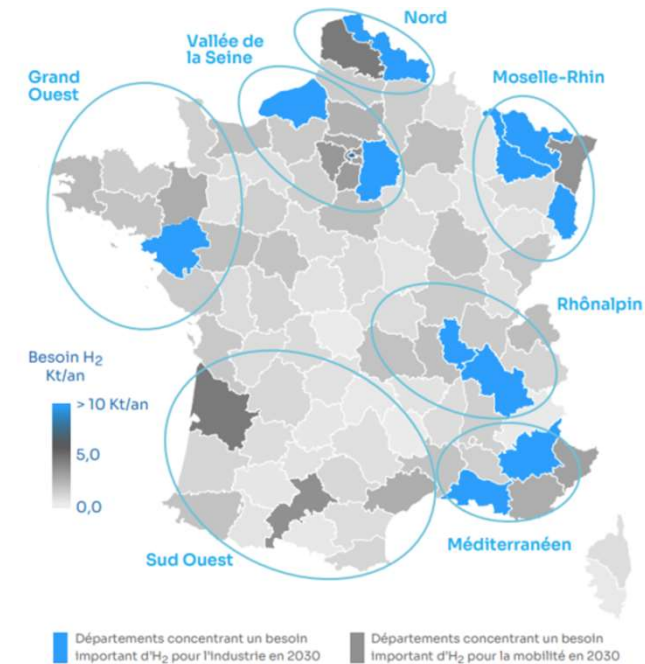


# NOT SO FAST ! THE HUB STRATEGY

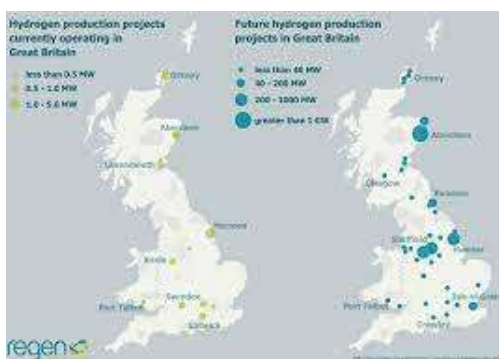
## ● What's a Hub ?

- An identified hotspot, for demand and potential offer
- Work on an independant, self-sufficient regional ecosystem
- Plan the logistics accordingly
  - Trucks to start with, but local pipes an option

## ● Connect the dots to create a global network



In France, 90% of potential needs identified in 7 key regions

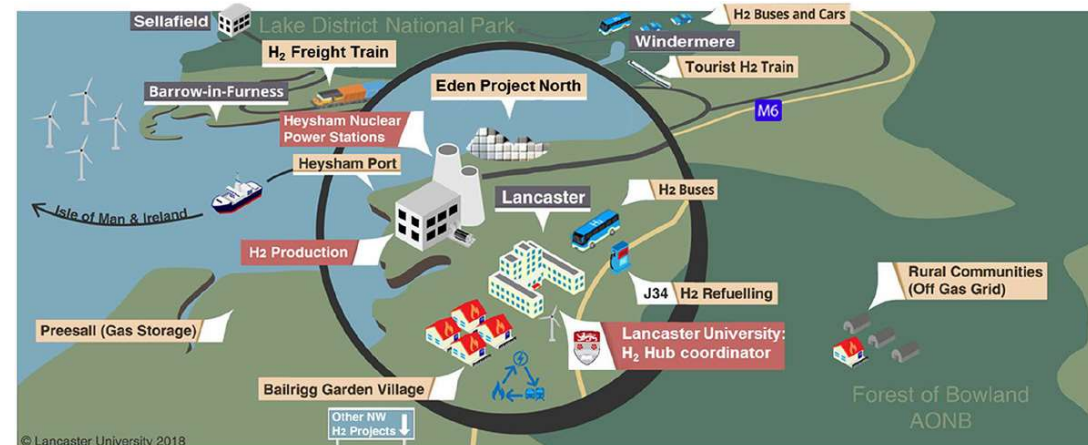


HyNTS  
Developing a UK  
hydrogen backbone

HyNTS  
Project Union

Project Union will involve the potential phased implementation of HTS pipelines to carry hydrogen and provide a hydrogen transmission backbone for the UK.

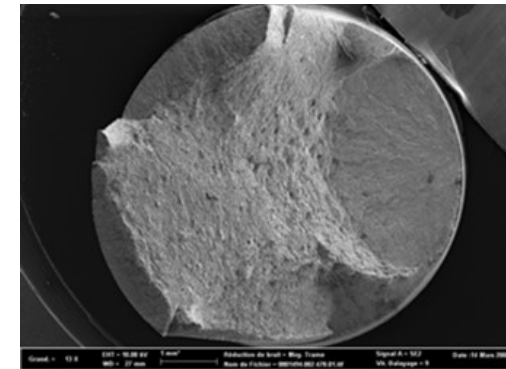
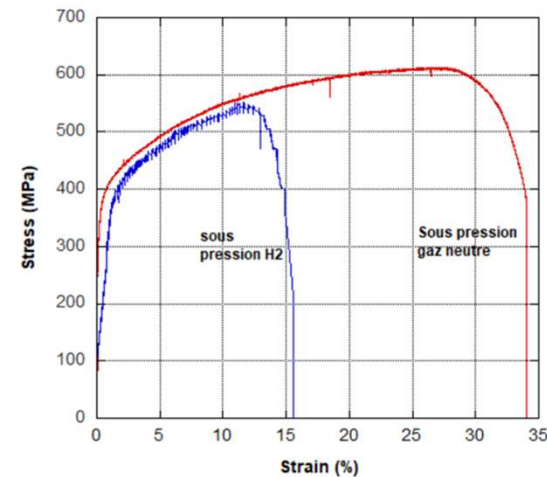
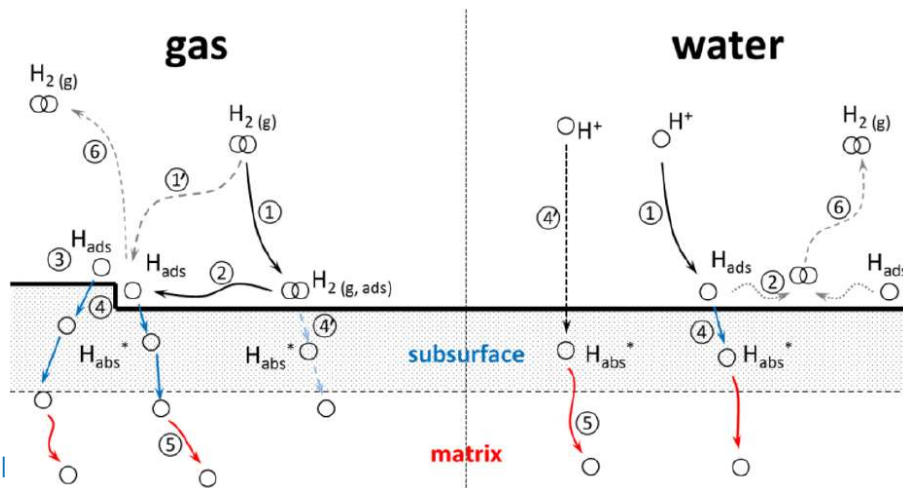
Teesdale - St Fergus
Humber - Teesside
Humber - Merseyside
South Wales - Bristol - Humber
Southampton - Cornwall





# THE PROBLEM OF STEEL ENBRITTLEMENT

- Hydrogen will dissociate on steel surface and penetrate the structure
  - Hydrogen will interact with the steel microstructure and degrade properties
  - Less plasticity, decreased fatigue properties and tenacity
- Nothing new under the sun, but...
  - New pipes : no issue adequate grade can be used, conditions not so severe
  - But old gas pipelines were not designed to handle hydrogen



# THE PROBLEM OF STEEL ENBRITTLEMENT

## ● Current focus of research by the gas industry

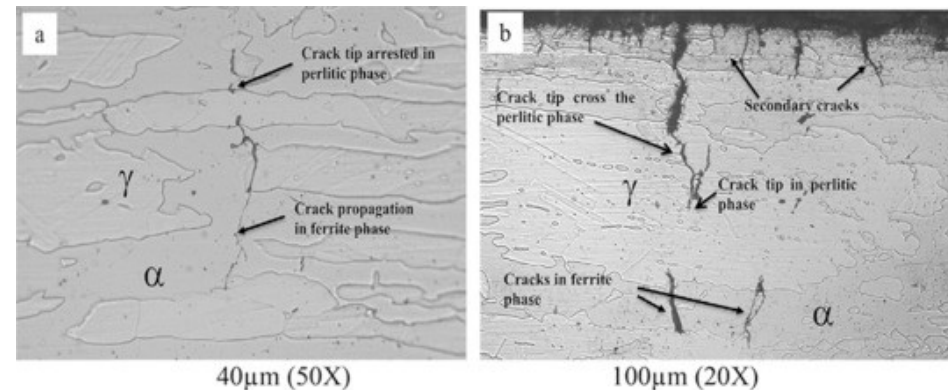
- First results not catastrophic
- Careful to influence of pipe age, impurities

## ● Solutions ?

- Be careful : avoid older tubings, reconsider dates for end of life
- Inner coating of the pipes as an ultimate solution ?



Arcelor Mittal

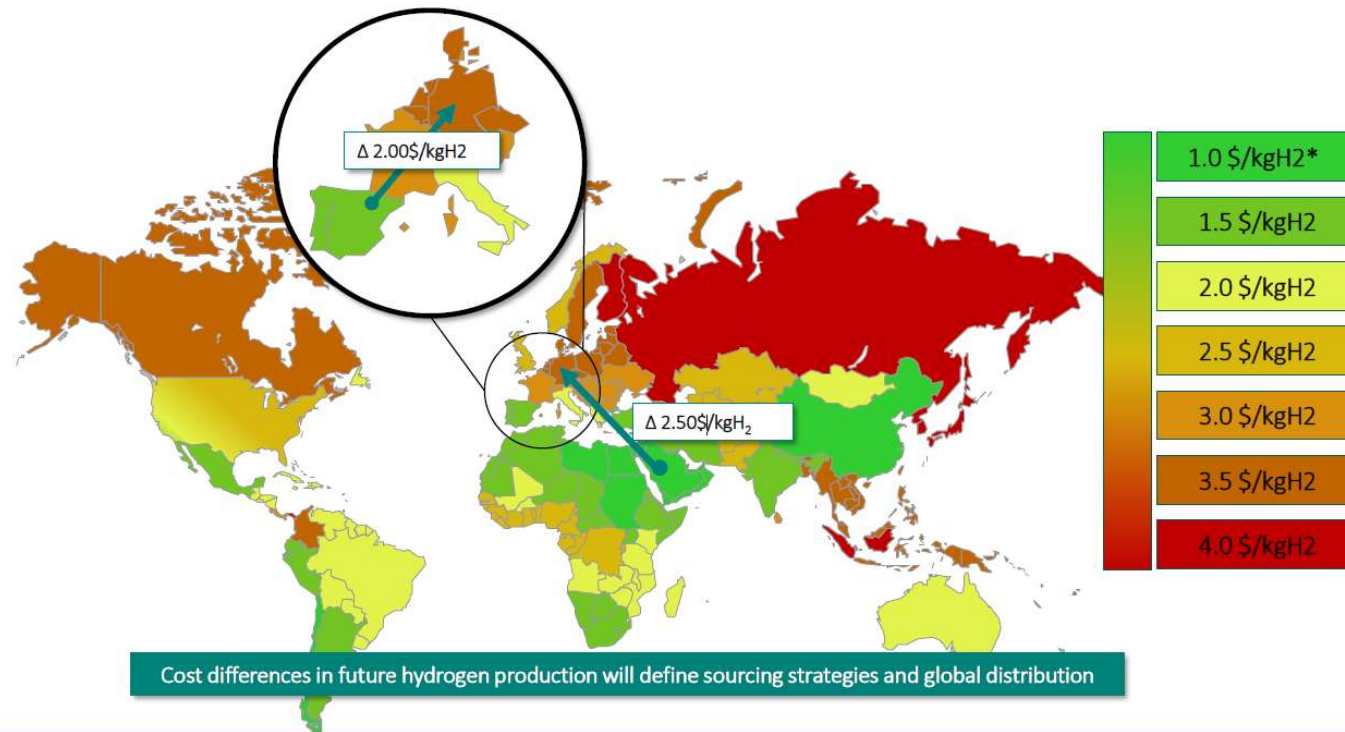


Bourkoot et al. 2019 International of Hydrogen Energy

## INTERNATIONAL MARITIME HYDROGEN TRANSPORT. WHY ?

### ● Exploiting disparities in production costs of H<sub>2</sub>

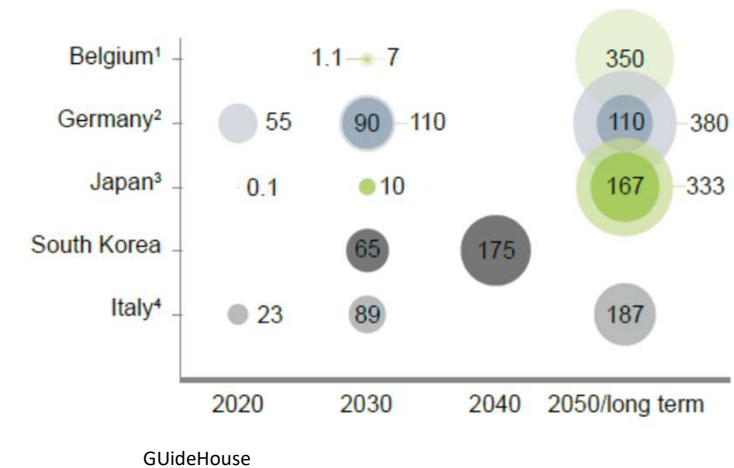
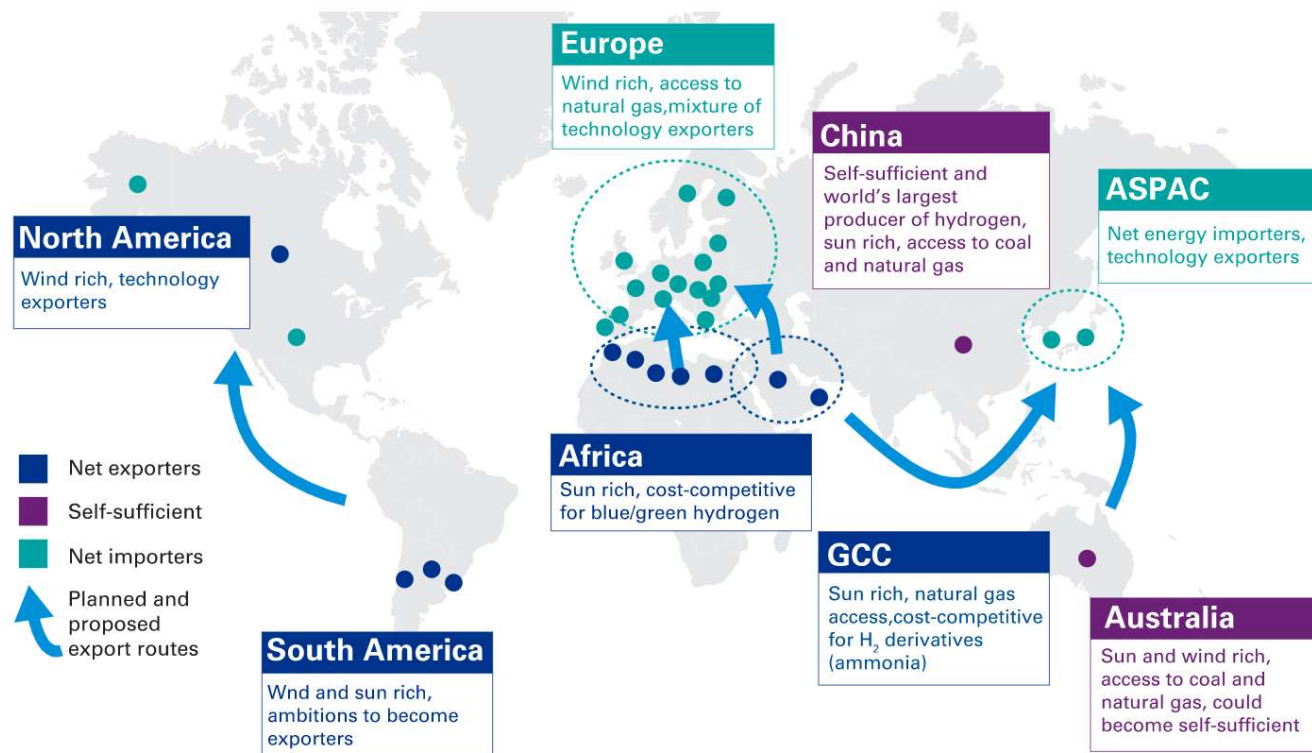
- Driven by the productivity of renewables
- Up to 2-3 \$/kg gradients



Sources: IEA 2019, The future of hydrogen and \*IEA 2019, Hydrogen: A renewable energy prospective (value PV costs 2050)

## INTERNATIONAL MARITIME HYDROGEN TRANSPORT. WHY ?

- Can potentially correspond to durable discrepancies between offer and demand
  - Japan could import the equivalent of 1/3 of its electricity



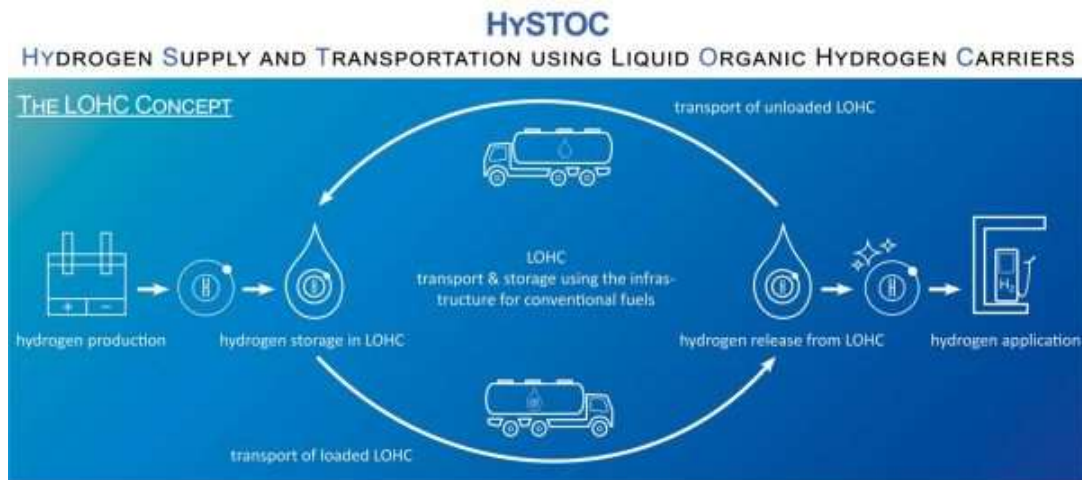


## OPTIONS FOR NAVAL TRANSPORT

- Direct transport of liquid hydrogen
- Liquid Organic Hydrogen Carriers
- Don't transport hydrogen, but e-fuels or ammonia



## OPTIONS FOR NAVAL TRANSPORT : LOHC

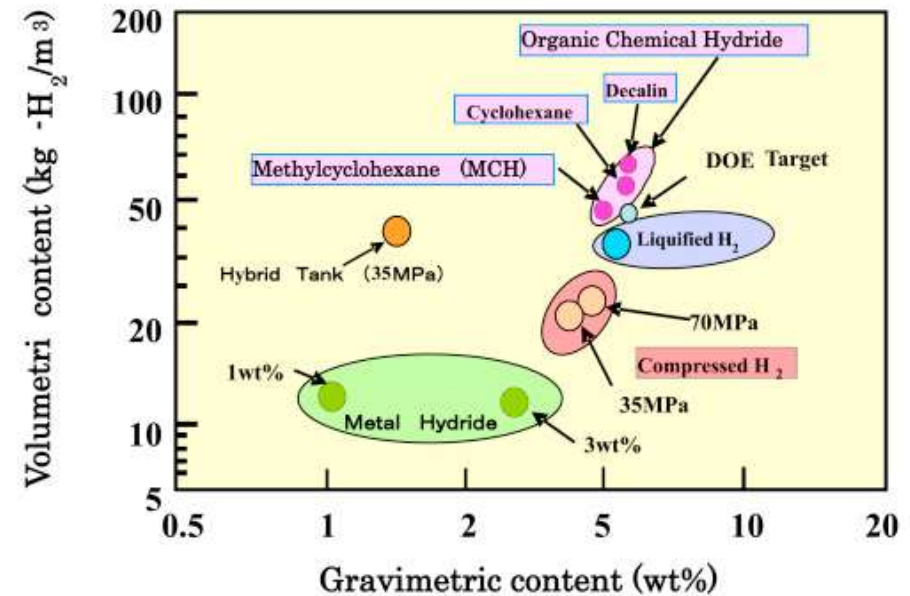


This LOHC hydrogen storage system is part of a project that has received funding from the Fuel Cells and Hydrogen 2 Joint Undertaking under the European Union's Horizon 2020 research and innovation programme under grant agreement No 779694.



### ● The best of two worlds :

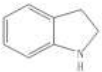
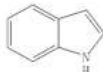
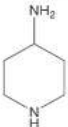
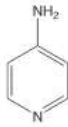

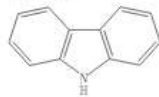

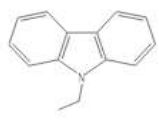
- The energy density of hydrocarbons
- With actual  $\text{H}_2$  for use





























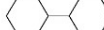
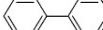
# OPTIONS FOR NAVAL TRANSPORT : LOHC

## What to look at ? :

- H<sub>2</sub> content
- (De-)hydrogenation energy
- Carrier loss per cycle
- Viscosity
- Toxicity of either forms

	Products of dehydrogenation	Hydrogen capacity (wt%)	Calculated $\Delta H$ (kJ/mol <sub>H<sub>2</sub></sub> )	Catalyst and activity	Ref.
Indoline 	indole 	1.7	51.9	5% Pd/C: 110 °C, conversion 100% 5% Ru/C: 110 °C, conversion 100%	[47] [47]
4-Aminopiperidine 	4-aminopyridine 	6.1	56.7	10% Pd/SiO <sub>2</sub> : 170 °C, conversion 66%	[48]
Dodecahydrocarbazole 	carbazole 	6.7	51.1	5% Pd/C: 170 °C, conversion 100%	[19]
Dodecahydro-N-ethylcarbazole 	N-ethylcarbazole 	5.8	50.6	5% Pd/C: 170 °C, conversion 100% 4% Pd/Lithium aluminate: 197 °C 5% Pd/SiO <sub>2</sub> : 170 °C IrH <sub>2</sub> [2,6-C <sub>6</sub> H <sub>3</sub> -(OPBu <sup>t</sup> ) <sub>2</sub> ] <sub>2</sub> : 200 °C	[19] [46] [52] [53]

Entry	Storage media	Hydrogenated form	Dehydrogenated form	Density (g mL <sup>-1</sup> )	(and experimental) $\Delta H_{H_2}^0$ (kcal mol <sup>-1</sup> )	wt%	g L <sup>-1</sup>
N-substituted heterocycles							
5	Dodecahydrocarbazole			1.298	12.2	6.7	87
6	Dodecahydro-N-ethylcarbazole			0.931	12.1 (12.4) <sup>d</sup>	5.8	54
7	Indoline			1.063	12.4	1.7	18.1
8	4-Aminopiperidine			0.945	13.3	6.0	57
9	Piperidine-4-carboxamide					4.7	
10	Perhydro-4,7-phenanthroline			0.958		7.2	69
11	2-Methyl-1,2,3,4-tetrahydroquinoline					2.7	
12	2,6-Dimethyldodecahydro-1,5-naphthyridine					6.0	
1,2-BN-heterocycles							
13	1,2-BN-cyclohexane			1.011		4.7	48
14	3-Methyl-1,2-BN-cyclopentane			0.894		4.7	42

media	Hydrogenated form	Dehydrogenated form	Density (g mL <sup>-1</sup> )	Calculated (and experimental) $\Delta H_{H_2}^0$ (kcal mol <sup>-1</sup> )	Hydrogen content	
					wt%	g L <sup>-1</sup>
cyclohexane			0.779	15.6 (16.42)	7.2	56.0
cyclohexane			0.770	16.3	6.2	47.4
cyclohexane			0.896	( <i>cis</i> -) 15.1 (15.29) ( <i>trans</i> -) 15.8 (15.91)	7.3	65.3
cyclohexyl			0.883	16.0	7.3	64.2

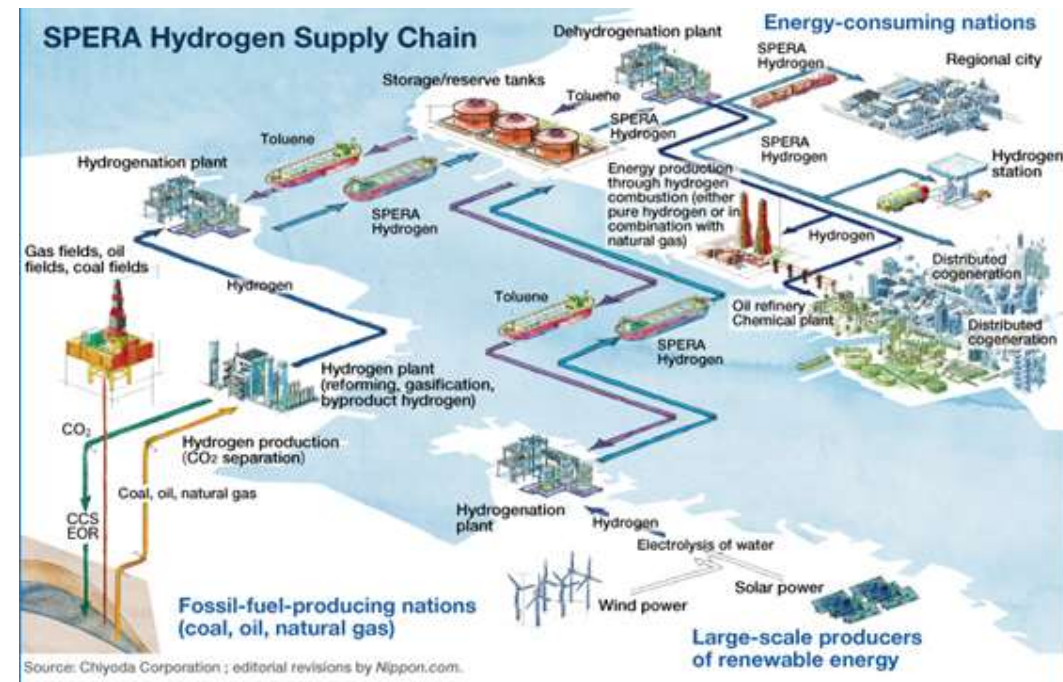
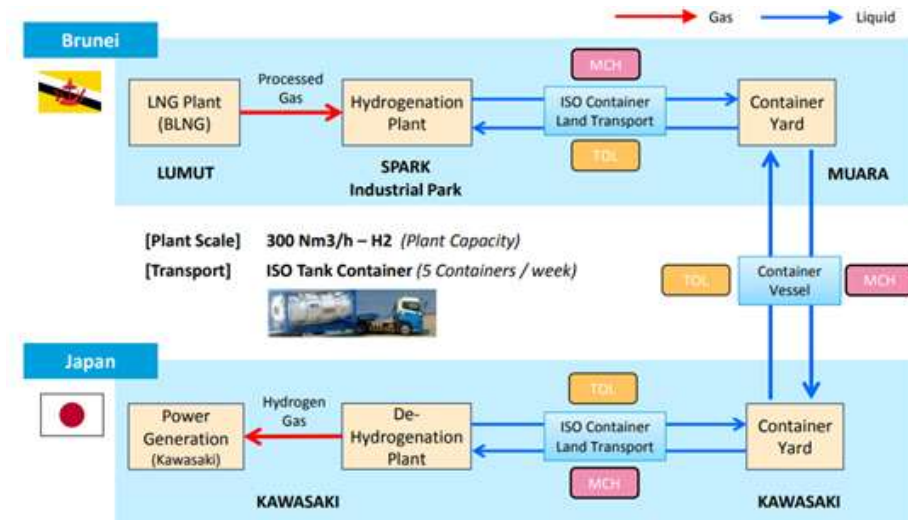
## OPTIONS FOR NAVAL TRANSPORT : LOHC

- Logistics is heavy

- Tankers circulate loaded in AND out
- Petrochemical installations at both end

- But technologies are quite mature

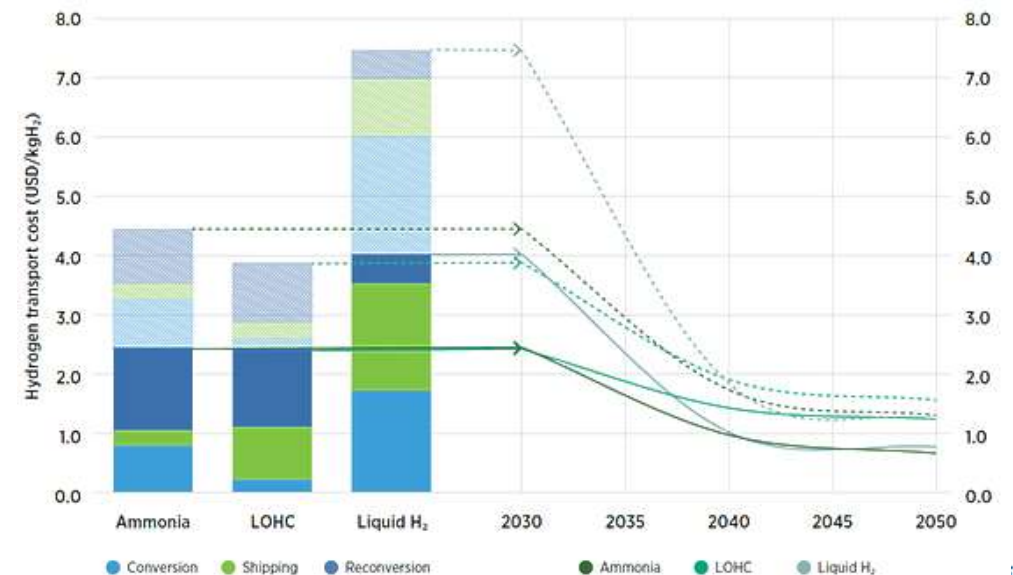
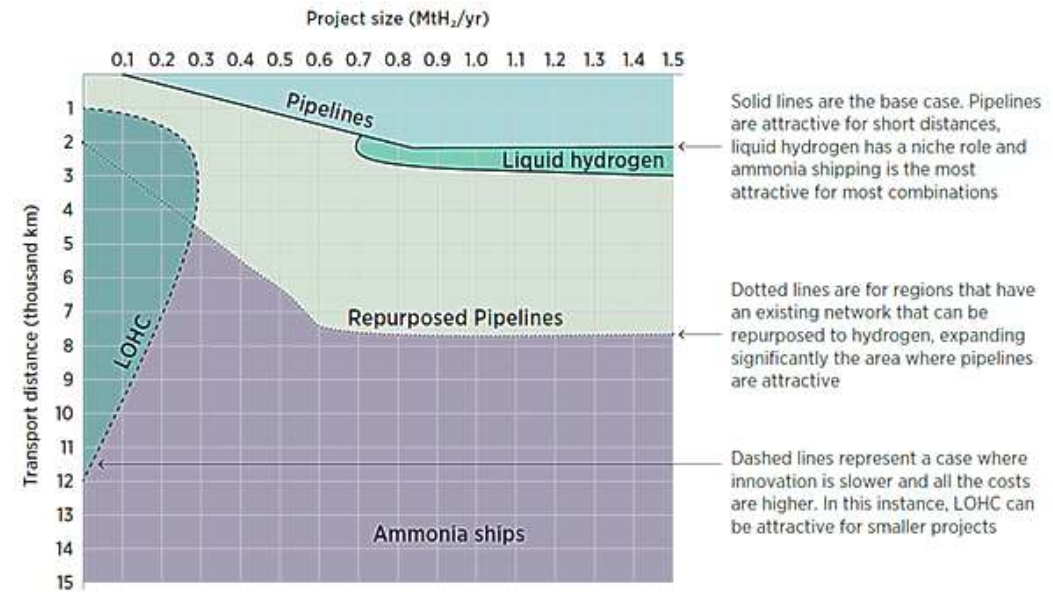
- Mostly adaptations of existing refining technologies





## ECONOMY : A CLOSE CALL

- Liquid hydrogen seems expensive
  - But is directly usable at reception
- LOHC and ammonia competes closely...
  - Around 2 or 3 \$/kg, compatible with anticipated contrasts in production costs
- Ammonia wins if used directly
  - But it is still a horrible molecule
- Require engaging decision, on a permanently evolving field



## BEYOND HYDROGEN : BALANCING THE ELECTRICITY GRID

- If intermittence is high on electric production, you need to store energy

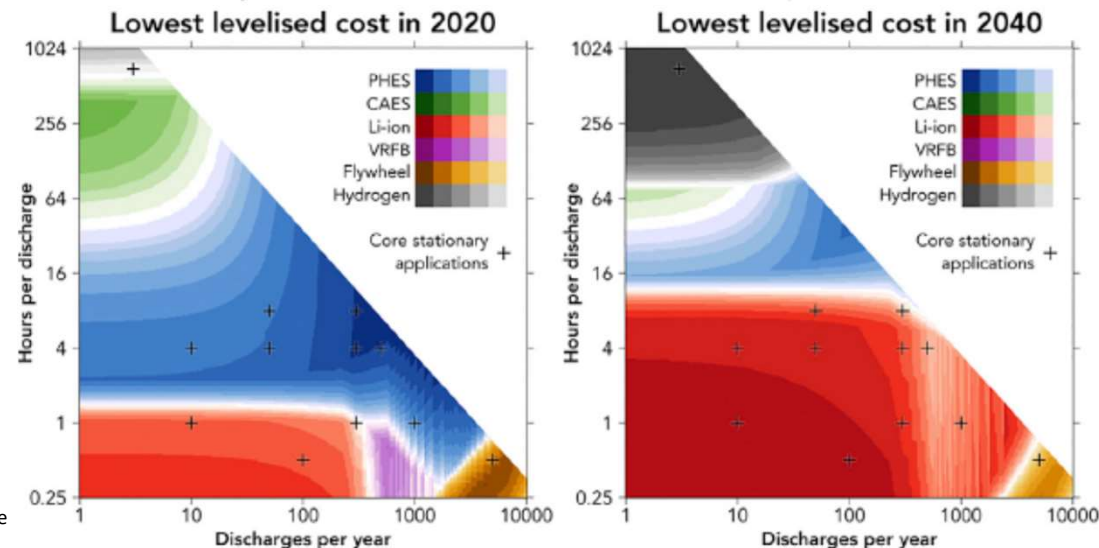
- Intermittence increasing drastically with renewables
- Produce hydrogen during peak production, use it to generate during peak demand

- Except H2 is a terrible way of storing electricity !

- Only relevant for long term, seasonal storage, and far in the future
- 30% efficiency of the power-to-gas-to-power chain

- Far better to insert hydrogen in a global strategy of your energy needs

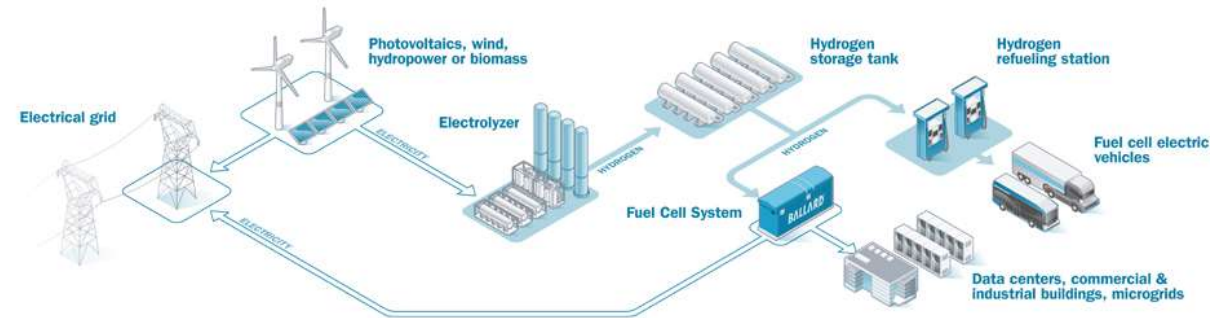
- Produce H2 during peaks, ok, but use it as hydrogen
- Could actually work even with a high proportion of renewables



# YOU REALLY WANT H2-ELECTRICITY ? WHAT'S YOUR OPTIONS ?

## ● Large scale fuel-cells

- Most efficient
- Huge CAPEX



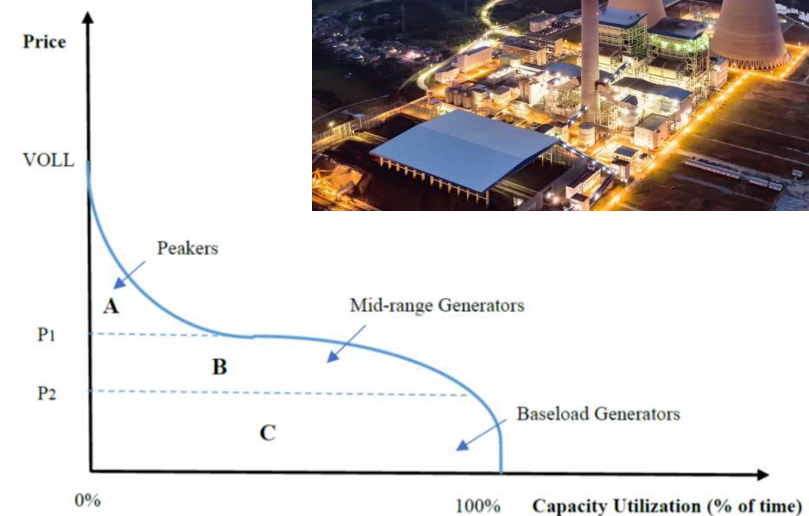
## ● Converted thermal powerplant

- Not as efficient as fuel-cell
- Could also burn ammonia, biomethane...
- Option to use existing (adapted) power plants

## ● Thermal plants are better peaker systems

- Using even a very expensive fuel make sense
- Starts on demand
- Low CAPEX very important for peakers

Future importing countries (Japan, Germany)  
bet on converting their power plants



## GERMAN CASE : GEOGRAPHIC UNBALANCE

### ● German energy geography

- Potential for renewables (wind) and naval imports are in the North
- Industry concentration in the South

### ● Hydrogen transfer to balance ?

- Electric infrastructure not up to the task for North/South transfers
- Hydrogen pipes could do the job
- And import through pipes could come from the South



APSIS







## SAFETY AND QUALITY

## VERY DEMANDING REQUIREMENTS ON PURITY

### ● A very demanding list

- It's long !
- Some very low values here : sulphur, water , oxygen...
- Some complex to define : « total sulphur compounds »

### ● Justified by aging problems on fuel cell

- No problem for chemical or combustion usage

### ● Challenging to even measure

- Techniques are making progress, but at a point complete characterization was almost impossible
  - Now it would still cost 20 000 \$
- No total online quality control possible !
- You have to guarantee no pollution can happen

#### Maximum concentration of individual contaminants

Water (H <sub>2</sub> O)	5 $\mu\text{mol mol}^{-1}$
Total hydrocarbons (methane basis) <sup>b</sup>	2 $\mu\text{mol mol}^{-1}$
Oxygen (O <sub>2</sub> )	5 $\mu\text{mol mol}^{-1}$
Helium	300 $\mu\text{mol mol}^{-1}$
Total Nitrogen (N <sub>2</sub> ) and Argon	100 $\mu\text{mol mol}^{-1}$
Carbon dioxide (CO <sub>2</sub> )	2 $\mu\text{mol mol}^{-1}$
Carbon monoxide (CO)	0.2 $\mu\text{mol mol}^{-1}$
Total sulphur compounds (H <sub>2</sub> S basis) <sup>c</sup>	0.004 $\mu\text{mol mol}^{-1}$
Formaldehyde (HCHO)	0.01 $\mu\text{mol mol}^{-1}$
Formic acid (HCOOH)	0.2 $\mu\text{mol mol}^{-1}$
Ammonia (NH <sub>3</sub> )	0.1 $\mu\text{mol mol}^{-1}$
Total halogenated compounds <sup>d</sup> (Halogenate ion basis)	0.05 $\mu\text{mol mol}^{-1}$
Maximum particulates concentration	1 mg kg <sup>-1</sup>

Bacquart et al. 2018 International Journal of Hydrogen Energy

## ORIGIN OF POLLUTIONS

### ● SMR production

- Sulphur, CO, CO<sub>2</sub>...
- Actually well handled by refineries massive purification systems



AICHE

### ● Electrolysis

- Water, O<sub>2</sub>...
- Less problematic impurities, but standards still to be built



Peric

### ● But the problems don't end at production

- Keeping the hydrogen pure is a problem at those standards
- Transport ?
  - A 50 years-old methane pipeline is not going to be mirror-clean...
- Compression ?
  - Need to develop lubricant-less technologies

## PURIFICATION METHODS AND STRATEGY

### ● Option 1 : Keep it clean, at all cost !

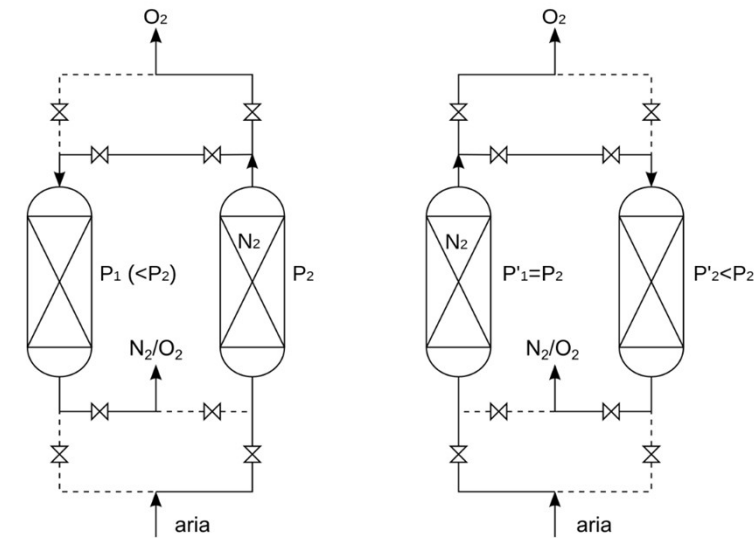
- Current strategy for commercial  $H_2$
- Large-scale, centralized purification at the plant
- Perfect control of downstream conditions

### ● Option 2 : Guarantee purity at the consumer end

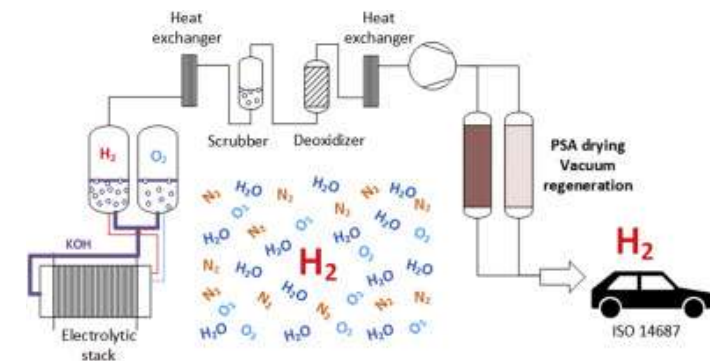
- You can (somehow) tolerate degrading purity in transport
- Last step of purification just before consumer delivery
- Small purifiers on each delivery points
- Don't bother with users who don't use a fuel cell

### ● Important, because purification techniques are scale dependant

- Massive PSA/TSA units in large plants
- Small, limited purifiers still to be optimized



Wikipedia



Ligen et al. 2020 International Journal of Hydrogen Energy



## SAFETY : AN EXPLOSIVE GAS

### ● Flamability, explosivity

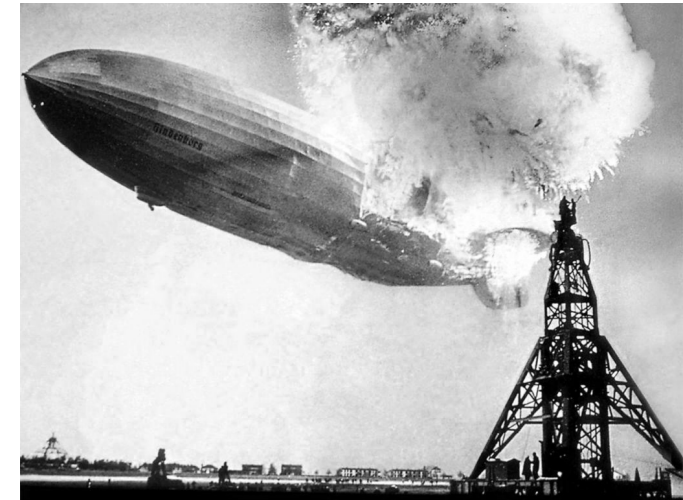
- Flamability range : 4%-75% at ambient pressure and T°
  - Methane : 5-15%
- Detonation range : 11%-59% (supersonic shockwave)
- Ignition energy 17  $\mu\text{J}/\text{mole}$ 
  - 15 times less than methane (290  $\mu\text{J}/\text{mole}$ )
- Any mix of hydrogen and oxygen can burn, with barely a spark

### ● Hydrogen flame is often too hot to be visible

- 2000°C... mostly in the ultraviolet range

### ● A leaky gas...

- Small molecule, will permeate tiny cracks and many materials
- Known adverse effect on steels



Imagesensors.org

## SAFETY : SENSORS

### ● Hear it : leak detector

- Leaking gas make a distinctive sharp gas
  - Even more specific in ultrasound
- Huge progress using AI to isolate the noise specific signature

### ● Smell it : gas detector

- Detect the presence of the gas in the atmosphere
  - Catalyst, infrared
- 1/10 of the inferior explosivity range

### ● See it : flame detector

- Detect the light emitted by a flame
- Huge progress again in discrimination capacity



MSA



## SAFETY : ON THE ROAD

### ● A safety challenge

- Vehicles are disseminated in a free environment
- Wide variety of potential problems

### ● Hydrogen disseminate quickly

- Can make the situation less dangerous then with liquid fuels
- Won't save you in a confined environment

### ● Two main design principle

- Maintain mechanical integrity of the reservoir box
  - On the model of deforming the structure to preserve the passengers
- « Leak before break », especially in a fire situation
  - Because breakage most likely lead to an explosion
  - Thermal pressure release systems



Hyundai



U.S. Department of Energy

## SAFETY : STATIONNERY STORAGE AND SOCIAL ACCEPTANCE

### ● Most serious accidents implied stationnary storage

- More hydrogen stored
- Confined space
- Industrial handling of hydrogen is an old story, though



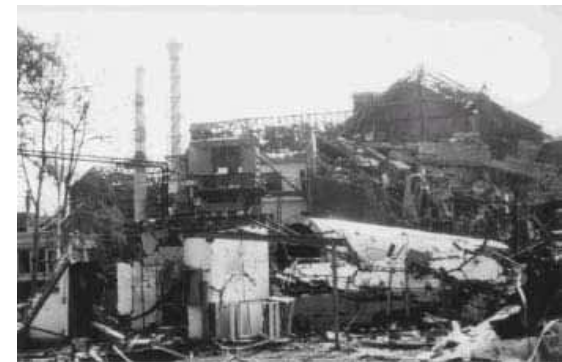
### ● Pressure from industrial to relax constraints on stored hydrogen

- Limits that were sent for a niche chemical are very constraining for mobility applications
- But those limits were set for a reason...
- Growth of the sector is very fast, actor's skills a real issue



### ● Social acceptance

- Public image of hydrogen is generally very good, and projects easily accepted
- One serious accident can change this reputation rapidly







## CASE STUDY

## YOUR TIME TO WORK !

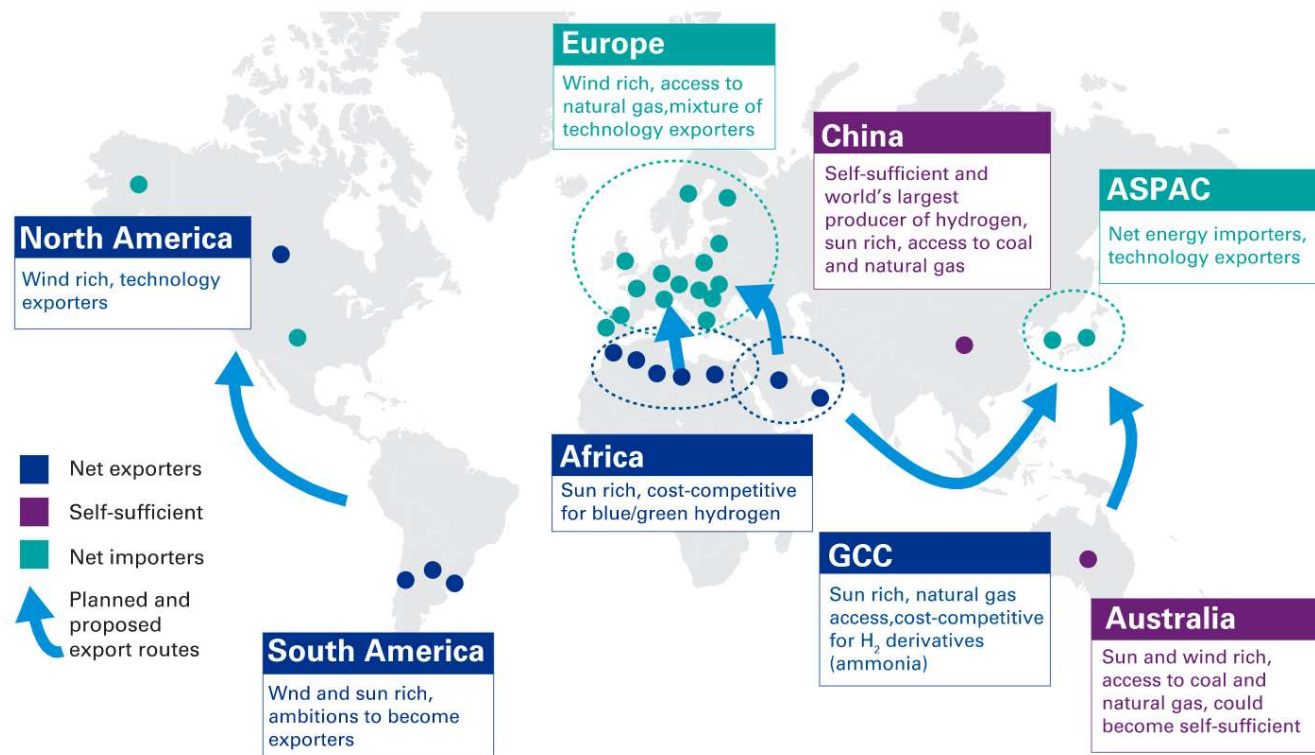
- India's hydrogen strategy: global

- How do you think India will use hydrogen to balance its energy needs
  - Hubs ? Transport ?
  - Import-export ?

- I don't have a correction !

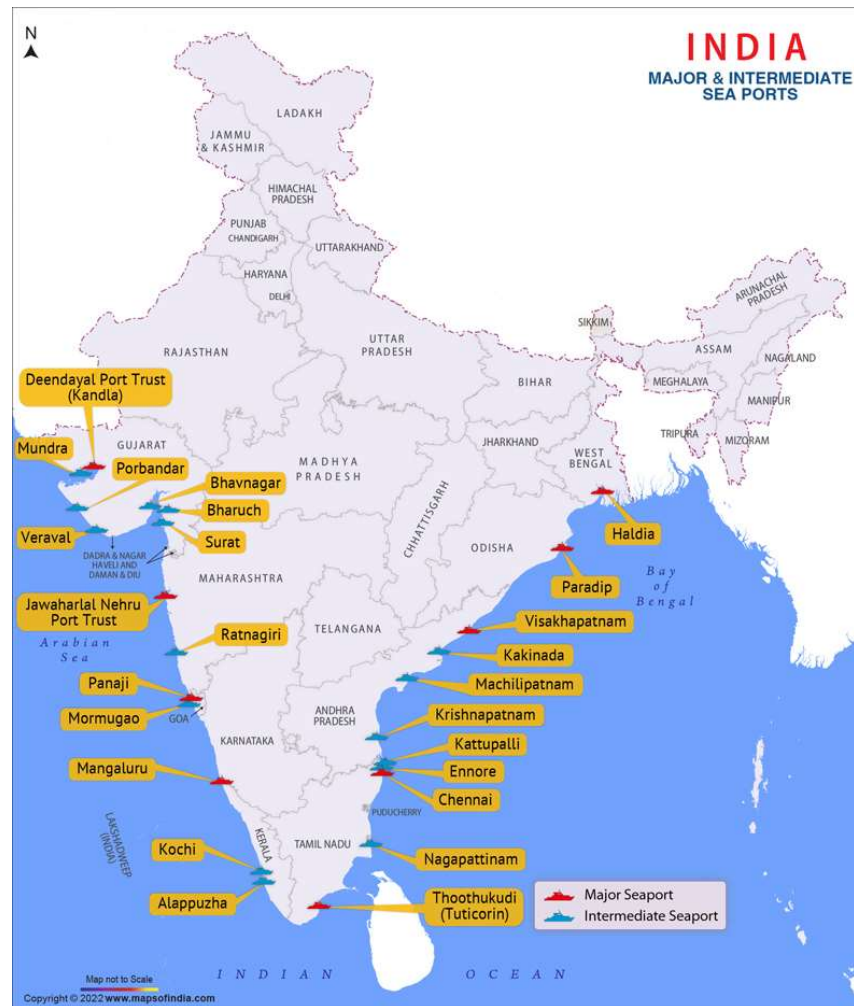
- You are the experts concerning India, not me...
- But I have a few éléments and we can debate

## BOTH A MASSIVE PRODUCER AND A MASSIVE CONSUMER



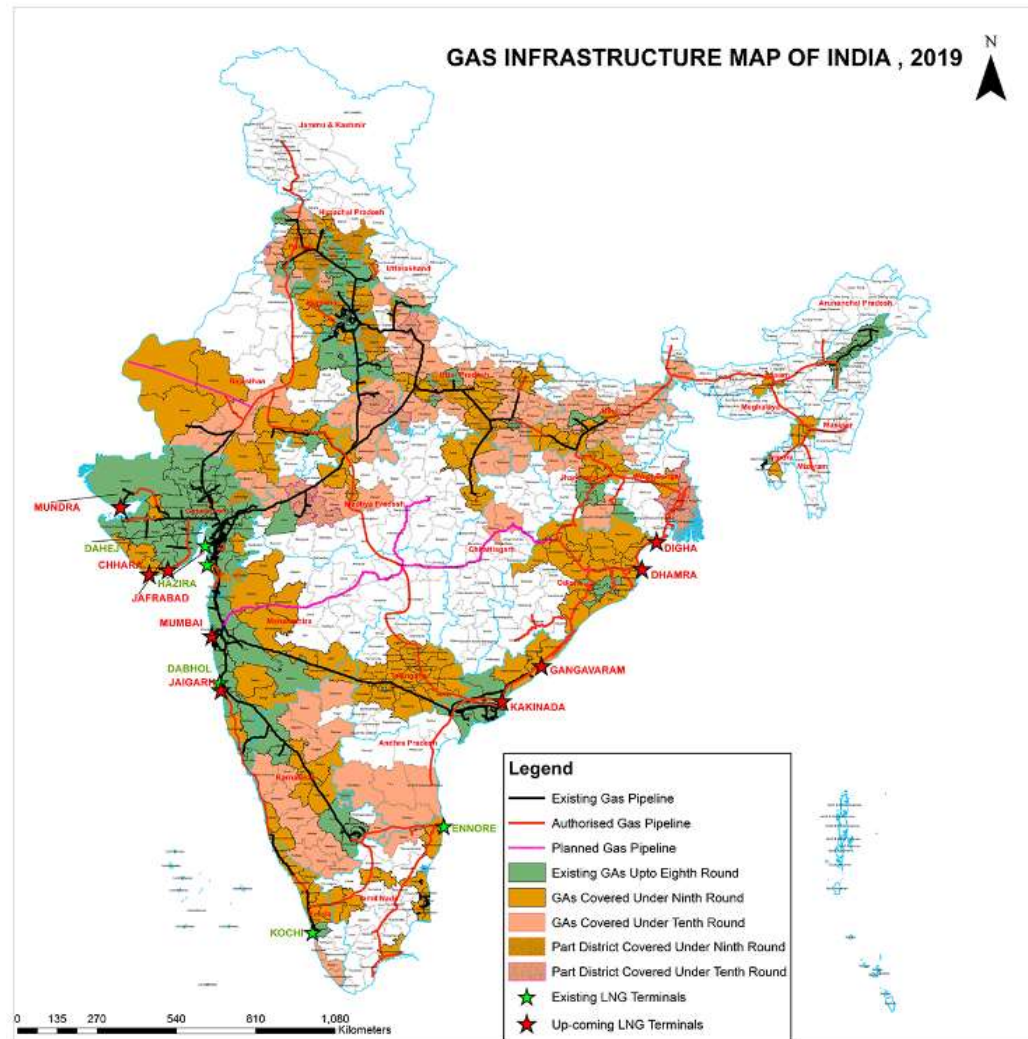
Global hydrogen hot spots and distribution corridors (not exhaustive).

## PORTS AS POTENTIAL HUBS

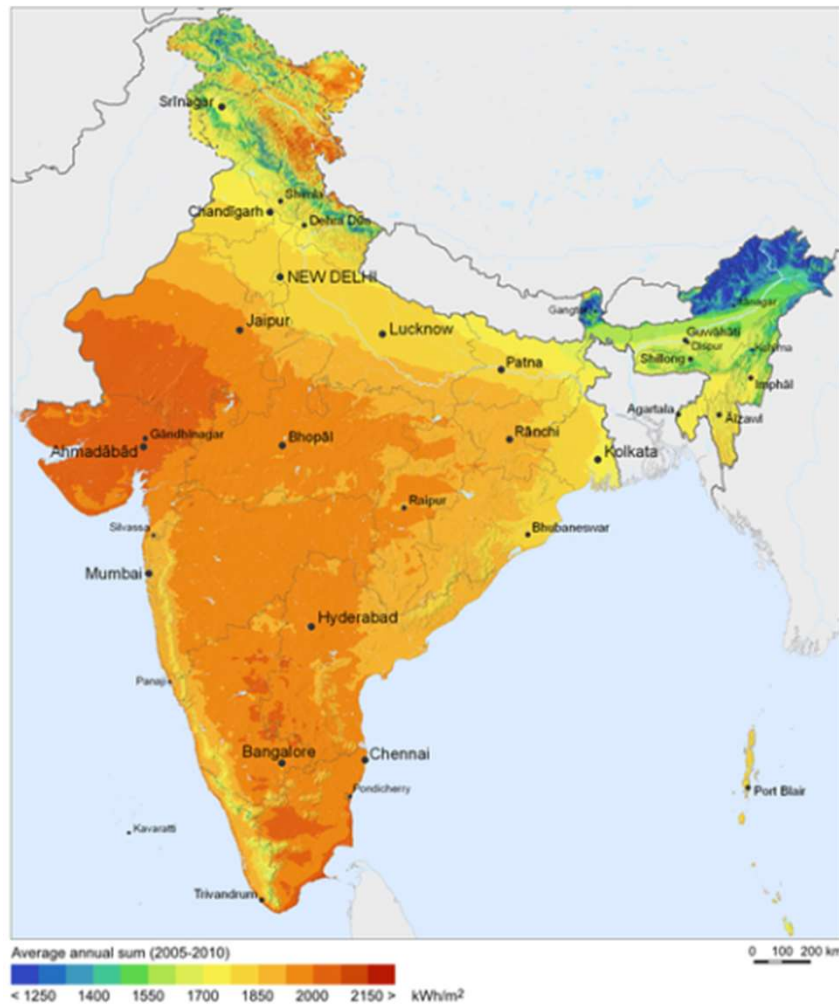




## A RESPECTABLE GAS GRID



## WEST-EAST UNBALANCE ?





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